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## Project for the handling and storage of vitrified high-level waste

Saint Gobain Techniques Nouvelles Oct 1977

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PROJECT FOR THE HANDLING AND STORAGE OF VITRIFIED HIGH LEVEL WASTE

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## I. INTRODUCTION

1.1. The disposal of high level wastes

The high level wastes (HLW) originating from the reprocessing of spent fuels contain $99.9 \%$ of the fission products, a maximum content of 1 . of the uranium and plutonium present in the spent fuels, and most of the other transuranic elements (1).

These wastes will be solidified in the reprocessing plant of La Hage aiving an homogeneous borosilicate glass of a very low leachability. The ILLW corresponding to one ton of uranium reprocessed are finally enclosed in a glass block of 150 liters. The composition of these HIW, hence their radioactivity and their heat release depends on the burn-up of the fuels. The related figures will be higher for PWR fuels which can be irradiated up to $33.000 \mathrm{MWd} / \mathrm{t}$, than for BWR fuels whose burn-up is limited to $28.000 \mathrm{MWd} / \mathrm{t}$. Our calculations are based on PWR fuels in order to be conservative (see Annex 1).

The g]ass blocks enclosed in a sealed containermade of refractory stainless steel will be cooled in the reprocessing plant for the first years (up to 10 years), till their heat release will be lower than 7.6 watts/liter, i.e. 1140 watts/canister.

The glass containerswill then be shipped to Sweden at a rate of 300 canisters/year, after 1990 . The shielded casks to be used for this transportation will be very similar to the casks presently used for spent fuels transportation.
-/..
(1) This content of $1 \%$ is a conservative figure. Actually, the losses recovered in the HLW of the reprocessing plant are in the range of $0.5 \%$ for $U$ and Pu. Basically the safety problem of the disposal of HLW is the same for both cases.

The best way to dispose of these vitrified wastes is to bury them permanently underground. Furthermore, the cristalline bedrock in Sweden offers favorable geological conditions.

The glass containers will be deposited in pits accessible by tumels drilled at 500 m below ground surface. The density of storage will depend on the heat release of the canister, and on the heat conductivity of granite and of the different materials inserted between the granite and the canisters. The glass is subject to devitrification when its temperature is higher than $600^{\circ} \mathrm{C}$. It crystallizes, becomes brittle, and its leachability increases though remaining in a good range.

It is therefore advisable to keep the maximum temperature of the glass at the center line of the container below $500^{\circ} \mathrm{C}$. This is normally achieved in the cooling ponds in La Hague, and in the shipping casks. If the canisters were to be disposed of immediately after reception in Sweden, i.e. after 10 years maximum of cooling, the temperature rise in the final storage would greatly limit the density of storage in the granite, and also the corrosion effects on the final packing would be enhanced.

In order to realize the best conditions of storage, it is preferable to put the glass containers in an interim storage for at least an additional period of 30 years. This storage will be cooled by forced air, and possibly by natural convection. After this total period of 40 years of cooling, the maximum heat release will be :

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        3.42 watts/liter of glass
        or 513 watts/container
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This heat release will allow a density of storage of one container per 100 m 2.

The potential health hazard of the HLW varies greatly along the time. If we consider the decay curves of the HLW we see that the decrease of activity (and correlatively of heat release) is relatively fast during the first 600 years. This corresponds to the decay of the Iission products whose activity is:
312.000 Ci $\beta /$ ton of uranium (i.e. per canister) after 10 years 32 Ci $\beta /$ ton of uranium after 600 years

The remaining activity, due to the actinides (uranium and transuranio elements) decrease very slowly:
3.130 cio/ton of uranium after 10 years
$135 \mathrm{Ci} \alpha /$ ton of uranium after 600 years

Beyond this time, the activity of actinides is predominant.

Taking thege facts into account, the strategy of the final storage will be the following :

1/ We shall provide the best isolation possible of the glass container by means of an encapsulation during the first 600 years in order to avoid any leaching of the fission products, constituted mainly by cesium and strontium.
$2 /$ After this period, the health hazard will be reduced by a factor of 10.000 . The release of the radioisotopes to the environment will remain limited by :

- Leachability of Pu appears to be somewhat lower than for the fission products (Cs/Sr) after long period of leachinge Leachability of Am is lower than for Pu.
- the diffusion of the actinides into the ground in the rock;
- the space between the container and the rock will be filled with a material having a low permeability and some ion exchange capacity.

Thus the rock, the encapsulation and the low leachability of the glass and the filling material constitute the four barriers provided against the migration of radioactive nuclides to the biosphere.

### 1.2. Basic concept

According to the hercabove considerations, the main steps toward the final disposal of the vitrified HLW will be the following :
1.2.1. Transportation

The glass containerswill be shipped from the reprocessing plant of la Hague to Sweden in shielded casks similar to those presently used for spent fuels. For example, the NTL 12 cask used for transportation of 12 PWR fuel elements with a heat release of 120 KW can be used for transportation of 15 glass containerswith a maximum heat release of 17 KW The $\gamma$ shielding is more than enough. The neutron shielding will also be sufficient.

### 1.2.2. Intermediate storage

This storage will be built 30 meters under ground, for physical protection, and will allow an additional cooling of at least 30 years. Its capacity will be in a first step of 3000 containers, extensible to 6000 containers.

The transportation casks brought by trucks will be discharged in a special cell. After monitoring the containers will be stored in vertical pits cooled by forced air. This type of storage will be quite similar to the storage to be put in operation in early 1978 in Marcoule. All safety measures will be taken to maintain the necessary ventilation. In case of power failure of long duration (some days) natural convection will maintain the glass below a maximum temperature of $500^{\circ} \mathrm{C}$.

The design of the whole facility will comply with the regulations in force in nuclear installations.
1.2.3. Encapsulation

After the additional 30 years cooling, the glass containerswill be put into a titanium canister prior to final disposal. This enveloppe (thickness 6 mm ) will have an excellent resistance to corrosion by ground water.

Between the titanium canister and the glass container, a filling of 10 cm of lead will be poured for the following purposes :

- Mechanical strength to avoid the deformation of the canister under a water pressure of 50 bars.
- Additional protection against corrosion.
- Attenuation of the $\gamma$ flux to minimize the radiolysis of ground water to a level, which will not be damageable for the titanium enveloppe.

This containering is performed in a special hot cell at the outlet of the intermediate storage.

The canisters are then transferred into the final repositery in a shielded cask. This cask is lowered to the mine through a material shaft by a drum hoist. This shaft is connected to the intermediate storage by an underground gallery.
1.2.4. Final repository

- The final repository will be located at a single level 500 meters below the rock surface (as an alternative a location at 1000 m depth could also be considered).
- The tunneling system will have 41 tunnels, each with a length of 1 km , with a $\mathrm{c} / \mathrm{c}$ distance of 25 meters and will then cover an area of $1 \mathrm{~km} \times 1 \mathrm{~km}$.
- The canisters will be placed in vertical holes with a diameter of 1 meter and a depth of 5 meters with a c/c distance of 4 meters. One canister will be placed in each hole. Backfilling of the holes will be made with a filling material having a good ion-exchange property (bentonite + quartz sand). This material will have a thickness of 20 cm . It will also prevent any damage to the canister from a possible tectonic movement.
- The holes will be drilled into the rock from the floor of tunnels having a width of 3.5 meters and a height of 3.5 meters.
- The repository will have four vertical shafts for communication and ventilation.
- Placing of canisters will begin at one end of the repository and proceed towards the other. Backfilling of the tunnels and the shafts will be made when all canisters have been placed, thus 30 years after the first canister has been placed.
- Until then the ventilation of the entire tunneling system will be maintained. This ventilation will be designed in accordance with the Mines Safety Regulations, taking into account the additional heat flux of the stored canisters.

Nuclear specifications will only apply to shielding against radiations ( $\gamma$ and neutron).

- Retrievability of the canisters before backfilling of the tunnels, has not been considered as an target. However it will remain possible.


## II. PROPERTIES OF THE HLW

2.1. Composition

We are assuming the HLW are coming from the reprocessing of PWR ruel elements,
with a burn-up of $33.000 \mathrm{MWd} /$ ton of U
a specific power of $30 \mathrm{MW} /$ ton of $U$
which is the most pessimistic case

Moreover, we suppose the reprocessing occurs 10 years after discharge of the reactor, which is an upper limit. In this case, the content of americium and its decay product will build-up in the spent fuels.

The detailed composition of the HLW is given in Annex 1 . These HLW contain all the fission products, $1 \%$ of uranium and plutonium present in the spent fuels, most of the other transuranic elements, and some metallic impurities from the process.

After solidification in the glass, the HLW content can be splitted as follows, per ton of uranium :

| Fission products | (as oxides) | $: 35,75 \mathrm{~kg}$ |  |
| :--- | :---: | :--- | :--- |
| Actinides | $"$ | $: 10,5$ |  |
| Metallic impurities | $"$ | $:$ | 2,25 |
| (Al, $\mathrm{Cr}, \mathrm{Fe}, \mathrm{Ni}, \mathrm{Cr}, \mathrm{Si}, \mathrm{Na})$ |  |  |  |

$$
\text { Total } \quad 58,5 \mathrm{~kg} / \text { ton of } \mathrm{U}
$$

2.2. Activities

With the same assumptions, the activities of the HLW are given in Annex 2. (This annex shows the spectrum of $\gamma$ curies which is used for the shielding calculations).

- 10 years after discharge of the reactor, the activity will be the following :
Fission products : $312.000 \mathrm{Ci} \beta /$ ton of U

$$
\text { Actinides } \quad: \quad 3.130 \mathrm{Ci} \alpha / \mathrm{t}
$$

$$
\text { Total }: 315.130 \mathrm{Ci} / \mathrm{t}
$$

- 40 years after discharge :

Fission products : $143.000 \mathrm{Ci} \beta / \mathrm{t}$
Actinides : $1.930 \mathrm{Ci} \mathrm{\alpha} / \mathrm{t}$ Total : $144.930 \mathrm{Ci} / \mathrm{t}$
2.3. Heat flux

The heat flux will be :

- 10 years after discharge :

Fission products : $1030 \mathrm{~W} / \mathrm{t}$
Actinides : $108 \mathrm{~W} / \mathrm{t}$ Total : $1138 \mathrm{w} / \mathrm{t}$

- 40 years after discharge :

Fission products : $\quad 447 \mathrm{~W} / \mathrm{t}$
Actinides $\quad: \quad \underline{66 \mathrm{w} / \mathrm{t}}$
Total : $513 \mathrm{~W} / \mathrm{t}$

The curve of the heat flux along time is given in Annex 3 .
2.4. Glass composition

The average glass composition will be the following :

| $\mathrm{SiO}_{2}$ | $: 60 \%$ |  |
| :--- | :--- | :--- |
| $\mathrm{Na}_{2} \mathrm{O}$ | $: 10 \%$ |  |
| $\mathrm{~B}_{2} \mathrm{O}_{3}$ | $: 16 \%$ |  |
| HLW (oxides) | $: 14 \%$ |  |
|  | (including transuranium elements and |  |
|  | Total | 100 |

This corresponds to a concentration of 150 liters of glass per ton of uranium, which volume will be poured in single container.
2.5. Glass container

The glass canister is made of refractory stainless-steel
(type Z $15 \mathrm{CN} 24-12$ : chrome $23-24 \%$, nickel 12-13 \%, carbon $0.15 \%$ )
The main dimensions are :

| Diameter | $: 400 \mathrm{~mm}$ |
| :--- | :--- |
| Overall height | $: 1500 \mathrm{~mm}$ |
| Opening diameter | $: 200 \mathrm{~mm}$ |
|  |  |
| Thickness - body | $:$ |
| - head | $:$ |
| Useful volume | $: \quad 150 \mathrm{~mm}$ |
| U liters |  |

The cover is automatically welded in hot cell with a plasma torch. The stainless-steel material has a good resistance to water or in air even at temperatures of $300^{\circ} \mathrm{C}-400^{\circ} \mathrm{C}$. This resistance will be very limited in saline water.
2.6. Glass properties

The glass block is homogeneous, isotropic and without porosity.

- Fusion temperature : about $1150^{\circ} \mathrm{C}$
- Crystallization temperature : above $600^{\circ} \mathrm{C}$
- Density : $2.8 \mathrm{~g} / \mathrm{cm} 3$
- Thermal conductivity : $1.2 \mathrm{~W} / \mathrm{m} \mathrm{x}^{\circ} \mathrm{C}$
- Leachability by water at $25^{\circ} \mathrm{C}$ :
- for fission products : $10^{-6}$ to $10^{-8} \mathrm{~g} / \mathrm{cm} 2 \mathrm{x}$ day
(higher figure for cesium , lower figure for ruthenium)
- for actinides : $10^{-7}$ to $10^{-9} \mathrm{~g} / \mathrm{cm} 2 \mathrm{x}$ day
(higher figure for plutonium, lower figure for americium)

The leachability increases with the temperature. It is 10 times higher at $70^{\circ} \mathrm{C}$.

The stability to radiation is very good. After irradiation up to $10^{11}$ rads, there is no energy build-up in the glass and the leachability is the same.

The incorporation of $\alpha$ emitters at high concentrations, simulating in one year the actual dose integrated in 1000 years gave the same results.

Bibliography : "La Vitrification en France des Solutions de Produits de Fission" by R. BONNIAUD (CEA-MARCOULE), Nuclear Technology Vol. 34, Aug. 77, pp. 449-460.

## III. INTERMEDTATE STORAGE AND ENCAPSULATION

3.1. General Description of Facilities

This installation has the function of :

- receiving the NTL 12 transport casks containing the containers of vitrified HLW coming from the reprocessing plant;
- unloading the containerscontained in the transport casks;
- transferring the containersto an intermediate storage for activity decay;
- transferring the containers, after decay, from the intermediate storage to a encapsulating cell where they will be put into titanium-lead canisters, before final storage.

It comprises :

- A cask receiving hall, in which the container transport cask arrives on its road trailer.
- A cask monitoring room, in which the internal activity of the cask is monitored before the containers are unloaded.
- An unloading cell, in which the following operations are carried out :
- the unloading of the cask;
- the provisional storage of the containers contained in the cask;
- their evacuation to the temporary storage.
- A recanning cell for damaged containers, in which the damaged and contaminating containers are put into tight containers before being removed to the temporary storage.
- An intervention cell, which permits : the transfer of equipment in the cells, its maintenance, its decontamination and conditioning in the event of its being evacuated outside the installation.
- A temporary storage, in which the containers are stored for the time necessary for their activity decay and temperature decrease before they are stored definitively.
- An encapsulation cell, in which, after temporary storage, the containers are conditioned before being put into their final storage pits.


### 3.2. Transport Cask

The containerscoming from the reprocessing plant are transported in a shielded cask, type NTL 12, in conformity with the prescriptions of the latest safety regulations of the IAEA for radioactive material transportation.

The conception of the packaging is of the type "all in" comprising between the stainless steel containment vessel and the outer shell :

- a lead gamma shielding;
- a solid neutron shielding;
- a fireproof shielding.

Inside the tight containment vessel, the cooling is ensured by natural convection between the needles and allows the thermal transfer towards the cavity walls.


The external cooling is performed by air and also by natural convection. All the control and safety parts are sunk in the packaging body, at its upper part and protected against impact and fire by a shockabsorbent hood.

- A thermal transier by conduction and an air cooling by natural convection allow the guarantee of the package cooling whatever the circumstances and the orientation of the package may be, after an accident.
- A fireproof shielding allows the appreciable limitation of the temperatures inside the package in case of real fire outbreak and thus the avoidance of the neutron and gamma shielding destruction and, especially, a slackening of the primary heat-carrier fluid, even when filtrated to the atmosphere.
- A solid neutron shielding guarantees its existence whithout restriction under normal conditions of use as well as under accidental conditions.
- Inside the package, there is a part for centering the containers, constituted by an aluminium alloy block (ALPAX), comprising five pits, each capable of taking three containersvertically, i.e. a total of 15 containersper cask.
- Several devices and fittings ensure the safety on the highway and allow the performance of each necessary control when loading and unloading the packaging :
- safety valves provided to avoid dangerous overpressures in case of an accident worse than in legal tests;
- self-closing connections coupled with taps used for possible draining of the internal fluid;
- self-closing cornections also coupled with taps fitted for sampling and pressure measurements;
- temperature registration for thermal equilibrium control;
- openings for cap-tightness checking.

Technical characteristics

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Total weight (loaded packaging) : approwimately 95 T
Total length (with shock-absorbing cover) : approximately : 6 m
Diameter : 2.5 m
Cavity length : 4.59 m
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3.3. Reception and Intermediate Storage
3.3.1. Cask Receiving Hall

Coming from the reprocessing plant, the cask, on its road trailer, is brought into this hall which is equipped with :

- a door isolating the access corridor,
- a device for washing by jet to rid the cask of the dirt resulting irom its transport by road.

The washing wastes are recovered in a tank installed in a trench below the low level of the hall,

- a 3 T travelling crane for raising and handing the shockabsorbers fitted to the ends of the cask.

The receiving hall communicates through an aperture in the ceiling with the air-lock which permits the connection without communication of ambiance with the monitoring and unloading room.

The air-lock is equipped with a 120 T crane which manoeuvers the cask by means of a handling-frame from the receiving room to the monitoring room.

Two motorized trap-doors permit the closing of the transfer apertures.
3.3.2. Monitoring and Unloading Room

This room is equipped with :

- a self-propelling fork-lift truck which transports the cask from the transfer aperture with the air-lock to the door of the unloading cell, where, by means of its lifting movement, it connects the cask of the cell,
-. a crane with nacelle which allows the operators to unbolt the cask lid, to place a handling-frame on it for in-cell handing and to connect the piping of the activity monitoring circuit,
- a 3 T travelling crane for moving the lid handing-frame,
- an activity monitoring device connected to the cask by flexible piping, comprising an air circulation device with filter, an installation for measuring the activity of the air.
3.3.3. Unloading Cel1 This is made up of concrete walls to ensure the biological shielding of the operators.

4 working stations, each equipped with a shielded viewing window and a pair of "Master-Slave" telemanipulators which permit operations to be carried out inside the cell.

These stations are served by a working area situated in level
$-48$.

Inside the cell, handing is performed by a travelling crane with a capacity of 8 tons. When it is not being used or in case of maintenance, the crane rests in an intervention cell which communicates with the unloading cell by an opening closed by a shielded door.

The introduction of the containersinto the cell is made through an aperture in the bottom of the cell, which communicates between this and the unloading room.

The cask, placed on its transport cart, is connected to this aperture, which is closed by a shielded trap-door when not in use.

After the connection of the cask, its lid is removed by the cell crane and placed in an obturable box to avoid the propagation of the possible contamination of the lid.

The containers are removed from the cask and placed in an air cooled storage area by the cell crane which is equipped with a special grapple for this purpose.

After unloading the cask and closing the transfer aperture, each container, as it is evacuated, is placed in a receptacle situated to the right of the evacuation aperture in the cell roof. This aperture is closed by a shielded trap.

If the internal activity monitoring of the transport cask indicates that one or more containers have become contaminated following deterioration, all the containers contained in the cask are introduced, through the unloading cell into a recanning cell.

### 3.3.4. Recanning cell

This has the function of putting contaminating containers, or those which have become accidentally contaminated, into a second container, in order to avoid the contamination of the temporary storage.

It comprises two working stations, at level - 55, each equipped with a shielded window and with a pair of "Master-Slave" telemanipulators.

2 apertures which can be closed by a shielded door make the connection with the unloading cell. One of these apertures is used for the introduction of the containers, the other for their evacuation.

A shielded transfer device permits the introduction of the empty containers and their lids into the cell from the intervention zone.

These containers are placed on a merry-go-round which carries them under the introduction aperture where they receive a container. At the next position of the merry-go-round, a lid is placed on the full container and automatically welded by a plasma torch device. After this operation, the container is brought under the evacuation position to be introduced into the unloading cell by means of the crane of the latter.

A trap-door situated in the ceiling of the conditioning cell permits the transfer of the cell equipment to the intervention cell for maintenance operations.

### 3.3.5. Container Transfer Cask

This is used to transport a container from the unloading cell to the temporary storage and from there to the encapsulation cell. Of a type similar to the AVM cask of CEA Marcoule, it is constituted by a lead wall lined internally with stainless steel, and can be closed at the bottom by a shielded door. An electric hoist with a cyclical grapple at the end of its cable carries out the gripping of the container from the top of the unloading or of the temporary stage.

An autonomous ventilation decice constituted by a fan and filters at the air inlet and outlet permits the internal cooling of the cask when a container is inside it.

The cask is connected to the transfer apertures of the unloading cell temporary stage by means of a movable shielded drawer.
3.3.6. Temporary Storage

This is constituted by two parallel concrete trench, the walls of which are covered with a carbon steel lining. On a metal frame on each of these are set out 150 steel pits which can each hold ten containers vertically, ine. a total for the two trenches of 3,000 containers. Spaces are provided batween the tubes to ensure the cooling ventilation of the pits.

Each trench is closed horizontally by a concrete slab in which, at right angles to each pit, there is an aperture closed by a plug. This plug has a rod in its centre, which is terminated on the interior side of the storage by a disc which closes the pit concerned when it is empty so as to prevent the cooling air from passing through it. This disc can be opened from outside the storage so as to permit the ventilation of a pit containing one or more canisters.

A 45 -ton crane carries out handing operations above the trenches.

When a canister is placed in a pit, a shielded drawer is drawn across the pit selected, as also is a cask for putting the pit plug into. After the plug has been removed, the shielded drawer is closed again and the cask with the plug is taken away and replaced by the cask with the container After the containerhas been put down into the pit, the reverse procedure is carried out.

At the end of the storage is situated the storage ventilation installation area. Two ducts, one for the blowing and the other for air extraction, link the ventilation room to the storage.
3.4. Encapsulation
3.4.1. Encapsulation cel1

The function of this is encapsulation of the containers coming from the temporary storage prior to final storage.

It consists of a concrete cell comprising 5 working-points, each one provided with a shielded window fitting and a pair of "MasterSlave" telemanipulators.

Handling within the cell is carried out by an 8-ton travelling crane, which is kept, when not in use, in the intervention cell communicating with the conditioning cell by means of a passage closed off by a shielded door.

The containers, coming from the temporary storage under shielded cask, are put in from an aperture, which can be closed off by a shielded trap, situated on the upper part of the cell at level

- 40 .

The container is placed in the cell on a cart which serves :

- The device for placing the canister on the container, constituted by a titanium jacket shielded with lead. These canisters are introduced into the cell through an aperture in the ceiling of the latter.
- The lead-pouring station at which the container in its canister is turned in order to allow an intervening pouring with molten lead. This lead is introduced into the container through a duct supplied from a lead-melter situated on the upper part of the cell.
- The lead-machining station, where, after cooling, the lead in the canister is surface-levelled by a machining-head in order to facilitate the putting on of the lid.
- The canister-lid weiding station. After the lid has been placed on the canister with a remote-control handler, an automatic welding head, type CEA-AVM, makes a tight weld by plasma torch around the lid-canister junction.
- After welding, the canister is returned and taken, by the cell crane, to the welding radiography point. At this point, the callister is set in rotation in order to pass the line of the weld in front of an X-ray emitter, which acts upon a film placed in a device which allows it to be inserted and removed from outside the cell.

After monitoring of the welding, the canister is taken up again by the crane and placed in a tight lid casing, in which the tightness monitoring is carried out in helium and in a vacuum.

After tightness monitoring, the canister is placed in the evacuation point from where it is transferred into a lead cask, which connects with a transfer shielded drawer placed in the ceiling of the cell. This cask, put on a motorized cart, is used to transfer the titanium canister to the final storage.
3.4.2. Titanium Canister Drawing $\mathrm{N}^{\circ}$ D OO14

The canister into which the containeris put for its final storage is constituted by a shell 6 mm thick, made of titanium in order to provide it with excellent resistance to corrosion by brackish water.

The interior of the drum is lined with a lead wall with a thickness of 50 mm , in the centre of which the canister rests.

The space between the lead 1 ining and the container is then filled with molten lead, which ensures, after cooling, a total lead shielding thickness of 100 mm and contact between the walls so as to :
－ensure good thermal dissipation，
－reduce the activity on contact and the radiolysis which can damage the titanium drum，
－increase the protection against corrosion and the mechanical strength under a water pressure of 50 bars．

After pouring the lead a titanium lid is welded on to the canister in a tight manner．

For handing，the canister comprises a gripping head identical to that of the containerso as to provide the same type of grip．

Characteristics of the Canister


Completion of the Canister

In order to obtain the proper adhesion to the titanium wall，the lead will be poured in portions into the canister in a horizontal position．After the cooling of one portion，the container will be rotated in accordance with an angular value，so that another portion can be poured．After pouring all the portions，the internal cavity will be put on the final side for machining．

To increase resistance to corrosion，the lead used will be pure．

Another method can be used. It consists to cast the lead against the titanium enveloppe by extrusion process and to close the canister with a prefabricated lid.
3.4.3. Intervention Cell

Situated between the unloading and the encapsulation cells at the same level, it comprises a concrete-walled enclosure, consisting of two working-points provided with shielding windows fittings and master-slave manipulators.

Its function is :

- To facilitate interventions on the cell crane.
- The possible decontamination of equipment coming from the three cells with which it communicates.
- To facilitate the transfer of this equipment to the outside.
- To perform maintenance or small repair operations on the cell equipment.

The intervention cell has on top of it an enclosure made of sheet steel in which there is a 8 ton crane. This enclosure communicates with the intervention cell by means of a shielded trap-door and its function is to maintain isolation with the handing hall when the intervention cell is open. It facilitates manual treating, for example under vinyl, of the equipment coming from the cell.

It also receives the mechanisms for opening the shielded doors of the passages between the unloading and intervention cells and of the latter with the treatment cell.

It also receives the drums of the supply cables for the cell cranes.
1.3. AUXILIARY SYSTEM

### 3.5.1. Temporary Storage Ventilation

3.5.1.1. General Description and Main Charactexistics

The cooling of the temporary storage is performed by a ventilation installation.
3.5.1.2. Introduction of Air

- This will be achieved by suction.
- Taking into consideration an air temperature of $+20^{\circ} \mathrm{C}$ at the inlet and $+80^{\circ} \mathrm{C}$ at the outlet, the necessary flowrate will be

$$
150,000 \mathrm{~m} 3 / \mathrm{h} \text { per storage. }
$$

- The collection of air is performed on the surface in a suction chamber fitted with external air intakes with a surface of approximately 21 m 2 .
- Before introduction into the storage this air is firstly filtered over a bank of filters with $90 \%$ efficiency for particles of 5 .
- The filtration system can be by-passed by a system of ducts and dampers.
- The connecting headers will have a minimum interior surface of 4.5 . m2.
3.5.1.3. Air Exhaust

At the storage outlet the air will be filtered over an absolute filtration device with $99.99 \%$ DOP efficiency.

The circulation of air and therfore of the exhaust will be performed by two half-flowrate fans, $N^{\circ} 1$ and $N^{\circ} 2$,

$$
\text { i.e. } 75,000 \mathrm{~m} 3 / \mathrm{h} .
$$

These fans will have a total manometric height of approwimately 35 mb corresponding to the following estimated needs :

5 nb for the suction
15 mb for the storage
7 mb for the filtration
8 mb for the exhaust

35 mb

Power absorbed approximately 150 Kw per fan.

These two fans will be doubled by a stand-by fan with identical characteristics, $N^{\circ} 3$.

The set of 3 fans and the filters will be laid out on level -- 30 m .

A by-pass circuit with an automatic damper will, in the event of a defect of the filters or of the fans, permit the by-passing of the filter and fan system.

At ground level and in the vicinity of the release stack, a second stand-by fan, $N^{o} 4$ connected to the exhaust circuit by a duct and damper device will permit the exhaust of $65 \%$ of the normal flowrate.

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### 3.3.2. Ventilation of Rooms

3.5.2.1. Purpose of the Installation

For safety reasons :

- A positive and negative pressure cascade is maintained according to the activity and contamination present or likely to be present in the various rooms, in order to prevent any non-controlled exchange of gas between them. These rooms are classified into four areas, according to the degree of activity, likely to be present in them.
- The exhaust air passes through absolute filters before being released to the outside.

For Reasons of Comfort :

- Pleasant atmospheric conditions are maintained whatever the season in rooms where personnel are present (heating, humidification, chilling of in-blown air).
3.5.2.2. Conditions Maintained

Temperatures_and_moistures_in the rooms

In accordance with the usual local conditions.

Pressure_and_negative pressure_in the rooms

White area : over pressurized
Green area : - 3 mm WG $P$ - 5 mm WG
Amber area : - 6 mm WG $\mathrm{P}-8 \mathrm{~mm} \mathrm{WG}$
Red area : -48 mm WG $P \quad-57 \mathrm{~mm}$ WG

Glove boxes and enclosures damp and dry channel : -25 mm WC

```
Average air renewals
White area : \(4 \mathrm{rn} / \mathrm{h}\)
Green area : \(6 \mathrm{rn} / \mathrm{h}\)
Amber area : \(6 \mathrm{rn} / \mathrm{h}\)
Red area: \(10 \mathrm{rn} / \mathrm{h}\)
```


### 3.5.3. Operating Principle

The air admitted into the building will be previously filtered and treated so as to obtain the requisite thermal conditions in the rooms. Maximum use will be made of the possibilities of transfers going from one area to another more active area.

In the red cells, the air will be compulsorily introduced by transfer, no blowing being authorized in order to avoid all risk of accidental overpressure.

These transfers will be performed from the intervention areas.

According to the particular conditions in the cell, the introductions of air will be equipped (if applicable) with transfer screws, filters, dampers and non-return valves.

The air exhaust will be performed by centrifugal fans, doubled on stand-by.

The air will be filtered prior to release to atmosphere.

Absolute filtration : $99.99 \%$ DOP
2 filter stages for the red areas
1 filter stage for the other areas.

```
1.6. OPERATIONS OF FACILITY
```

3.6.1. Cask Receiving llall
- Arrival of transport cask on its trailer.
- External washing of cask.
- Dismounting of shock-absorbers.
- Transfer of cask by 120 T crane into the monitoring and unloading
room.
3.6.2. Monitoring and Unloading Room
- Lowering of cask on to the transfer cart.
- Connection of flexible piping, circulation of air for monitoring
the internal activity of the cask.
- After monitoring, disconnection of the piping.
- Undoing of the cask lid fastening screw (bolt).
- Positioning of lid-handling frame.
- Moving of cask under the transfer aperture of the unloading cell.
3.6.3. Unloading Cell
- After connection of cask on transfer aperture in cell bottom, opening of cask and placing of its lid in its tight receptacle.

- Unloading of containers and placing them in the ventilated storage of the cell.
- Closing of cask and returning it to the receiving hall.
- Positioning in the cell of a container at the evacuation post.
- Positioning on the cell of the containertransfer cask.
- Opening of the shielded drawer of the cell.
- Lowering of the cask grapple and gripping of the container, which is then taken up into the cask.
- Transfer of the cask to the temporary storage.


### 3.6.4. Recanning Cell

In the case where the cask contains contaminated containers, these are taken from the cask by means of the unloading cell crane and transferred directly through the unloading cell into the conditioning cell after opening of the container introduction aperture.

- Placing of eachcontainer in the empty container brought by the merry-go-round right under the transfer aperture.
- Closing of the unloading cell and conditioning cell transfer apertures and return of the cask to the receiving hall.
- Lowering and welding of the lid of each full container.
- Opening of transfer aperture to the unloading cell and, with the crane in this cell, taking up of each container on the merry-go-round.
- When arriving in the unloading cell, each container is subjected to an external contamination control.
- In the case where traces of contamination are ascertained, the container is placed in a tight box where it is subjected to external washing at high pressure.
- In the opposite case, the container is put in the ventilated storage of the cell, to await its transfer towards the temporary storage.

```
3.6.5. Temporary Storage
    - Positioning by crane of shielded drawer above the selected
    storage pit.
    - Positioning on the drawer of the cask for the pit plug.
    - Opening of gates of drawer and cask and introduction of plug
    into cask.
    - Closing of drawer and placing on it of container transfer cask.
    - Opening of gates of drawer and cask and lowering of container into
        pit by the cask hoist.
    - Return of transfer cask to unloading cell.
    - Replacement of plug on pit from out of its cask.
    - Adjustment of pit ventilation valve.
3.6.6. Encapsulation Cell
After decay in the temporary storage, the containers are taken out and brought into the final conditioning cell. To this effect, the following operations are carried out :
- The shielded drawer and the plug cask are placed on the temporary storage pit and the pit is opened.
- The cask for transfer to temporary storage is placed above the pit and the container is introduced into the cask.
- The activity of the air ventilating the cask is monitored.
```

- If contamination is detected, the cask is directed towards the unloading cell where the containers will be treated in the recanning cell like the contaminated canisters before temporary storage. The transfer cask will then be decontaminated before being put back into service.
- The container transfer cask is then set down on the final conditioning cell and the container is lowered on to the in-cell cart.
- The titanium container is lowered by the in-cell crane.
- The container is tipped over.
- The cart is brought to the lead-pouring station and the final filling is carried out.
- After cooling, the cart is brought to the lead surface-grinding station.
- The cart is brought to the container-lid welding station and the lid is welded after being positioned by telemanipulator. After welding, the container is tipped over.
- The container is picked up by the cell crane and is set down at the welding inspection station.
- When this inspection has been completed, the container is taken by crane to the helium-tightness inspection station, station for checking the tightness by helium.
- After this operation, the cask is introduced into the cask for transfer to final storage.
3.9. Construction time schedule and costs

```
3.9.1. Evaluation of construction cost
    The following evaluation are included :
    .-. Special equipments in the cell.
    - Cell wal1s.
    - Fuel shielding casks.
    - Over head cranes.
    - Ventilation.
    - Electricity.
    - Radioprotection equipment.
    - Engineering and site monitoring.
```

    Are excluded :
    - The excavation cost.
    - The civil work other than cell and storage wall.
    - Lift, mobile truck.
    - Consumable material, titanium canister, lead, etc..
    3.9.1.1. Reception


39.500 .000 FF

Engineering and erection . . . . . .
12.000.000 FF
51.500.000 FF

### 3.9.1.3. Encapsulation - Canister transport in gallery



```
3.9.2. Time schedule
    1. Reception and intermediate storage
    Design - 12 months
    Fabrication {
    Erection - }12\mathrm{ months
    Tests - 12 months
                                    total 4 to 5 years
```

    2. Encapsulation
    Design - 12 months
    Fabrication - 18 months
    Erection - 10 months
    Tests - 6 months
    3.9.3. Operating personnel

```
Working in one shift/day.
```

 containers to intermediate storage :

1 Manager
2 Clerks
2 Crane-drivers
4 Operators
1 Electrician
1 Mecanician
3 Health physicists

Total 14 persons

- $2^{\text {nd }}$ phase : Retrieval of containers from intermediate storage, encapsulation and final disposal :

1 Manager
2 Clerks
1 Crane-driver
2 Truck-drivers
5 Operators
1 Electrician
1 Mecanician
3 Health physicists

Total 16 persons

## IV. FINAL STORAGE

4.1. General Description of Facilities

After conditioning in a titanium container, each glass container is directed towards the final storage situated 500 m below the rock surface.

The final storage is constituted by 41 parallel tunnels, 1 km long, each separated from the next by a distance of 25 m , i.e. an area of 1 km2.

This area is traversed by transport tunnels which separate it into four storage divisions which will be constructed in six stages. Each tunnel comprises storage pits, at intervals of four metres, into which the canisters will be put.

These pits, which number 9000 in total, are five metres deep and 1 m in diameter.

A shaft elevator performs the liaison between the temporary storage and encapsulation installation and the final storage.
4.2.K.B.S.
4.3. Transport to Place of Final Storage

The transport of the canister is performed by a shielded transfer cask mounted on a motorized cart which runs on rails.

The canister transfer cask is equipped with a hoist for introducing and setting down the canister, and with a lifting complementary shielding on the lower part which ensure a complementary shielding around the storage pit in the event of intervention on the cask while a canister is being set down. This skirt is raised up against the cask by an electrical jack device while the cask is being moved.

The cart's power supply comes from batteries. It runs on rails and in the final storage it is directed from the transport tunnel towards the storage tunnel by a turntable. Movable stops ensure that that it is positioned above the selected storage pit.

When all the pits in one tunnel are filled with canisters, the running rails are dismantled and reassembled in the next tunnel, as is the turntable. The rails in the transport tunnel are extended.

### 4.4. Final disposal

### 4.4.1. Placing of Canisters

The canister transfer cask is brought on its cart above the selected pit and positioned by the centering stops.

The lifting complementary shielding is placed on the pit, the cask trap is opened and the canister is lowered into the pit. At about 50 cm from the setting-down level, the hoist is stopped for a few minutes to stabilize the canister and then the descent continues until it is set down.

The cyclical grapple is then raised, as is the complementary shielding.

### 4.4.2. Scaling of Final Storage

When the canister has been set down, the pit is filled with a mixture of sand and bentonite.

The filliny is performed from a shielded hood mounted on a cart which runs on the same rails as the transfer cart.

This hood is constituted by a lead jacket comprising the bentonite tank and the injection pump on the outside and, on the inside, an annular pouring device equipped with 3 vibrating needles.

This device can be lowered into the pit around the canister and gradually raised again as the filling progresses, at the same time being endowed with an alternating angular motion of $120^{\circ}$ to increase the efficiency of the vibrating needles.

When it is positioned above the pit, the filling cart is connected to the transfer cask before the latter is moved from its unloading position.

After connecting the two carts, the whole assembly is moved so as to bring the filling cart over the pit. This operation is intended to prevent the breaking of the shielding before the pit completely filled.

When the filling cart is over the pit, the transfer cask can be brought back to the encapsulation cell.

After the filling with bentonite, the filling cart is moved towards the next pit and a granitecover is placed on the pit which has just been filled.
5.1. HANDLING
5.1.1. Cask Handling Cranes

- To ensure handling safety, the travelling cranes for the casks are classified in a mechanical group of the European Federation of Handing which is one higher in relation to the class of actual use and requirements.
- They are designed to have very slow loaded speeds so as to avoid serious dynamic reactions.
- The speed reducing gear of the winches is of the irreversible type and the braking devices are doubled.
- The lifting cables are subjected to frequent examinations and are changed periodically.
- The handing frames are constructed of steel selected from grades of 80 to $90 \mathrm{~kg} / \mathrm{mm} 2$ and calculated for a workrate of $6 \mathrm{~kg} / \mathrm{mm} 2$.
- The links between frames and casks are fixed positevely after being put in place.

The strength of the casks and the equipment in the handling zones is calculated to take into account the possible dropping of a cask in relation to the maximum height necessary for handling it.

### 5.1.2. In-Cell Cranes

These cranes, which operate in a hostile medium, are designed in such a way as to be able to terminate the operation in progress in the event of an incident and to return under their own power into an intervention zone where they can be repaired.

For this purpose, all the movements of raising - lowering, travelling and direction are performed at two speeds, one slow and the other fast, obtained from an independant source.

The electricity supply comes from two cables, one for the HV motors and one for the LV motors. In this way an operation performed at a chosen speed can be terminated, in the event of a power-supply incident or a motor breakdown, at the other speed, the crane then being immediately brought back into the intervention cell to be repaired.

### 5.2. IN-CELL OPERATIONS

All the in-cell equipment is designed in order to :

- Complete the operating sequence by manual means in case of failure of the motorized controls.
- To remove from or introduce into the cells the in-cell equipment without the necessity of the operators having to enter.
- Permit its maintenance and small repairs by remote control.
- Alarm or stopping devices are provided in the machines operating cycle or for operating sequences to give warning of any operating defect and to prevent the deterioration of the equipment.


### 5.3.1. Description

Each storage cell, capable of 3000 containers, includes an independent ventilation facility.

According to the calculation (annex 9), the nominal air flowrate is $150000 \mathrm{Nm} 3 / \mathrm{h}$ and the total normal differential pressure is 35 mbars.

The total flowrate is given by two identical fans, $75000 \mathrm{Nm} 3 / \mathrm{h}$ each and a third stand-by fan in parallel set-up in an underground room.

For the most conservative case, i.e. 3000 canisters regularly loaded in 10 years, the total thermal power is 3000 Kw . With such conditions, the temperature difference of air between entrance and exit is $60^{\circ} \mathrm{C}$. So the exit is at $80^{\circ} \mathrm{C}$ when the atmosphere is at $20^{\circ} \mathrm{C}$.

A supplementary emergency fan is set-up at the ground level. The characteristics of it are the same as for the others.

Automatic check-gates are put in the ducts for by-passing the filters and by-passing the supplementary fan.
5.3.2. Normal operating

Two fans are operating in parallel in the underground chamber. The total flowrate is $150000 \mathrm{Nm} 3 / \mathrm{h}$,
the exit air temperature : $80^{\circ} \mathrm{C}$
the upper canister surface temperature : $86,4^{\circ} \mathrm{C}$

### 5.3.3. First emergency

A fan is stopped, the stand-by third fan is started, the ope rating conditions are the same.
5.3.4. Second emergency

Two fans are stopped, the third one only is operating. The flowrate falls to $65 \%=100000 \mathrm{Nm} 3 / \mathrm{h}$ 。

The temperature of exit air gradually rises within 40 hours up to $112^{\circ} \mathrm{C}$ when normal loading is completed or $66^{\circ} \mathrm{C}$ for half total thermal power.

The upper canister surface temperature to $123^{\circ} \mathrm{C}$ when the normal loading is completed or $72^{\circ} \mathrm{C}$ for half total thermal power.
5.3.5. Third emergency

The three fans of the underground chamber are stopped.

It is possible to start the emergency ground level fan and have access to the others.

The underground filters are by-passed.

The flowrate is approximately $100000 \mathrm{Nm} 3 / \mathrm{h}$ and the conditions are the same as preceding.
5.3.6. Total break-down

In that case, all by-pass gates are open. The filters and blowers are out off the circuit and the air is circulating in natural convection.

The exit air temperature gradually rises within 40 hours up to $336^{\circ} \mathrm{C}$, when normal loading is completed.

The upper canister surface temperature is $370^{\circ} \mathrm{C}$. The center temperature in glass is $434^{\circ} \mathrm{C}$.

In such a case, the ambiance temperature could be high in the underground chamber. It will be necessary to start the upper fan before having access to the chamber of the others.

This very exceptional event could cause some damages to the concrete of the storage cell. So this cell is to be provided with steel coating and frame in order to be sufficiently strong to withstand the above temperature during the necessary time to repair the ventilation system.
5.4. NUCLEAR SAFETY OF THE ACTIVE FACTLITY

> 5.4.1. Nature of the risk

In this facility are received HLW incorporated in glass enclosed in sealed canisters. These canisters are normally transferred to the intermediate storage and, after at least 30 years cooling, conditioned in special containers and lowered to the final storage.

The nuclear safety will take into account the risk of irradiation through $\chi$ and neutron radiations by proper shieldings.

The risk of contamination is not normally present as the canisters are tight, and have been decontaminated in the vitrification facility before stripping.

Remark

> The transportation of containers comply with the safety regulations of radioactive materials (IAEA regulations). After exceptional events, it could happen that a container is ruptured. In this event some glass dust could exceptionally be found in the transport cask. This activity is likely to be fixed and not contaminating, but if some dust may be entrained by air sweeping during monitoring of the cask, this activity would be detected on filtexs. In that case washing of the cask and recanning of the containers could be performed, giving way to liquid effluents with some risk of contamination in the cells specially foreseen for this operations. This risk has to be taken into consideration for the cells where damaged containers can be transferred.

### 5.4.2. Hot cells facility

In these cells the operations are remotely controlled from behind biological shielding.

The main risk is coming from irradiation due to the rather high activity of the glass handled.

The construction of the buildings, rooms and cells that form the facility will comply with the principles for shielding and prevention against the action of ionizing radiation defined in "Norm ISO R 1710 " of 1970 for radioactive facilities.

- Irradiation

The biological shielding will be designed for the activity calculated as per Annex2.

The equivalent dose rate will not exceed :

- $0.25 \mathrm{mrem} / \mathrm{h}$ in zone 1
- $2.5 \mathrm{mrem} / \mathrm{h}$ in zone 2 (operating zone)
(Where operators are frequently present, the dose rate is limited to about $0.25 \mathrm{mrem} / \mathrm{h}$ ).

The transfer of the glass canisters or final containers in shielded casks is a routine operation. It must be checked that the equivalent dose rate at the surface of the cask remains lower than $2.5 \mathrm{mrem} / \mathrm{h}$.

- Contamination

The containment of activity will be ensured by the cascade of negative pressures of the ventilation system explained in ch. 3.5.2.

In normal operation, there will be no contamination inside the cells. In case of accidental contamination of canisters by glass dust, monitoring of the air extracted from the cell (zone 4) will be provided by means of $\alpha$ or $\beta$ sniffers with monitoring instruments installed in the operating zone (zone 2). Linings and equipment inside the cell shall be easy to decontaminate.

Telemanipulation means will make mechanical decontaminations possible and will be used to remove contaminated equipment in the intervention cell where it will be packaged for take-off.

Outside the cells, the surface contamination of rooms and equipment will have to be less than the following values :

- Non-fixed contamination : Maximum admissible limit is :
$2 \times 10^{-6} \mu \mathrm{Ci} / \mathrm{cm} 2$ for $\alpha$ contamination
$3 \times 10^{-4} \mu \mathrm{Ci} / \mathrm{cm} 2$ for $\beta$ contamination
- Fixed contamination : Maximum admissible limit is :
$2 \times 10^{-5} \mu \mathrm{Ci} / \mathrm{cm} 2$ for $\alpha$ contamination
$10^{-3} \mu \mathrm{Ci} / \mathrm{cm} 2$ for $\beta$ contamination

These latter limits are taken into account, only when they are more restraining than those for irradiation.
5.4.3. Liquid wastes

Liquid wastes arise from the rinsing of the transport cask in case of detection of non-fixed contamination or from the washing of contaminated canisters.

They are received in a tank installed in a separate cell located on the lower level of the active facility and equipped with a drip-tray. These wastes will be of medium-level activity (in the range of $1 \mathrm{Ci} \beta / \mathrm{m} 3)$. They have to be transferred periodically by means of a syphoning device in a shielded flask for evacuation to the ground surface. Monitoring of their activity will be made before transfer.

Production of these wastes will be very occasional and their volume can be estimated to some cubic meters per year during reception of glass canisters.

### 5.4.4. Intermediate Storage of Glass Canisters

This storage receives sealed glass canisters or canisters formerly recanned in hot cells in case of external contamination. There is normally no risk of contamination of the cooling air during the 30 years of storage. The experience of the PIVER glass canisters storage in Marcoule (about 5 million curies) has shown no detectable contamination of this air even from non sealed canisters. Nevertheless absolute filters are provided on the outlet of the air before release to the stack for additional safety.

Multiple redundancy has been foreseen on the ventilation system as described in ch. 5.3 . In the most improbable case of power failure of the four fans cooling will be achieved by natural convection of air, and it has been shown in the calculations of Annex 2 that the temperature of air will increase in a time of about 50 hours to $330^{\circ} \mathrm{C}$ in the worst case i.e. at the end of the filling of the storage.

In this event the maximum temperature of the hottest glass will remain below the crystallization temperature of $600^{\circ} \mathrm{C}$, and far below the melting point. Then no risk of contamination of the cooling air will appear. But if these conditions are maintained for too long, some damages (cracks) will result in the concrete structure of the storage, not-withstanding the thermal shield.

The atmosphere of the intervention zone, even in the hereabove conditions, will remain at a higher pressure than the cooling air, due to the natural draft of the stack.

### 5.4.5. Organization of the Safety

Safety monitoring wiil be the duty of a Health Physics team. Health Physics will be permanently attached to the facility for checking of the instruments. They will attend all the intervention operations to survey the conditions of work.
5.5. CONTAINMENT OF THE RADIOACTIVITY

This containment is ensured by the successive barriers :

1/ The glass itself
2/ The stainless-steel container
3/ The lead shielding
4/ The titanium canister
5/ The sand + bentonite filling of the pit
6/ The granite and the underground water

After completion of loading of the underground storage and backfilling of the tunnels with bentonite + quartz sand, the final storage will be exposed to ground water.
5.5.1. The titanium enveloppe will constitute the main barrier to corrosion during the first period of decay of the HLW. This enveloppe will have a thickness of 6 mm . The welding of the cover will be carefully executed and controlled in a hot cell. The excellent resistance of titanium in ground water can be appreciated on the following basis (1) :
(1). DE GELAS

Résistance à la ./.. (Informations Chimie - No 92 - Janvier 1971)

- DE GELAS

Principes d'emploi du titane en génie chimique
(Informations chimie - No 136 - Octobre 1974)

- Titanium metal is protected by the formation in air of a protecting film of titanium oxide, which is remarkably stable, except for very low pH values.
- The rupture voltage of this film in chlorine solution is relatively high, of the order of 10 Volts.
- Welding properties of titanium are good under inert gas (argon). Corrosion around weld is excluded by using an alloy of Titanium.
- Pitting corrosion does not occur even in saline water below $140^{\circ} \mathrm{C}$. The temperature of the titanium surface in saline water will not exceed $80^{\circ} \mathrm{C}$.
- Stress corrosion does not occur in saline water.
- General corrosion is very low in saline water and difficult to measure.

A sample immersed during 18 years in sea water did not show appreciable corrosion, except a slight decoloration.

According to tests performed in U.S.A. with samples immersed in sea water, the corrosion thickness calculated over 1000 years would be less than $25 \mu$.

- The chemical composition, the pH and the oxyde-reduction potential of the saline water can be modified by the effect of radiolysis due to the $\gamma$ and neutron flux coming out of the glass. These flux
are minimized by the lead shielding of 10 cm thickness and will be of the order of $0.1 \mathrm{rad} / \mathrm{S}$ at the time of disposal, after 40 years cooling. According to calculations made by Atomenergi, the radiolysis process on the saline water will very rapidly (after 10 hours) come to an equilibrium state corresponding to a concentration of some ppb (parties per billion) of radiolyzed elements in the saline water. This would have a negligible effect on the corrosion.

From the foregoing, it can be concluded that the corrosion resistance of the titanium enveloppe will last more than 1000 years, at which time the concentration of radioactive fission products will be very low and their potential health hazard become negligible. Then will remain only the radioactivity due to the actinides.
5.5.2. The lead shielding will have 3 main functions :

- The first function, already mentioned in 5.5.1. is to minimize the radiolysis effect. The thickness of 10 cm has been chosen, so as to reduce the $\gamma$ flux outside the container to the order to magnitude of the neutron flux (see Annex $n^{\circ}$ 5).
- The lead will give the container the mechanical resistance necessary to withstand the hydraulic pressure of the underground water, i.e. 50 bars.

We have calculated that the crushing pressure of the lead will be 70 bars (reckoned with an empty canister). As an indicative figure, the buckling pressure is about 700 bars.

- Lastly, the lead shielding will provide an additional resistance to corrosion. The resistance of lead to ground water is good, but not comparable to that of titanium, and is more sensitive to the
presence of oxygen in the solution. In the absence of oxygen, the corrosion at $50^{\circ} \mathrm{C}$ has been estimated to be less than 2 mm in 1000 years, but with unfavourable conditions the corrosion could be 10 to 100 times higher, hence a lifetime of 500 to 5000 years.

Therefore, we can consider the lead shielding as a corrosion barrier additional to the titanium enveloppe or at least as safety barrier in case of failure of the titanium enveloppe.
5.5.3. The stainless-steelcontainer is a barrier to possible contamination in the intermediate storage. On the other hand, its resistance to corrosion by saline water is very weak, and it cannot be considered as a reliable barrier in the final storage.
5.5.4. The glass will ultimately come into contact with the underground water after destruction of the titanium and the lead, i.e. after a time delay of the order of 1000 years. At that time, the heat power of the glass canister will be reduced to 33 watts, so its temperature will have decreased in the range of $20^{\circ} \mathrm{C}-30^{\circ} \mathrm{C}$. The release of radioisotopes of actinides in the underground water will be limited by the leachability of the glass.

The leaching rate of the radioisotopes from the glass by sea water or natural water are not significantly different. They are not affected by the internal irradiation ( $\alpha, \beta$ or $\gamma$ ) during the decay of the HLW.

The measured rates at $25^{\circ} \mathrm{C}$ are the following for the actinides :

$$
10^{-7} \text { to } 10^{-9} \mathrm{~g} / \mathrm{cm} 2 \mathrm{x} \text { day }
$$

The value for the plutonium is in the lower range, the value for americium in the higher range.

The surface of the glass considered as a solid block in the container is about $2 \times 10^{4} \mathrm{~cm} 2$. Actually, the surface of the glass block can be increased by a factor of 5 during the transportation due to possible cracks. The leachable surface will then be taken as :

$$
S=10^{5} \mathrm{~cm}
$$

The leaching rate of an element $M$ is defined by the formule :

$$
L=\frac{a}{A} \cdot \frac{P}{S} \mathrm{~g} \mathrm{~cm}^{-2} \mathrm{day}^{-1}
$$

where a is the weight of $M$ leached out per day, A is the total weight of $M$ in the canister, $P$ is the weight of the glass in the canister and $S$ the effective surface as defined above. $P$ is equal to $420 \mathrm{~kg}(420000 \mathrm{~g})$. Taking the value $\mathrm{L}=2 \times 10^{-7} \mathrm{~g} \mathrm{~cm}^{-2}$ day for the actinides, which is probable for Pu and probably conservative for the other actinides, the quantity of $M$ leached out per day should be :

$$
a=\frac{\mathrm{A} .2 \times 10^{-7} \times 10^{-5}}{420000}=4.76 \times 10^{-8} \mathrm{~A} \mathrm{~g}
$$

According to appendix 1, the weights of the actinides 100 years after removal from the reactor are :

| Pu | $:$ | 99.2 | g |
| :--- | :---: | :---: | :---: |
| Am | $:$ | 458 | g |
| Cm | $:$ | 2.83 | g |
| Np | $:$ | 541 | g |

Using these values in the formula above gives the following release of radioisotope from the canister :

$$
\begin{aligned}
& \mathrm{Pu}: c \cdot 7 \mu \mathrm{~g} / \text { day } \\
& \mathrm{Am}: 22 \\
& \mathrm{Cm}: \\
& \mathrm{Cm} \\
& \mathrm{~Np}: 0.13 \mu \mathrm{~g} / \text { day } \\
& \mathrm{g} / \text { day } \\
&
\end{aligned}
$$

The further movement of these actinides in the underground system and their eventual release to the surface will depend on several parameters, among others the water flow. Calculations of those movements are beyond the scope of this study.
5.5.5. The bentonite has a very high ion exchange capacity. Tests are presently performed in France by the CEA and the B.R.G.M. (Bureau de Recherches Géologiques et Minières) under contract of the EEC. First results have shown that bentonite is one of the best materials for absorption of actinides by ion-exchange. The absorption takes place very rapidly by exchange of the sodium included in the molecule of bentonite with the radioisotope, and the absorption is not reversible. The distribution coefficient between bentonite and the solution is higher than $10^{4}$ for plutonium, neptunium, americium, and also cerium.

The absorption capacity of the bentonite is of the order of :

$$
60 \mathrm{Ca}^{++} \mathrm{meg} / 100 \mathrm{~g}
$$

On this basis, it has been calculated (see Annex 7) that a quantity of 826 kg of bentonite would be sufficient to absorb all the leached radioisotopes released.

This has to be compared with the volume of filling of the pit which is approximately 5 m 3 . The weight of this filling will be about 10 tonnes, of which $10 \%$ i.e. 1000 kg will be constituted by bentonite.

## A N N E X 1

## COMPOSITION OF HLW CONTAINER

The weight of elements included in one 150 liters container corresponding to 1 equivalent metric ton of heavy metal charged in the reactor, are calculated by the Origen program.

The assumed nuclear fuel charged in the reactor is:

- PWR enriched 3.3\%
- Regular irradiation in 1100 days at $30 \mathrm{MW} / \mathrm{MT}$
- Total burn-up $33000 \mathrm{MWD} / \mathrm{MT}$


## RESULTS

1.1 - Actinides - Storage period
1.2 - Actinides - After reprocessing at 10 years
1.3 - Fission products - Storage period
1.4 - Fission products - After reprocessing at 10 years.

[^0]POWER $=30.00 \mathrm{MH}$, BURNUP $=33000$. MWD, FLUX $=2.93 E+9 \mathrm{~N} / \mathrm{CM} * 2-$ SEC
ELEMENT CONCENTRATIONS, GKAMS
BASIS = MT OF HEAVY PETAL CHARGED TO HEACTOR

|  |  |  |  |  |  |  |  |  |  |  | REPRO |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Charge | DISCHARGE | 30. | 60. | 120. D | 18 C |  |  |  |  |  |  |
| HE | 0.0 | $2.32 \mathrm{E}-01$ | $2.56 \mathrm{E}-01$ | 2.78F-01 | 3.15E-01 | $3.47 \mathrm{E}-01$ | $3.84 \mathrm{E}-01$ | $4.13 \mathrm{~F}-01$ | $4.91 \mathrm{E}-01$ | $5.47 \mathrm{E}-01$ | $\begin{aligned} & 3652=0 \\ & 9.04 t=01 \end{aligned}$ | $\begin{array}{r} 0 \\ +00 \end{array}$ |
| TL | 0.0 | $1.86 \mathrm{E}-12$ | $1.99 \mathrm{E}-12$ | $2.17 \mathrm{~F}-12$ | $2.58 \mathrm{~F}-12$ | $2.97 \mathrm{E}-1$ ? | $3.58 \mathrm{E}-12$ | $4.21 \mathrm{E}-12$ | 6.80 E | 9. |  |  |
| P日 | 0.0 | $4.61 E-07$ | $5.14 \mathrm{E}-07$ | $5.711-07$ | $6.98 \mathrm{E}-07$ | $8.50 \mathrm{E}-07$ | 1.12E-06 | $1.43 \mathrm{E}-0 \mathrm{o}$ | $3.22 \mathrm{E}-96$ | 5.86E-00 | $4.16 \mathrm{~F}-05$ |  |
| 81 | 0.0 | 1.66E-10 | $1.79 \mathrm{E}-10$ | 1.92F-10 | $2.17 \mathrm{E}-10$ | $2.41 \mathrm{E}-10$ | 2.78t-10 | 3.17E-10 | 4.76E-10 | 6.35F-10 | 1.39E-09 | 1.94E-09 |
| PO | 0.0 | 8.97E-14 | $9.73 \mathrm{E}-14$ | $1.05 \mathrm{E}-13$ | 1.22E-13 | $1.41 \mathrm{E}-13$ | $1.72 \mathrm{E}-13$ | $2.07 \mathrm{E}-13$ | $3.87 \mathrm{E}-13$ | 6. $46 \mathrm{E}-13$ | 6.92E-12 | 3.56E-11 |
| AT RN | 0.0 | 1.19E-19 | $6.90 \mathrm{E}-20$ | $3.56 t-20$ | 2.21E-20 | $2.16 \mathrm{E}-20$ | $2.22 \mathrm{E}-2 \mathrm{C}$ | $2.29 \mathrm{E}-20$ | $2.59 \mathrm{E}-20$ | 2.88F-20 | 5.14E-20 | 9.09E-20 |
| RN | 0.0 | 1.75E-12 | $1.87 \mathrm{E}-12$ | $2.03 \mathrm{E}-12$ | $2.40 \mathrm{E}-12$ | 2.76E-12 | $3.31 \mathrm{E}-12$ | 3.88E-12 | $6.24 \mathrm{E}-12$ | $8.00 \mathrm{E}-1$ | 1.90 E | 2.37E-11 |
| FR R A | 0.0 | 5 | 1.01E-15 | $7.26 \mathrm{E}-10$ | $6.47 \mathrm{E}-16$ | 6.89E-10 | $7.63 \mathrm{E}-16$ | 8.37e-16 | 1-13E-1 | 1.43E-15 | $3.37 \mathrm{E}-15$ | $5.95 E-15$ |
| AC | 0.0 | 2. <br> $1.39 E-08$ <br> $.36-08$ | ? $.72 \mathrm{E}-08$ $3.42 \mathrm{E}-08$ | 2.87E-08 | $3.27 E-08$ | 3.57E-0と | 4.12E-08 | $4.699 \mathrm{E}-08$ | 7.20E-08 | $1.00 \mathrm{~F}-07$ | 3.41E-07 | $8.85 \mathrm{E}-07$ |
| Th | 3.0 | $1.16 \mathrm{E}-33$ | $1.20 \mathrm{E}-03$ | $1.24 \mathrm{E}-03$ | $1.31 \mathrm{E}-03$ | 1.39E-03 | 2.10t-08 | 2.361-08 | $3.37 \mathrm{E}-08$ | $4-36 E-08$ | $1.09 \mathrm{E}-07$ | $1.92 \mathrm{E}-\mathrm{U} 7$ |
| PA | 0.0 | $5.29 \mathrm{E}-04$ | S. $31 \mathrm{E}-04$ | $5.32 \mathrm{E}-0.4$ | 5.33E-04 | ). 3 . 35 E-04 | 5.37E-04 | $1.63 \mathrm{E}-03$ $5.38 \mathrm{E}-04$ | $2.10 E-03$ $5.46 E-04$ | $2.58 E-03$ $5.54 t-04$ | $5.94 E-03$ $6.07 \mathrm{E}-04$ | $1.11 \mathrm{E}-02$ $0.83 \mathrm{E}-04$ |
| U | $9.99 \mathrm{E}+05$ | $9.55 \mathrm{E}+05$ | $9.55 E+05$ | $9.55 E+05$ | $9.55 E+0.5$ | $9.55 \mathrm{E}+05$ | $9.55 \mathrm{E}+05$ | $9.55 \mathrm{E}+05$ | $9.55 \mathrm{E}+05$ | $9.55 E+05$ | $9.55 \mathrm{E}+05$ | 9.85E + C5 |
| NP | 0.0 | $5.53 E+02$ | $4.82 E+02$ | $4.82 \mathrm{E}+02$ | $4.82 \mathrm{E}+02$ | $4.82 E+02$ | $4.82 \mathrm{E}+02$ | $4.82 \mathrm{E}+02$ | $4.82 \mathrm{E}+02$ | $4.83 E+02$ | $4.86 \mathrm{E}+02$ | $4.94 \mathrm{E}+02$ |
| PU | 0.0 | $8.90 E+03$ | $8.98 \mathrm{E}+03$ | $8.98 \mathrm{E}+03$ | $8.97 \mathrm{E}+03$ | $8.97 \mathrm{E}+03$ | $8.96 \mathrm{E}+03$ | $8.94 \mathrm{E}+03$ | $8.90 \mathrm{~F}+03$ | $8.86 E+03$ | $8.60 E+03$ | $8.34 \mathrm{E}+03$ |
| AM | 0.0 0.0 | $1.21 E+02$ $4.28 E+01$ | $1.25 E+02$ $4.16 E+01$ | $1.29 E+02$ | $1.37 \mathrm{E}+02$ | $1.45 E+02$ | $1.57 \mathrm{E}+02$ | $1.69 E+02$ | $2.15 E+02$ | 2.58f+02 | $5.08 \mathrm{EF}+02$ | 7.43E+02 |
| CM | 0.0 | $4.28 E+01$ $2.19 E-06$ | $4.16 \mathrm{E}+01$ | 4. $1.94 \mathrm{E}+01$ | $3.84 \mathrm{E}+01$ | $3.68 E+01$ | $3.50 \mathrm{E}+\mathrm{C1}$ | $3.37 E+01$ | $3.09 \mathrm{~F}+01$ | $2.95 E+01$ | 2. $30 \mathrm{E}+01$ | $1.64 \mathrm{E}+\mathrm{C1}$ |
| BK | 0.0 | 2-19E-06 | $2.05 \mathrm{E}-06$ | $1.92 \mathrm{E}-06$ | $1.68 \mathrm{E}-00$ | 1.47E-06 | $1.20 E-00$ | $9.78 \mathrm{E}-07$ | $4.37 E-07$ | 1.95E-07 | $6.93 t-10$ | 2.18t-13 |
| C F | 0.0 | $9.03 \mathrm{E}-07$ | $1.04 \mathrm{E}-06$ | 1.17E-06 | $1.40 \mathrm{E}-00$ | $1.61 \mathrm{E}-06$ | 7.87E-06 | $2.07 \mathrm{E}-06$ | 2.58E-06 | 2.79E-06 | 2.82E-06 | ?.68E-06 |
| ES | 0.0 | 1.22E-10 | $1.02 \mathrm{E}-10$ | $5.55 \mathrm{E}-11$ | $1.12 \mathrm{E}-11$ | $1.77 \mathrm{t}-12$ | $9.76 \mathrm{E}-14$ | $5.12 \mathrm{E}-15$ | $2.88 \mathrm{E}-20$ | 1.42E-25 | 0.0 |  |
| TOTALS | $9.99 \mathrm{E}+05$ | $9.65 E+05$ | $9.65 \mathrm{E}+05$ | $9.6 .5 E+05$ | $9.65 E+05$ | $9.65 E+05$ | $9.65 \mathrm{E}+75$ | $9.65 \mathrm{E}+05$ | $9.65 \mathrm{E}+05$ | $9.65 E+05$ | $9.65 \mathrm{~F}+05$ | 9.6 - ${ }^{\text {a }}$ (05 |

reference pur equilierium fuel cycle -- waste decay times
POWER $=30.00 \mathrm{MW}$. BURNUP $=33000$. MWD. FLUX $=2.93 \mathrm{E}+13 \mathrm{~N} / \mathrm{CM} * 2-$ SEC

| REPRO |  |  |  | ELEMENT CONCENTRATIONS, GRAMS |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | charge | - | 5. Y | 10. | 30. Y | 90. Y | 190. |  |  |  |  |  |
| He | 0.0 | $9.04 \mathrm{E}-01$ | $1.02 \mathrm{E}+00$ | $1.13 \mathrm{E}+00$ | $1.47 \mathrm{E}+00$ | $2.20 \mathrm{E}+00$ | $3.14 \mathrm{E}+00$ | $5.19 \mathrm{E}+00$ | 7.06E+00 | 1990. $8.42 F+00$ | 4990. $9.30 E+C O$ | $9990 . \mathrm{Y}$ |
| TL | 0.0 | $1.98 \mathrm{E}-11$ | $3.47 E-12$ | $8.04 \mathrm{E}-13$ | $2.86 E-13$ | 2.43E-93 | $1.88 \mathrm{E}-13$ | 1.59E-13 | 7.06E +00 $1.68 E-13$ | $8.42 F+00$ $2.00 E-13$ | $9.30 E+C 0$ $3-61 F-13$ | $1.01 \mathrm{E}+01$ |
| P3 | 0.0 | $4.16 E-25$ | $5.68 \mathrm{E}-05$ | 5.96t-05 | $6.15 \mathrm{E}-0.5$ | $6.48 \mathrm{E}-05$ | 6.83E-05 | $1.59 E-13$ $7.49 \mathrm{E}-05$ | $1.68 E-13$ $8.93 E-05$ | $2.00 E-13$ $1.47 E-04$ | $3.61 E-13$ $6.52 t-04$ | $7.96 E-13$ $2.87 E-03$ |
| $\pm 1$ | 0.0 | 1.39E-09 | 6.61E-1U | 0.91t-10 | $1.60 E-09$ | $1.05 \mathrm{E}-08$ | 7-52E-08 | 1.27E-06 | $1.12 \mathrm{E}-05$ | $9.90 \mathrm{E}-05$ |  | $2.87 E-0.3$ $1.27 E-02$ |
| PO | 0.0 0.0 | $6.92 E-12$ $5.14 E-20$ | $1.71 \mathrm{E}-11$ $5.26 \mathrm{~F}-20$ | $3.21 E-11$ $5.60 E-20$ | 1.27E-10 | 5.06E-10 | 1.45t-09 | 5.13E-09 | 1.31E-08 | $3.63 E-08$ | 1.37E-07 |  |
| RN | 0.0 | 1.90E-11 | $5.78 \mathrm{E}-12$ | $3.60 E-20$ $4.62 E-12$ | $9.14 \mathrm{E}-20$ $9.00 \mathrm{~F}-12$ | $4.14 E-19$ $2.40 E-11$ | $1.72 \mathrm{E}-18$ $5.08 \mathrm{E}-11$ | 1.21E-17 | $5.36 E-17$ | 2.35E-16 | $1.52 \mathrm{E}-15$ | 5.53E-15 |
| FR | 0.0 | $3.37 E-15$ | $4.48 \mathrm{E}-15$ | 5.44E-15 | $8.31 \mathrm{E}-15$ | 1.37t-14 | 2.02E-94 | 1. $21 E E-10$ $1.21 E-13$ | $3-83 \mathrm{E}-10$ | 1.06E-09 | $4.00 \mathrm{E}-\mathrm{CS}$ | $9.09 \mathrm{~F}-09$ |
| RA | 0.0 | $3.41 \mathrm{E}-07$ | $4.47 \mathrm{t}-07$ | $6.22 \mathrm{E}-07$ | $1.385-06$ | 3.72E-06 | $2.02 E-14$ $7.90 t-06$ | $1.21 E-13$ $2.34 E-05$ | $5.02 E-12$ $5.47 E-05$ | $2.17 E-12$ $1.05 E-04$ | $1.40 E-11$ $0.23 E-0.4$ | $5.07 \mathrm{E}-11$ |
| AC | 0.0 | $1.09 t-07$ | 1.50E-07 | $1.855^{-1}-07$ | 2.80E-07 | $3.73 \mathrm{t}-07$ | $7.901-06$ $3.921-07$ | 2. $34 E-05$ $4.07 E-07$ | $5.47 E-05$ $4.29 E-07$ | $1.65 E-04$ $4.75 E-07$ | $0.23 E-\cap 4$ $6.38 E-\cap 7$ | $1.51 E-03$ $8.71 E-07$ |
| TH | 0.0 | 5.99E-03 | $5.99 \mathrm{E}-03$ | $6.01 \mathrm{E}-03$ | $6.12 \mathrm{E}-03$ | $6.52 \mathrm{E}-03$ | 7.35E-03 | 1.05E-02 | 1.63E-02 | 4.90E-02 |  | $7.71 \varepsilon-07$ $1.59 E-01$ |
| PA | 0.0 $9.99 E+0.5$ | $0.07 t-04$ $9.55 t+03$ | $6.08 E-04$ $9.55 E+03$ | 6.08E-04 | $6.10 t-04$ |  | 6.23E-0.4 | 6.46E-04 | 6.81E-04 | 7.48E-04 | $9.42 \mathrm{E}-04$ | $1.39 E-01$ $1.28 E-03$ |
| NP | 0.0 | $4.56 E+02$ | $4.89 E+02$ | $4.92 \mathrm{E}+02$ | $9.55 E+03$ $5.05 E+02$ | $9.55 E+03$ $5.41 E+02$ | $9.55 E+03$ $5.94 E+02$ | 9-56E+03 | $9.50 E+03$ | $9.50 E+03$ | $9.58 \mathrm{E}+03$ | $9.60 \mathrm{E}+03$ |
| PU | 0.0 | $8.60 E+01$ | $8.82 \mathrm{E}+01$ | $9.00 \mathrm{E}+01$ | $9.49 \mathrm{E}+0$ i | $9.92 \mathrm{E}+01$ | $5.94 E+02$ $9.96 E+01$ | 7.10E+02 | $8.13 E+02$ | $8.81 \varepsilon+02$ | $8.97 \mathrm{E}+0$ ? | $8.96 E+02$ |
| AM | 0.0 | $5.08 t+02$ | $5.06 E+02$ | $5.04 \mathrm{E}+02$ | $4.93 \mathrm{E}+02$ | $4.58 E+02$ | $4.03 \mathrm{E}+02$ | $2.80 E+02$ | 1.01E+02 | $1.03 E+02$ | 1.06E+C2 | 1.06E +02 |
| CM | 0.0 | $2.30 \mathrm{E}+31$ | $1.94 \mathrm{E}+01$ | $1.64 E+01$ | 8.79E+00 | $2.83 E+00$ | $2.16 \mathrm{~F}+00$ | $2.09 \mathrm{E}+00$ | -13Et | $9.67 \mathrm{t}+\mathrm{Cl}$ | $6.06 \mathrm{~F}+01$ | $3.84 \mathrm{E}+01$ |
| 3 K | 0.0 | $6.93 \mathrm{E}-10$ | $1.23 \mathrm{E}-11$ | 2.18E-13 | 4.58E-20 | $2.42 \mathrm{E}-20$ | $2.41 \mathrm{E}-20$ |  | -94E+00 | 1.82F+u0 | $1.40 \mathrm{E}+0 \mathrm{CO}$ | 8.99E-01 |
| CF | 0.0 | 2.82E-06 | 2.74E-06 | $2.68 \mathrm{E}-06$ | $2.50 \mathrm{E}-00$ | $2.20 \mathrm{E}-06$ | 1.82E-06 | $2.38 E-20$ | 2.33E-20 | 2. $24 \mathrm{E}-20$ | 1.991-20 | 1.63E-20 |
| totals | $9.94 E+05$ | $1.07 t+04$ | $1.07 \mathrm{E}+04$ | $1.07 \mathrm{E}+04$ | $1.07 E+04$ | $1.07 \mathrm{E}+04$ | $1.07 \mathrm{E}+04$ | $1.07 \mathrm{E}+04$ | 4.27E-07 | $8.61 \mathrm{E}-08$ | 3.89F-09 | 7.99E-11 |

POWER $=30.00 \mathrm{MW}$, EURNUP $=33000$. MWD. FLUX $=2.93 E+13 \mathrm{~N} / \mathrm{CM} * * 2-$ SEC
ELEMENT CONCENTRATIONS, GRAMS
BASIS = MT OF HEAVY METAL CHARGED TO REACTOR

|  | charge | discharge | 30. D | 60. 0 | 120. D | 182. D | 274. | 365. | 73 | 1096. D | 36 | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| H | 0.0 | 7.32E-02 | 7.28E-02 | 7.25E-02 | 7.18E-02 | 7.11E-02 | 7.01E-02 | 6.91E-02 | $6.54 \mathrm{E}-02$ | $6.18 \mathrm{E}-02$ | $4.16 \mathrm{E}-02$ | $2.37 \mathrm{E}-02$ |
| LN | 0.0 | 2.17E-05 | $4.73 \mathrm{E}-10$ | $1.03 \mathrm{E}-14$ | 4.91E-24 | 1.14E-33 | 5.81E-48 | 4.22E-62 | 0.0 | 0.0 | 0.0 | 0.0 |
| GA | 0.0 | $1.20 \mathrm{E}-05$ | $2.06 \mathrm{E}-10$ | $4.49 \mathrm{E}-15$ | 2.14E-24 | $4.98 \mathrm{E}-34$ | 2. $53 \mathrm{E}-48$ | 1.84E-62 | 0.0 | 0.0 | 0.0 | 0.0 |
| GE | 0.0 | 3.71E-01 | 3.71E-01 | 3.71E-01 | 3.71E-01 | 3.71E-01 | 3.71E-01 | $3.71 \mathrm{E}-01$ | 3.71E-01 | 3.71E-01 | 3.71E-01 | 3.71E-01 |
| AS | 0.0 | $8.70 \mathrm{e}-02$ | 8.46E-02 | 8.46E-02 | 8.46E-02 | 8.46E-02 | 8.46t-02 | 8.46E-02 | 8.46t-02 | $8.46 E-02$ | 8.46E-02 | 8.46E-02 |
| SE | 0.0 | 5.20E+01 | $5.20 E+01$ | $5.20 \mathrm{E}+01$ | 5-20E+01 | $5.20 E+01$ | $5.20 E+01$ | $5.20 E+01$ | $5.20 E+01$ | $5.20 E+01$ | $5.20 E+C 1$ | $5.20 \mathrm{E}+01$ |
| 日R | 0.0 | $1.53 \mathrm{E}+01$ | $1.53 E+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | 1.53E+01 | $1.53 E+01$ | $1.33 \mathrm{E}+01$ |
| KR | 0.0 | $3.75 \mathrm{E}+02$ | $3.75 \mathrm{E}+02$ | $3.74 t+02$ | $3.74 \mathrm{E}+02$ | $3.74 \mathrm{E}+02$ | 3.73E 02 | $3.73 E+02$ | $3.71 E+02$ | $3.70 E+02$ | $3.61 E+02$ | 3. $54 \mathrm{E}+02$ |
| RB | 0.0 | $3.33 \mathrm{~F}+02$ | $3.33 \mathrm{E}+\mathrm{C2}$ | $3.33 E+02$ | $3.34 \mathrm{E}+02$ | $3.34 \mathrm{E}+02$ | $3.35 E+C 2$ | $3.35 E+02$ | $3.37 E+02$ | $3.38 \mathrm{E}+0$ ? | $3.47 \mathrm{E}+02$ | $3.54 \mathrm{E}+02$ |
| Sk | 0.0 | $9.24 \mathrm{E}+02$ | $9.14 \mathrm{E}+02$ | $9.07 E+02$ | $8.99 \mathrm{E}+02$ | $8.94 \mathrm{E}+02$ | $8.89 \mathrm{E}+02$ | $8.85 \mathrm{E}+02$ | $8.72 \mathrm{E}+02$ | $8.59 \mathrm{E}+0$ ? | $7.78 \mathrm{E}+02$ | $0.84 E+02$ |
| $Y$ | 0.0 | $4.80 E+02$ | $4.77 E+02$ | $4.74 \mathrm{E}+02$ | $4.71 E+02$ | $4.69 E+02$ | $4.67 E+02$ | $4.67 E+02$ | $4.67 \mathrm{E}+02$ | $4.67 \mathrm{E}+02$ | $4.67 \mathrm{E}+02$ | $4.67 \mathrm{E}+02$ |
| ZR | 0.0 | $3.68 \mathrm{E}+03$ | $3.68 \mathrm{E}+03$ | $3.67 E+03$ | $3.67 E+03$ | $3.67 \mathrm{E}+03$ | $3.67 E+03$ | $3.67 E+03$ | $3.68 E+03$ | 3.69E+03 | $3.78 \mathrm{E}+03$ | 3.87 E +03 |
| NB | 0.0 | $3.53 E+31$ | $3.27 E+01$ | 2.76E+01 | 1.73E+01 | $9.78 \mathrm{E}+00$ | $3.90 \mathrm{E}+00$ | $1.52 E+00$ | $3.35 \mathrm{E}-02$ | $2.21 \mathrm{E}-03$ | 3.94F-U3 | ?.34E-0. |
| 190 | 0.0 | 3.37E+03 | $3.39 E+03$ | $3.40 E+03$ | $3.43 E+03$ | $3.45 E+03$ | $3.46 E+03$ | $3.46 E+03$ | $3.47 E+03$ | $3.47 \mathrm{E}+03$ | $3.47 E+03$ | 3.4 ? $\mathrm{E}+\mathrm{O}$ ? |
| TC | 0.0 | $8.38 \mathrm{E}+02$ | $8.41 E+02$ | $8.41 E+02$ | $8.41 E+02$ | $8.41 E+02$ | $8.41 E+02$ | $8.41 \mathrm{E}+02$ | $8.41 \mathrm{E}+02$ | $8.41 \mathrm{E}+02$ | 8.41E+02 | $8.41 \mathrm{E}+02$ |
| Ru | 0.0 | $2.35 E+03$ | $2.32 E+03$ | $2.31 \mathrm{E}+03$ | $2.28 \mathrm{E}+03$ | $2.26 E+03$ | $2.24 E+03$ | 2.23F+03 | $2.19 \mathrm{~F}+03$ | $2.17 E+03$ | 2.15E+03 | C. $15 \mathrm{E} E+03$ |
| RH | 0.0 | 3.51E+02 | $3.65 \mathrm{~F}+02$ | $3.74 \mathrm{E}+02$ | 3.83E+02 | 3.86E+02 | $3.88 \mathrm{E}+02$ | 3.88E +02 | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{~F}+02$ | $3.88 \mathrm{E}+42$ |
| PD | 0.0 | $1.25 E+03$ | $1.26 E+03$ | $1.27 E+03$ | $1.28 E+03$ | $1.30 \mathrm{E}+03$ | $1.32 E+03$ | $1.33 E+03$ | 1.37E +03 | $1.39 \mathrm{~F}+03$ | $1.41 \mathrm{E}+03$ | $1.41 E+03$ |
| M G | 0.0 | $0.08 \mathrm{E}+01$ | $6.06 E+01$ | $6.05 E+01$ | $6.04 \mathrm{E}+01$ | $6.03 \mathrm{E}+01$ | $6.02 \mathrm{E}+01$ | 6.01E+01 | $5.99 \mathrm{E}+01$ | $5.99 E+01$ | $5.98 \mathrm{E}+01$ | $5.98 E+01$ |
| C | 0.0 | $8.31 \mathrm{E}+01$ | $8.34 E+01$ | $8.35 E+01$ | $8.36 E+01$ | $8.37 \mathrm{E}+01$ | $8.38 \mathrm{E}+01$ | $8.39 E+01$ | $8.41 \mathrm{E} \geqslant 01$ | $8.41 \mathrm{E}+01$ | $8.42 \mathrm{E}+01$ | $8.41 \mathrm{E}+01$ |
| 1 N | 3.0 | $1.20 E+00$ | 1. $22 \mathrm{E}+00$ | $1.22 \mathrm{E}+00$ | 1.23E+00 | $1.23 \mathrm{E}+00$ | $1.23 E+00$ | $1.23 E+00$ | $1.24 E+00$ | 1.24E*00 | $1.25 E+00$ | $1.26 E+00$ |
| SN | 0.0 | $5.23 E+01$ | $5.20 E+01$ | $5.18 \mathrm{E}+01$ | 5.16E+01 | $5.15 \mathrm{E}+01$ | S. $13 \mathrm{E}+01$ | 5-12E+01 | $5.11 \mathrm{E}+01$ | $5.11 \mathrm{E}+01$ | $5.11 \mathrm{E}+01$ | $5.11 E+01$ |
| SB | 0.0 | $1.76 t+01$ | $1.74 \mathrm{E}+01$ | $1.73 \mathrm{E}+09$ | 1-72E+01 | $1.70 E+01$ | $1.67 E+01$ | $1.64 E+01$ | $1.51 \mathrm{E}+01$ | 1.39F+61 | 1.07E+01 | $1.01 \mathrm{E}+01$ |
| TE | 0.0 | $5.71 \mathrm{E}+02$ | $5.65 E+02$ | $5.65 \mathrm{E}+02$ | $5.64 E+02$ | $5-64 E+02$ | $5.65 E+02$ | $5.05 \mathrm{E}+02$ | $5.66 E+02$ | $5.67 \mathrm{E}+02$ | $5.71 \mathrm{E}+02$ | $5.71 \mathrm{E}+02$ |
| 1 | 0.0 | $2.75 t+02$ | $2.68 E+07$ | $2.68 \mathrm{E}+02$ | $2.69 E+02$ | $2.69 E+02$ | $2.69 E+02$ | $2.70 E+02$ | $2.70 \mathrm{E}+02$ | $2.705+02$ | $2.70 E+02$ | $2.70 E+02$ |
| XE | 0.0 | $5.43 \mathrm{E}+03$ | $5.43 \mathrm{E}+03$ | $5.43 \mathrm{E}+0.3$ | $5.43 E+03$ | $5.43 \mathrm{E}+03$ | $5.43 \mathrm{E}+03$ | $5.43 E+03$ | $5.43 \mathrm{E}+03$ | $5.43 \mathrm{E}+03$ | $5.43 \mathrm{~F}+03$ | $5.43 E+03$ |
| CS | 0.0 | $2.75 \mathrm{E}+03$ | $2.76 E+03$ | $2.75 \mathrm{E}+03$ | $2.73 E+03$ | $2.72 E+03$ | $2.70 E+03$ | $2.68 \mathrm{E}+03$ | $2.61 E+03$ | $2.56 \mathrm{E}+03$ | 2.32E*03 | $2.12 E+03$ |
| BA | 0.0 | $1.37 \mathrm{E}+03$ | $1.36 E+03$ | $1.37 E+03$ | $1.38 \mathrm{E}+03$ | $1.40 \mathrm{E}+03$ | $1.42 \mathrm{E}+03$ | $1.44 \mathrm{E}+03$ | $1.50 \mathrm{E}+03$ | $1.56 E+03$ | $1.79 \mathrm{E}+03$ | $2.00 E+03$ |
| LA | 0.0 | $1.27 E+03$ | $1.27 E+03$ | 1. $27 \mathrm{E}+03$ | $1.27 E+03$ | 1.27E+03 | 1.27E+03 | $1.27 \mathrm{E}+03$ | $1.27 \mathrm{E}+03$ | $1.27 E+03$ | $1.27 \mathrm{E}+03$ | $1.27 t+03$ |
| CE | 0.0 | $2.86 E+33$ | $2.82 E+03$ | $2.79 E+03$ | 2.74E+03 | $2.706+03$ | $2.66 E+03$ | 2.62E+03 | $2.54 \mathrm{E}+\mathrm{C3}$ | $2.50 E+03$ | $2.48 \mathrm{E}+03$ | $2.48 \mathrm{E}+03$ |
| PR | 0.0 | $1.17 E+03$ | 1.18E+03 | $1.19 \mathrm{E}+03$ | $1.20 F+03$ | $1.20 E+03$ | 1.20E+03 | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | 1.20E+03 | $1.20 E+03$ |
| ND | 0.0 | $3.76 E+03$ | $3.79 \mathrm{E}+03$ | $3.82 E+03$ | $3.86 E+03$ | $3.89 E+03$ | 3.94E*03 | $3.97 \mathrm{E}+03$ | $4.06 E+03$ | $4.09 E+03$ | $4.12 \mathrm{E}+03$ | $4.12 \mathrm{E}+03$ |
| PM | 0.0 | 1.15E+02 | $1.15 \mathrm{E}+02$ | 1. $13 \mathrm{E}+02$ | $1.08 \mathrm{E}+02$ | $1.03 \mathrm{t}+02$ | $9.63 \mathrm{E}+01$ | $9.02 \mathrm{E}+01$ | $6.92 E+01$ | $5.31 \mathrm{~F}+01$ | $8.34 \mathrm{E}+00$ | 5.91E-01 |
| SM | 0.0 | $7.91 \mathrm{E}+02$ | $7.96 E+02$ | $7.99 E+02$ | $8.04 \mathrm{E}+02$ | $8.09 E+0 ?$ | $8.16 E+02$ | $8.22 \mathrm{E}+02$ | $8.42 E+02$ | $8.58 \mathrm{E}+02$ | $9.00 \mathrm{E}+02$ | $4.05 \mathrm{C}+02$ |
| EU | 0.0 | $1.88 \mathrm{E}+02$ | $1.86 E+02$ | $1.85 \mathrm{E}+02$ | 1.84E+02 | $1.83 E+02$ | 1.82E+02 | 1.81E+02 | $1.79 \mathrm{E}+02$ | $1.76 E+02$ | 1-66E+02 | $1.58 \mathrm{E}+02$ |
| GD | 0.0 | $9.60 E+01$ | $1.00 E+02$ | $1.01 \mathrm{E}+02$ | $1.02 \mathrm{E}+02$ | $1.03 E+02$ | $1.04 \mathrm{E}+02$ | $1.05 \mathrm{E}+02$ | $1.08 \mathrm{E}+02$ | $1.11 \mathrm{E}+\mathrm{Cz}$ | $1.23 E+02$ | $1.35 E+0$ ? |
| TB | 0.0 | $1.90 E+00$ | 1.87E+00 | $1.84 \mathrm{E}+00$ | $1.82 \mathrm{E}+00$ | $1.80 \mathrm{E}+00$ | $1.79 \mathrm{t}+00$ | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | $1.78 t+00$ | $1.78 \mathrm{E}+00$ |
| Dr | 0.0 | 9.86E-01 | $1.02 \mathrm{E}+00$ | $1.04 \mathrm{E}+00$ | 1.07E+00 | $1.09 E+00$ | 1.10E+00 | $1.10 \mathrm{E}+00$ | $1.10 E+00$ | $1.11 \mathrm{E}+00$ | 1.11E+00 | $1.11 \mathrm{E}+00$ |
| H0 | 0.0 | $1.02 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | 1.01E-01 | $1.01 \mathrm{E}-01$ | 1.01E-01 | 1.01E-01 |
| ER | 0.0 | 3.06E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 | 3.08E-02 |
| totals | 0.0 | 3.49E*04 | $3.49 E+04$ | $3.49 E+04$ | $3.49 E+04$ | $3.49 E+04$ | 3.49E+04 | $3.49 \mathrm{E}+04$ | $3.49 \mathrm{E}+04$ | $3.49 \mathrm{E}+04$ | $3.49 \mathrm{E}+04$ | $3.49 E+04$ |

REference phr equilierium fuel cycle -- waste decay times
POWER $=30.00 \mathrm{MW}$, BURNUF $=33000$. MWD, FLUX $=2.93 E+13 \mathrm{~N} / \mathrm{CM} * 2-$ SEC

| PEPRO |  |  |  |  |  | heavy | CH | TO REA |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | CHARGE | 0\#94***E | 5. r | 10. Y | 30. Y | 90. Y | 190. Y | 490. Y | 990 | 1990 | 4990. | 9990. |
| H | O. 0 | $4.16 \mathrm{E}-32$ | $3.14 E-02$ | 2.37E-02 | 7.68E-03 | <-61E-0. | $9.33 \mathrm{E}-07$ | $4.24 E-14$ | 2.45E-26 | 0.0 | 0.0 | 0.0 |
| GE | 0.0 | 3.71E-31 | 3.71E-01 | 3.71E-01 | 3.71E-01 | 3.71E-01 | $3.71 \mathrm{E}-01$ | 3-71E-01 | $3.71 \mathrm{E}-01$ | 3.71E-01 | $3.71 t-01$ | 3.71E-01 |
| AS | 0.0 | 8.46E-02 | 8.46E-02 | $8.46 \mathrm{E}-02$ | $8.46 E-02$ | $8.46 \mathrm{t}-02$ | $8.46 \mathrm{E}-02$ | $8.46 \mathrm{E}-02$ | 8.46F-02 | $8.46 \mathrm{E}-02$ | $8.46 \mathrm{E}-02$ | $8.46 \mathrm{E}-02$ |
| SE | 0.0 | $5.20 E+01$ | $5.20 E+01$ | $5.20 \mathrm{E}+01$ | $5.20 E+01$ | $5.20 \mathrm{E}+01$ | $5.20 E+01$ | $5.20 F+01$ | $5.20 \mathrm{E}+01$ | $5.19 \mathrm{~F}+01$ | $5.17 \mathrm{E}+01$ | 5.15E*01 |
| BR | 4.0 | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 \mathrm{E}+01$ | $1.53 E+01$ | $1.53 t+01$ | $1.54 E+01$ | $1.54 E+01$ | $1.55 \mathrm{E}+01$ | $1.56 \mathrm{E}+01$ | $7.59 \mathrm{E}+01$ |
| RB | 0.0 | $3.47 \mathrm{E}+02$ | 3-47E+02 | $3.47 \mathrm{E}+02$ | $3.47 \mathrm{E}+02$ | $3.47 \mathrm{E}+02$ | $3-47 E+02$ | $3.47 E+02$ | $3.47 E+02$ | $3.475+02$ | $3.47 \mathrm{E}+02$ | $3.47 E+02$ |
| SR | 0.0 | $7.78 E+02$ | 7.28E+02 | $6.84 E+02$ | $5.54 t+02$ | $3.96 E+02$ | $3.53 E+02$ | $3.5 U E+02$ | $3.50 \mathrm{E}+02$ | $3.50 \mathrm{E}+02$ | $3.50 \mathrm{E}+02$ | $3.50 E+02$ |
| $Y$ | 0.0 | $4.67 E+02$ | $4.67 \mathrm{E}+02$ | $4.67 \mathrm{E}+0$ ? | $4.67 \mathrm{E}+02$ | $4.06 t+02$ | $4.66 E+02$ | $4-66 E+02$ | $4.60 \mathrm{E}+02$ | $4.66 E+02$ | $4.66 E+02$ | - 6 EE +02 |
| ZR | 0.0 | $3.78 \mathrm{E}+03$ | 3.82E +03 | 3.87F+03 | $4.0 \cup \mathrm{E}+03$ | $4.16 E+03$ | $4.20 E+03$ | $4.20 E+03$ | $4.20 E+03$ | $4.20 E+013$ | $4.20 \mathrm{E}+03$ | $4.20 E+03$ |
| NB | 0.0 | 3.94E-03 | $5.64 E-03$ | 7.34F-03 | $1.41 \mathrm{E}-02$ | $3.45 \mathrm{E}-02$ | $6.85 \mathrm{E}-02$ | 1.70E-01 | 3.40E-01 | $6.80 \mathrm{E}-01$ | $1.70 \mathrm{E}+00$ | $3.30 \mathrm{E}+00$ |
| Y0 | 0.0 | $3.47 t+33$ | 3.47E+03 | $3.47 \mathrm{E}+03$ | $3.47 \mathrm{E}+03$ | $3.47 t+03$ | $3.47 t+03$ | $3.47 \mathrm{E}+03$ | $3.47 \mathrm{E}+03$ | $3.47 E+03$ | $3.47 E+03$ | $3.4 \mathrm{PF}+03$ |
| 10 | 0.4 | 8.41E+02 | $8.41 E+02$ | $8.41 E+02$ | 8.41E+02 | $8.41 E+02$ | 8.41E 402 | $8.40 \mathrm{E}+02$ | $8.39 E+02$ | $8.30 E+02$ | $8.28 \mathrm{E}+0$ ? | $8.14 \mathrm{E}+02$ |
| RU | 0.0 | 2.15E+03 | $2.15 E+03$ | $2.15 E+03$ | $2.15 E+03$ | <-15E+03 | $2.15 \mathrm{E}+03$ | $2.15 E+03$ | $2.15 \mathrm{E}+03$ | $2.15 \mathrm{E}+03$ | $2.16 E+03$ | $2.17 \mathrm{E}+03$ |
| RH | 0.0 | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{E}+02$ | $3.88 E+02$ | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{E}+02$ | 3.88E+02 | $3.88 \mathrm{E}+02$ | $3.88 \mathrm{E}+02$ | $3.88 E+02$ |
| PD | 0.0 | $1.41 \mathrm{E}+03$ | $1.41 \mathrm{E}+03$ | $1.41 \mathrm{E}+03$ | $1.41 E+03$ | $1.41 \mathrm{E}+03$ | 1.41E*03 | 1.41E +03 | $1.41 \mathrm{f}+03$ | $1.41 \mathrm{E}+03$ | $1.49 \mathrm{~F}+03$ | 1.4.4E+03 |
| A G | 0.0 | $3.98 E+01$ | $5.98 \mathrm{E}+01$ | $5.98 \mathrm{E}+01$ | $5.985+01$ | $5.98 \mathrm{E}+01$ | $5.98 \mathrm{E}+01$ | $5.99 E+01$ | $5.99 E+07$ | $5.99 \mathrm{E}+01$ | $6.00 E+01$ | 6.01E+01 |
| CD | 0.0 | $8.42 E+01$ | $8.41 E+01$ | 8.41E+01 | $8.41 \mathrm{E}+01$ | 8.4.4E+01 | $8.41 E+01$ | $8.41 E+01$ | $8.41 E+01$ | $8.41 E+01$ | $8.41 E+01$ | $8.41 E+01$ |
| IN | 0.0 | 1.25E+00 | 1.26E+00 | 1.26E+00 | 9.27E +00 | $1.28 \mathrm{E}+00$ | 1.28t*00 | $1.28 \mathrm{E}+00$ | $1.28 \mathrm{E}+00$ | $1.28 \mathrm{E}+00$ | $1.285+00$ | $1.28 \mathrm{E}+00$ |
| SN | 0.0 | S.11E+01 | $5.11 E+01$ | $5.11 \mathrm{E}+01$ | $5.11 E+01$ | $5.11 t+01$ | $5.11 t+01$ | $5.10 \mathrm{~F}+01$ | $5.10 E+01$ | $5.08 \mathrm{E}+07$ | $5.04 E+01$ | $4.98 \mathrm{E}+01$ |
| SE | 0.0 | $1.07 E+01$ | $1.03 \mathrm{E}+09$ | $1.01 \mathrm{E}+01$ | 1.01E+01 | $1.07 \mathrm{E}+01$ | $1.01 E+01$ | $1.01 \mathrm{E}+01$ | $1.01 \mathrm{E}+01$ | $1.01 \mathrm{t}+01$ | $1.01 \mathrm{E}+\mathrm{C} 9$ | 1.01E+01 |
| TE | 0.0 | $5.71 \mathrm{E}+02$ | $5.71 \mathrm{E}+02$ | $5.71 \mathrm{E}+02$ | $5.71 E+02$ | $5.71 \mathrm{t}+02$ | $5.71 \mathrm{E}+02$ | 5.71E+02 | $5.71 \mathrm{E}+02$ | 5.71E+02 | $5.72 \mathrm{E}+0 \mathrm{C}$ | $5.72 \mathrm{E}+02$ |
| 1 | 0.0 | $2.70 \mathrm{E}+02$ | $2.70 E+02$ | $2.70 E+02$ | $2.70 E+02$ | $2.70 E+02$ | $2.70 \mathrm{E}+02$ | $2.70 \mathrm{E}+02$ | $2.70 E+02$ | $2.70 E+02$ | $? .70 \mathrm{E}+02$ | $2.70 E+02$ |
| XE | 0.0 | 0.0 | $4.68 \mathrm{E}-05$ | $9.36 E-05$ | $2.81 \mathrm{E}-04$ | 8.43E-04 | $1.78 \mathrm{E}-03$ | $4.59 \mathrm{E}-03$ | $9.27 \mathrm{E}-03$ | $1.86 \mathrm{E}-02$ | $4.67 E-02$ | $9.35 \mathrm{E}-122$ |
| CS | 0.0 | $2.32 \mathrm{E}+03$ | $2.21 E+03$ | $2.12 E+03$ | $1.83 \mathrm{E}+03$ | $1.46 E+03$ | $1.35 E+03$ | $1.33 E * 03$ | $1.33 \mathrm{E}+03$ | $1.33 \mathrm{~F}+03$ | $1.33 E+03$ | $1.33 E+03$ |
| BA | 0.0 | $1.79 \mathrm{E}+03$ | $1.90 \mathrm{E}+03$ | $2.00 \mathrm{E}+03$ | $2.29 E+03$ | $2.66 E+03$ | $2.77 E+03$ | 2.78E +03 | $2.78 E+03$ | $2.78 \mathrm{E}+03$ | $2.78 \mathrm{E}+03$ | -.78E+03 |
| La | 0.0 | $1.27 E+03$ | 1.27E +03 | 1.27E+03 | $1.27 \mathrm{E}+03$ | $1.27 E+03$ | $1.27 \mathrm{E}+03$ | $1.27 E+03$ | $1.27 E+03$ | $1.27 \mathrm{E}+03$ | $1.27 E+03$ | $1.27 E+03$ |
| C | 0.0 | $2.48 E+03$ | $2.48 E+03$ | $2.48 \mathrm{E}+03$ | $2.48 \mathrm{E}+03$ | $2.48 \mathrm{E}+03$ | $2.48 \mathrm{E}+03$ | 2.48E+03 | $2.48 E+03$ | $2.48 \mathrm{E}+03$ | $2.48 \mathrm{E}+03$ | $2.48 E+03$ |
| PR | 3.0 | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | $1.20 E+03$ | $1.20 \mathrm{E}+03$ | $1.20 E+0$ ? | $1.70 E+03$ |
| ND | 0.0 | $4.12 t+03$ | $4.12 \mathrm{E}+03$ | $4.12 E+03$ | $4.12 \mathrm{E}+03$ | $4.12 \mathrm{E}+03$ | $4.12 E+03$ | $4.12 \mathrm{E}+03$ | $4.12 E+03$ | $4.12 \mathrm{E}+03$ | $4-12 \mathrm{E}+03$ | $4.12 E+G 3$ |
| PM | 0.0 | $8.34 \mathrm{E}+00$ | 2.22E*00 | 5.92E-01 | 2.97E-03 | $3.79 \mathrm{E}-10$ | $1.22 \mathrm{E}-21$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| SM | 0.0 | $9.00 E+02$ | $4.05 E+02$ | $9.05 \mathrm{E}+02$ | $9.00 E+02$ | 8.87E+02 | $8.76 E+02$ | 8.67E+02 | $8.66 E+02$ | $8.66 E+02$ | $8.66 E+02$ | $8.66 E+02$ |
| EU | 0.0 | $1.66 \mathrm{E}+32$ | $1.61 E+02$ | 1.58E+02 | $1.52 \mathrm{E}+02$ | $1.57 \mathrm{E}+02$ | $1.67 E+02$ | 1.76E +02 | 1.77E+02 | $1.77 \mathrm{E}+02$ | $1.77 \mathrm{E}+02$ | 1.77E+02 |
| GD | 0.0 | $1.23 E+02$ | $1.30 E+02$ | $1.35 E+02$ | $1-46 E+02$ | 1.54E+02 | $1.55 \mathrm{E}+02$ | $1.55 \mathrm{E}+02$ | $1.55 \mathrm{E}+02$ | 1.55E+02 | 1.5SE+02 | $1.55 E+02$ |
| 18 | 0.0 | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | 1.78E+00 | $1.78 \mathrm{E}+00$ | $1.78 \mathrm{E}+00$ | 1.78E*00 | $1.78 \mathrm{E}+00$ |
| DY | 0.0 | 1. $11 \mathrm{E}+00$ | $1.11 \mathrm{E}+00$ | $1.11 E+00$ | $1.11 \mathrm{E}+00$ | $1.11 \mathrm{E}+00$ | $1.11 \mathrm{E}+00$ | 1.11E*00 | $1.11 \mathrm{E}+00$ | $1.11 \mathrm{~F}+00$ | $1.11 \mathrm{E}+00$ | $1.11 \mathrm{E}+00$ |
| HO | 0.0 | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | 1.01E-01 | 1.01E-01 | $1.01 \mathrm{E}-09$ | 1.01E-01 | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ |
| ER | 5.0 | $3.08 \mathrm{E}-02$ | 3.08E-02 | 3.08E-02 | 3.08E-02 | $3.08 \mathrm{E}-02$ | 3.08E-02 | 3.09E-02 | 3.10E-02 | 3.11E-02 | 3.12E-02 | $3.12 \mathrm{E}-02$ |
| TOTALS | 0.0 | $2.91 E+04$ | 2.91E+04 | $2.91 \mathrm{E}+04$ | $2.91 E+04$ | $2.91 E+04$ | $2.91 E+04$ | $2.91 F+04$ | $2.91 E+04$ | $2.91 F+04$ | $2.91 E+04$ | $2.91 \mathrm{E}+04$ |

## A N N E X 2

ACTIVITY SPECTRA OF H L $W$ CONTATNER

The activity spectra are calculated by the origen program in following assumptions:

- The HLW quantity included in one container corresponds to one metric ton of heavy metal charged in reactor.
- The total volume of one container is 150 liters.
- The nuclear fuel charged in the reactor is a PWR type enriched $3.3 \%$.
- The irradiation has been at constant rate in 1100 days at a power $30 \mathrm{MW} / \mathrm{metric}$ ton, the total burn-up at discharge is 33000 MWL per ton.
- The irradiated elements are stored 10 years before reprocessing.
- When reprocessing is performed, we suppose $1 \%$ of fissile materials U and $P u$ stays in the wastes, but $O \%$ of volatile materials $K r$ and Xe remain.
- The spectra are given:
- in 18 energy groups for actinides
- in 12 energy groups for fission products
- The calculation is made separately for actinides with heavy materials and for fission products. Those spectra are reproduced in annexed tables.


## RESULTS

2.1 - For actinides
fuel elements storage period
2.2 - For actinides wastes after reprocessing
2.3 - For fission products - Storage period
2.4 - For fission products - After reprocessing at 10 years
$2.5-i, N$ neutron source - Storage period
$2.6-\chi . N$ neutron source - After reprocessing
2.7 - Spontaneous fission neutron - Storage period
2.8 - Spontaneous fission neutron - After reprocessing

[^1]Ex : For 10,40 and 1000 years after discharge, the total activity is :

|  | 10 years | 40 years | 1000 years |
| :---: | :---: | :---: | :---: |
| Actinides phot/s <br> Ci $\gamma$ | 2.31 E 13 <br> 624.3 | 2.23 E 13 <br> 602.7 | 6.21 E 12 <br> 167.8 |
| Fission <br> products phot/s <br> Ci $\gamma$ | 4.33 E 15 <br> $1.170 \mathrm{E} \mathrm{O5}$ | 1.71 E 15 <br> 4.62 E O | 8.32 E 10 <br> 2.25 |

reference phr equilibrium fuel cycle -- fuel decay times
POWER $=30.00 \mathrm{MW}$. BURNUP $=33000$. MWD. FLUX $=2.93 E+13 \mathrm{~N} \star 2-5 E C$
actinide photon release rates, photons/sec
GASIS = MT OF HEAVY METAL CHARGED TO REACTOR

| EAN: | time arter discharge |  |  |  |  |  |  |  | REPRO |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| (MEV) | INITIAL | 30. | 60. D | 120. D | 182. | 274 | 365. D | 30. 0 |  | 3652. |  |
| $3.00 \mathrm{E}-02$ | 5.24E+16 | $5.69 E+14$ | $2.63 E+13$ | $2.44 \mathrm{E}+11$ | $2.19 \mathrm{E}+11$ | $2.62 E+11$ | 3.05E+11 | $4.71 \mathrm{E}+11$ | 6-29E+11 | $1.54 \mathrm{E}+12$ | $2.40 \mathrm{E}+12$ |
| $4.00 E-02$ | $2.01 \mathrm{E}+17$ | $5.79 E+13$ | 7.56E+12 | $6.04 \mathrm{E}+12$ | 5.96E+12 | $5.88 \mathrm{E}+12$ | $5.83 E+12$ | $5.73 E+12$ | $5.71 E+12$ | 5.69E+12 | 5.67 |
| $6.00 E-02$ | $2.56 E+17$ | $5.85 E+14$ | $2.74 \mathrm{E}+13$ | $2.64 E+12$ | $2.95 t+12$ | $3.50 E+12$ | $4.03 t+12$ | $6.10 E+12$ | $8.07 \mathrm{E}+12$ | $1.94 t+13$ | $3.01 \mathrm{E}+13$ |
| $1.00 \mathrm{E}-01$ | $3.91 \mathrm{E}+17$ | $1.22 E+12$ | $1.18 \mathrm{E}+11$ | $6.26 E+10$ | $5.59 E+10$ | $4.90 E+10$ | $4.46 E+10$ | $3.85 \mathrm{E}+10$ | $3.88 E+10$ | $5.03 \mathrm{E}+9 \mathrm{C}$ | 6.13 |
| $1.50 \mathrm{E}-01$ | $1.80 E+17$ | $5.85 E+13$ | $1.83 E+12$ | 3-41E+11 | $3.35 E+11$ | $3.31 \mathrm{E}+11$ | $3.28 t+11$ | $3.23 E+11$ | $3.22 \mathrm{E}+11$ | 3-18E+11 | $3.16 \mathrm{E}+11$ |
| $2.00 \mathrm{E}-01$ | $1.42 E+17$ | 3.72E+14 | 1-64E+13 | $2.14 \mathrm{E}+11$ | $1.79 \mathrm{E}+11$ | $1.78 \mathrm{E}+11$ | $1.78 \mathrm{E}+11$ | 1.77E+11 | $1.75 E+11$ | $1.67 E+11$ | $1.58 \mathrm{E}+11$ |
| $3.00 \mathrm{e}-01$ | $9.57 \mathrm{E}+10$ | $4.98 E+13$ | $1.78 \mathrm{E}+12$ | $1.24 E+11$ | $1.20 \mathrm{E}+11$ | $1.20 E+11$ | $1.20 E+11$ | $1.19 \mathrm{E}+11$ | $1.19 E+11$ | $1.16 \mathrm{E}+11$ | 1.13 E |
| $6.30 E-01$ | $1.60 t+16$ | $2.54 \mathrm{E}+12$ | $7.19 \mathrm{E}+11$ | $6.79 \mathrm{E}+11$ | $6.48 \mathrm{E}+11$ | $6.15 E+11$ | $5.92 E+11$ | $5.55 \mathrm{E}+11$ | $5.47 E+11$ | . $45 \mathrm{E}+11$ | $5.44 \mathrm{E}+11$ |
| $1.10 \mathrm{E}+00$ | $8.61 E+15$ | 5.23E+11 | $1.39 E+11$ | $1.39 \mathrm{E}+11$ | $1.39 E+11$ | $1.39 E+1$ | 1. $39 \mathrm{E}+1$ | $1.39 E+11$ | $1.39 E+11$ | $1.39 E+11$ | $1.39 \mathrm{~F}+11$ |
| $1.55 \mathrm{E}+20$ | $2.77 \mathrm{E}+08$ | $2.46 E+08$ | $2.35 E+08$ | $2.17 E+08$ | $2.03 E+08$ | $1.87 \mathrm{E}+08$ | $1.76 \mathrm{E}+08$ | $1.55 \mathrm{E}+08$ | $1.47 \mathrm{E}+08$ | 1.18E+08 | $8.45 E+07$ |
| $1.99 \mathrm{E}+00$ | $1.37 \mathrm{E}+08$ | $1.29 E+08$ | 1.24E+08 | $1.14 \mathrm{E}+08$ | $1.06 t+08$ | $9.81 \mathrm{E}+07$ | $9.23 E+07$ | $8.11 E+07$ | $7.69 \mathrm{E}+07$ | -12E+07 | 4.37 |
| $2.38 \mathrm{E}+00$ | $6.74 \mathrm{E}+07$ | $6.43 E+07$ | $6.15 \mathrm{E}+07$ | $5.67 \mathrm{E}+07$ | 5.28E+07 | $4.86 \mathrm{E}+07$ | $4.56 \mathrm{E}+07$ | $3.99 E+07$ | $3.75 E+07$ | 2.87E +07 | $1.98 \mathrm{~F}+07$ |
| $2.75 \mathrm{E}+00$ | $5.04 E+07$ | $5.04 \mathrm{E}+07$ | S.09E+07 | $5.28 \mathrm{E}+07$ | $5.51 \mathrm{E}+07$ | $5.94 E+07$ | $6.44 E+07$ | $8.82 E+07$ | $1.13 E+48$ | 2-16E+08 | 1.98F4 |
| $3.25 E+00$ | $1.95 E+07$ | 1.86E +07 | $1.78 \mathrm{E}+07$ | $1.64 F+07$ | $1.53 \mathrm{E}+07$ | $1.41 E+07$ | $1.32 \mathrm{E}+07$ | $1.15 \mathrm{t}+07$ | $1.08 \mathrm{E}+07$ | $8.28 F+06$ | 7 |
| $3.70 \mathrm{E}+10$ | $1.25 t+07$ | 1.19E+07 | $1.14 \mathrm{t}+07$ | $1.05 \mathrm{E}+07$ | $9.81 \mathrm{E}+06$ | $9.62 E+06$ | 8.47E + 0 ó | $7.40 t+0 t$ | 0.96E+06 | . $32 \mathrm{E}+06$ | . $66 \mathrm{E}+00$ |
| $4.22 E+00$ | ?.89E+06 | 7.54E+06 | 7-20E +06 | $6.64 \mathrm{E}+06$ | -. 19E +06 | $5.64 \varepsilon+06$ | $5.35 E+06$ | $4.67 E+06$ | $4.39 E+06$ | $3.35 E+70$ | 2.31E*06 |
| $4.70 E+0 \cup$ | $3.74 \mathrm{E}+06$ | $3.57 \mathrm{E}+\mathrm{C} 6$ | $3.41 E+06$ | $3.14 F+06$ | $2.93 E+06$ | $2.69 E+06$ | 2.53E+06 | 2. $21 \mathrm{E}+06$ | $2.08 \mathrm{E}+06$ | 1.59E+00 | $1.09 E+0$ |
| $5.25 E+00$ | $2.35 E+00$ | $2.24 E+06$ | $2.14 E+06$ | $1.98 \mathrm{E}+06$ | $1.84 E+06$ | $1.69 t+76$ | 1.59E+06 | 1.39E+06 | $1.31 \mathrm{E}+06$ | $9.99 \mathrm{E}+0.5$ | $6.88 \mathrm{E}+0$ |
| total | $1.34 t+18$ | 1.70E+15 | $8.22 \mathrm{~F}+1$ | 1.65t+13 | 1.06E+13 | $1.11 \mathrm{E}+13$ | 1.16E+13 | $1.37 E+13$ | $1.58 \mathrm{E}+13$ | $2.80 E+13$ | $3.95 \mathrm{f}+1$ |
| MEV/SEC | $1.68 \mathrm{E}+17$ | 1.55E+14 | $7.44 t+12$ | $1.13 \mathrm{E}+12$ | 1.11E+12 | $1.12 \mathrm{E}+12$ | 1.14E+12 | $1.24 E+12$ | 1.35E+1 | . $06 \mathrm{E}+1$ |  |

ACTINIDE ENERGY RELEASE RATES, MEV/WATT-SEC


# PEFERFNCF PWR EQUILIBRIUM FUEL CYCLE -- HASTE DECAY TIMES <br> POWER $=30.00 \mathrm{MW}$, BURNUP $=33000$. MWD. FLUX $=2.93 E+13 \mathrm{~N} * 2-$ SEC 

aCtINIDE PHOTON RELEASE RATES, PHOTONS/SEC GASIS = MT OF HEAVY METAL CHARGED TO REACTIOR

| EMEAN | REPRO |
| :---: | :---: |
| (MEV) | $+\ldots++1$ |
| $3.00 E-02$ | $1.52 E+12$ |
| $4.00 E-02$ | $8.56 E+19$ |
| $6.00 E-02$ | $1.94 E+13$ |
| $1.00 E-01$ | $2.55 E+10$ |
| $1.50 E-01$ | $3.11 E+11$ |
| $2.00 E-01$ | $1.53 E+11$ |
| $3.00 E-01$ | $1.15 E+11$ |
| $6.30 E-01$ | $5.45 E+11$ |
| $9.10 E+00$ | $1.39 E+11$ |
| $1.55 E+00$ | $1.16 E+08$ |
| $1.99 E+00$ | $0.03 E+07$ |
| $2.38 E+00$ | $2.82 E+07$ |
| $2.75 E+00$ | $2.16 E+08$ |
| $3.25 E+00$ | $8.15 E+06$ |
| $3.70 E+00$ | $5.24 E+06$ |
| $4.22 E+00$ | $3.30 E+06$ |
| $4.70 E+00$ | $1.56 E+06$ |
| $5.25 E+00$ | $9.83 E+05$ |
| TOYAL | $2.31 E+13$ |
| $M E V / S E C$ | $1.85 E+12$ |

ACTINIDE ENERGY RELEASE RATES, MEV/WATT-SEC
BASIS = MT OF HEAVY METAL CHARGED TO REACTOR



Photon spectrum as a function of time for fission products

REFERENCE PWR EQUILIGRIUM FUEL CYCLE -- FUEL DECAY Times
POWER $=30.00 \mathrm{MW}$. BURNUP $=33000$. MWD. FLUX $=2.93 \mathrm{E}+13 \mathrm{~N} * 2-$ SEC
TWELVE GROUP FHOTON RELEASE RATES. PHOTONS/SEC
BASIS = MT OF HEAVY METAL CHARGED TO REACTOR


TWELVE GROUP ENERGY RELEASE RATES, MEV/WATT-SEC
EASIS = MT Of heavy Metal Charged to heactor

|  | EMEAN (MEV) | INITIAL | 30. 0 | $60 . \mathrm{D}^{\text {TIME }}$ | $\begin{gathered} \text { AFTER DI } \\ 120.0 \end{gathered}$ | $\begin{aligned} & \text { CHARGE } \\ & 182 . \text { D } \end{aligned}$ | 274. D | 365. D | 730. | 1096. D | $\frac{\text { REPRO }}{3652.0}$ | 3305.0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 3.00t-01 | $9.13 E+04$ | $2.19 \mathrm{E}+08$ | $1.31 E+08$ | $9.21 E+07$ | 7.49E+07 | $5.91 E+77$ | $4.81 E+07$ | 2.24E+07 | 1.12E+07 | $1.76 \mathrm{E}+10$ | 1.28E+06 |
|  | 6.30E-01 | $3.61 E+10$ | 3.66E +09 | 2.72E+09 | 9.74E+09 | 1.18E+09 | $7.72 E+08$ | $5.89 E+08$ | $3.64 E+C 8$ | $2.70 \mathrm{E}+08$ | 8.42E+07 | $5.43 E+07$ |
|  | $1.10 \mathrm{E}+00$ | $2.35 E+10$ | $1.82 \mathrm{E}+08$ | $9.54 F+07$ | $6.29 E+07$ | $5.39 \mathrm{E}+07$ | $4.56 E+07$ | $3.95 E+07$ | $2.38 E+07$ | $1.58 \mathrm{E}+\mathrm{C} 7$ | $4.99 \mathrm{E}+06$ | $3.02 \mathrm{~F}+06$ |
|  | $1.55 E+00$ | 2.01E+10 | $6.59 \mathrm{E}+08$ | $1.55 E+08$ | $3.30 E+07$ | $2.54 \mathrm{t}+07$ | $2.17 E+07$ | $1.88 \mathrm{E}+07$ | $1.11 E+07$ | $6.92 \mathrm{E}+06$ | $5.45 \mathrm{E}+05$ | $5.85 t+0.4$ |
|  | $1.99 E+00$ | $4.34 E+09$ | $5.27 E+07$ | $2.45 E+07$ | $1.98 \mathrm{E}+07$ | 1.67E+07 | $1.34 E+07$ | $1.08 \mathrm{E}+07$ | $4.59 \mathrm{E}+06$ | $1.95 \mathrm{E}+66$ | $8.19 \mathrm{E}+03$ | 1. $58 \mathrm{Et}+03$ |
|  | 2.38E+00 | $5.66 E+09$ | $3.36 E+07$ | $8.64 \mathrm{E}+06$ | $2.50 E+06$ | $2.02 \mathrm{E}+06$ | $1.69 E+06$ | $1.42 \mathrm{E}+06$ | $7.07 E+05$ | $3.52 E+05$ | $2.77 \mathrm{E}+03$ | 2.78E+00 |
|  | $2.75 E+00$ | $2.73 E+49$ | $2.36 E+05$ | $2.23 E+05$ | $1.99 E+05$ | $1.77 \mathrm{E}+05$ | $1.49 \mathrm{E}+05$ | $1.25 E+05$ | $6.27 E+04$ | $3.14 E+04$ | $2.51 \mathrm{E}+02$ | 2-33E-01 |
|  | $3.25 t+00$ | $5.63 \mathrm{E}+49$ | $8.78 \mathrm{E}+03$ | $8.30 k+03$ | $7.41 \mathrm{E}+03$ | $6.59 \mathrm{E}+03$ | $5.54 \mathrm{E}+03$ | $4.06 E+03$ | $2.34 E+03$ | $1.17 E+03$ | $9.39 E+C 0$ | 9.47E-03 |
|  | $3.70 \mathrm{E}+00$ | $3.78 \mathrm{E}+08$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
|  | $4.22 \mathrm{E}+00$ | 1.32E+09 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
|  | $4.70 \mathrm{E}+00$ | $6.76 E+C 8$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | C. 0 |
|  | $5.25 E+00$ | $1.90 \mathrm{E}+08$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | C.0 |
|  | total | 1.10E+11 | $4.81 E+09$ | $3.14 E+09$ | 1.95E+09 | $1.36 E+04$ | $9.14 \mathrm{E}+08$ | $7.08 \mathrm{E}+08$ | $4.27 E+08$ | 3.07E+08 | $9.15 E+07$ | S.80E + U7 |
| GAMMA | WATIS | $5.28 E+05$ | $2.315+04$ | $1.51 \mathrm{E}+04$ | $9.36 E+03$ | $6.52 E+03$ | $4.39 E+03$ | 3.41E+03 | $2.05 E+03$ | $1.47 \mathrm{E}+03$ | $4.40 \mathrm{E}+02$ | $2.82 \mathrm{E}+\mathrm{6} 2$ |

Photon spectrum as a function uf time for fission products

```
REfERENCE PWR EQUILIBRIUM fuel cycle -- waste oecay times
    POWER= 30.00 MW, BURNUP= 33000.MWO, FLUX= 2.93E*13 N**2-SEC
```

TWelve group photon release rates. photons/se
BASIS = T OF HEAVY METAL CHARGED TO REACTOR

| EMFAN | REPRO |  | II | AFter D | CHARGE |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| (MEV) | + | 5. Y | 10. Y | $30 . Y$ | 90. Y | 190. Y | 490. Y | 990. Y | 1990. Y | 4990. Y | 9990. Y |
| 3.00E-01 | $1.74 E+14$ | $1.46 E+14$ | 1.27E+14 | 7.06E+13 | $1.74 \mathrm{E}+13$ | $1.53 \mathrm{E}+12$ | $8.48 \mathrm{E}+09$ | $6.37 \mathrm{E}+09$ | $6.30 E+09$ | $6.17 \mathrm{E}+09$ | $5.97 E+09$ |
| $6.30 \mathrm{E}-01$ | $4.01 E+15$ | 3. $022 E+15$ | $2.58 E+15$ | $1.60 E+15$ | $3.95 E+14$ | $3.90 E+13$ | $1.13 E+11$ | $7.52 E+10$ | $7.47 \mathrm{E}+10$ | $7.31 \mathrm{E}+10$ | 7.06E+10 |
| 1.10E+00 | $1.36 \mathrm{E}+14$ | $1.03 \mathrm{E}+14$ | $8.23 E+13$ | $3.53 E+13$ | $3.03 E+12$ | $8.31 E+10$ | $3.26 E+07$ | $9.54 E+05$ | $5.35 E+05$ | $9.46 E+04$ | $5.26 E+03$ |
| $1.55 \mathrm{E}+00$ | $1.05 E+13$ | $2.65 E+12$ | 1.13E+1? | $4.90 E+11$ | 1.19E+11 | $1.09 \mathrm{E}+10$ | $1.66 \mathrm{E}+09$ | $1.65 \mathrm{E}+09$ | $1.64 \mathrm{E}+09$ | $1.61 F+09$ | $1.55 E+09$ |
| $1.99 E+00$ | $1.23 E+11$ | $2.90 E+10$ | $2.39 E+10$ | 1.45E+10 | $3.31 \mathrm{E}+09$ | $2.81 E+08$ | $1.72 \mathrm{E}+05$ | 7.54E-01 | $1.46 \mathrm{E}-11$ | 0.0 | 0.0 |
| $2.38 \mathrm{E}+00$ | $3.49 E+10$ | $1.10 E+09$ | $3=50 E+07$ | $3.50 E+01$ | 3-75E-17 | 0.0 | 0.6 | 0.0 | 0.0 | 0.0 | C. 0 |
| $2.75 \mathrm{E}+00$ | $2.74 \mathrm{E}+09$ | $8.68 E+07$ | $2.76 E+06$ | $2.81 E+00$ | 2.96E-18 | 0.0 | 0.0 | 0.0 | 0.0 | $0 . \mathrm{C}$ | 0.0 |
| $3.25 \mathrm{E}+00$ | 8.67E+07 | $2.75 E+06$ | $8.75 E+04$ | $8.90 E-02$ | $0.38 \mathrm{E}-20$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $3.70 E+00$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $4.22 E+00$ | 0.0 | 0.0 | 0.0 | 6.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $4.70 \mathrm{~F}+00$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $5.25 E+00$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| TOTAL | $4.33 t+15$ | $3.27 E+15$ | $2.79 E+15$ | 1.71E+15 | $4.15 t+14$ | $4.07 E+13$ | $1.24 \mathrm{E}+11$ | $8.32 E+10$ | $8.26 E+10$ | $8.69 \mathrm{E}+10$ | $7.82 F+10$ |

twelve group energy release rates, meviwatt-sec
BASIS = MT OF HEAVY METAL CHARGED TO REACTOR

| EMEAN | REPRO |  | T1M | AFter | harge |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| (MEV) |  | 5. Y | 10. Y | 30. Y | $90 . Y$ | 190. Y | 490. Y | 990. ${ }^{\text {Y }}$ | 1990. Y | 4990. Y | 9990. Y |
| $3.00 \mathrm{E}-01$ | $1.74 \mathrm{E}+0 \mathrm{C}$ | $1.46 \mathrm{E}+06$ | $1.27 E+06$ | 7.66E+05 | $1.74 \mathrm{E}+05$ | $1.53 E+04$ | $8.48 \mathrm{E}+01$ | $6.37 E+01$ | $6.30 \mathrm{E}+01$ | 6.17E+ 71 | $5.97 \mathrm{E}+01$ |
| $6.30 \mathrm{E}-01$ | $8.41 E+07$ | -. $34 \mathrm{E}+07$ | $5.43 E+07$ | $3.36 E+07$ | $8.29 E+06$ | $8-20 E+05$ | $2.38 E+03$ | $1.58 \mathrm{E}+03$ | $1.57 \mathrm{E}+03$ | $1.54 E+03$ | $1.48 \mathrm{E}+03$ |
| $1.10 \mathrm{E}+00$ | $4.99 E+00$ | $3.78 \mathrm{E}+76$ | 3.02E +06 | $1.29 E+06$ | $1.11 \mathrm{E}+05$ | $3.05 E+03$ | $1.20 \mathrm{E}+0 \mathrm{C}$ | $3.50 \mathrm{E}-02$ | $1.96 \mathrm{E}-02$ | 3.47E-03 | 1.93E-04 |
| 1.55E+00 | $5.45 \mathrm{E}+05$ | $1.37 E+05$ | $5.85 E+04$ | $2.53 \mathrm{E}+04$ | $5.74 \mathrm{E}+03$ | $5.64 E+02$ | $8.59 E+01$ | $8.53 E+01$ | $8.47 \mathrm{E}+01$ | $8.30 E+09$ | $8.02 \mathrm{E}+01$ |
| $1.99 \mathrm{E}+00$ | $8.19 \mathrm{E}+03$ | $1.92 \mathrm{E}+03$ | $1.58 \mathrm{E}+03$ | $9.65 E+02$ | 2. $20 \mathrm{E}+02$ | $1.86 E+01$ | $1.14 \mathrm{E}-02$ | $5.00 \mathrm{e}-08$ | $9.67 \mathrm{E}-19$ | 0.0 | 0.0 |
| $2.38 \mathrm{E}+00$ | $2.77 \mathrm{t}+03$ | $8.76 E+01$ | $2.78 \mathrm{E}+00$ | 2.82E-06 | $2.98 \mathrm{E}-24$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $2.75 \mathrm{E}+00$ | $2.51 \mathrm{E}+07$ | 7.96E+00 | $2.53 \mathrm{E}-01$ | 2.57E-07 | $2.71 \mathrm{E}-25$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $3.25 E+00$ | $9.39 E+00$ | 2.98E-01 | $9.47 \mathrm{E}-03$ | 9-64E-09 | $1.02 \mathrm{E}-26$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| 3.70E +00 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $4.22 \mathrm{E}+00$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $4.70 E+00$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| $5.25 E+00$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | C. 0 |
| total | $9.14 \mathrm{~F}+07$ | $6.88 E+07$ | $5.86 E+07$ | 3.56E*07 | 8.58E+06 | $8.39 E+05$ | $2.55 E+03$ | 1.73E+03 | $1.72 E+03$ | $1.68 \mathrm{E}+03$ | 1.62E+03 |
| WATtS | $4.40 E+02$ | $3.31 E+02$ | $2.82 E+02$ | $1.71 \mathrm{t}+02$ | $4=13 E+01$ | $4.03 E+00$ | $1.23 E-02$ | 8.31E-03 | $8.25 \mathrm{E}-03$ | $8.08 \mathrm{E}-0.3$ | $7.81 \mathrm{E}-03$ |

basis = mt of heavy metal charged to reactor

|  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | INITIAL | 30. 0 |  |  |  |  |  |  |  |  |  |
| 81211 | 3.44E-03 | 3.67E-03 | 3.91E-03 | 4.39E-03 | $4.88 \mathrm{E}-03$ | $5.62 \mathrm{E}-03$ | $6.34 E-03$ | $9.54 \mathrm{E}-03$ | 2 | 2 |  |
| B1212 | 3.76E-02 | $4.03 \mathrm{E}-02$ | $4.39 \mathrm{E}-02$ | 5.22E-02 |  |  | $8.52 \mathrm{E}-02$ | .38E-U1 | $1.89 \mathrm{E}-01$ | 4-01E-01 |  |
| B1213 | $9.61 \mathrm{E}-12$ | 5.56E-12 | .87E | 1.78 E |  |  |  | $2.09 \mathrm{E}-12$ | 2.32E-12 | 4.15 |  |
| P0210 | 6.74E-07 | 7.31E-07 | 7.91E-07 | $9.17 \mathrm{E}-07$ | 1.06E-06 | $1.30 E-06$ | 1-56E-06 | 2.93E-06 | $4.93 \mathrm{E}-06$ | 5.43 |  |
| P0211 | $1.61 \mathrm{E}-05$ | 1.72E-35 | $1.83 \mathrm{E}-05$ | $2.05 \mathrm{E}-05$ | 2.28E-05 | $2.62 \mathrm{E}-05$ | $2.96 E-05$ | $4.40 \mathrm{E}-05$ | $5.78 \mathrm{E}-05$ | 1.4 |  |
| P0212 | $1.06 E+01$ | $1.14 \mathrm{E}+01$ | $1.24 \mathrm{E}+01$ | $1.47 \mathrm{E}+01$ | $1.70 \mathrm{E}+01$ | $2.05 E+01$ | $2.40 E+01$ | $3.88 \mathrm{E}+0$ | $5.35 \mathrm{E}+01$ | 1.13 | 21E |
| P0213 | 1.76F-03 | 1.02E-03 | 5.25E-04 | 3.26E-04 | 3.18E-04 | $3.28 \mathrm{E}-04$ | 3.38E-04 | $3.83 \mathrm{E}-04$ |  | ? | 1.34E-0. |
| P0214 | 1-10t-04 | $1.14 \mathrm{E}-04$ | $1.18 \mathrm{E}-04$ | 1.29t-04 | $1.39 \mathrm{E}-04$ | $1.55 \mathrm{E}-04$ | 1 | $2.53 \mathrm{E}-04$ | $3.53 \mathrm{E}-04$ | $1.62 \mathrm{E}-03$ | 5.2 |
| P0215 | 3.18E-03 | 5.53E-03 | 5.89E-03 | $0.61 \mathrm{~F}-03$ | 7.36E-03 | $8.46 \mathrm{E}-03$ |  |  | 1.86E-02 | 4.66 E | 8.2 |
| P0216 | 6.43 E + | $6.89 E+00$ | . 50 E | 8.91 E | $1.03 \mathrm{E}+01$ | $1.24 \mathrm{E}+01$ | $1.46 \mathrm{E}+01$ | 2.35E+01 | $3.24 \mathrm{~F}+01$ | 6.85 E | $7.33 \mathrm{E}+0$ |
| P02 | . 468 | $4.60 E-05$ | $4.78 \mathrm{E}-05$ | S.19E-05 | $5.60 \mathrm{E}=05$ | $6.25 E-05$ | 6.94E-05 | $1.02 \mathrm{E}-04$ | $1.43 \mathrm{E}-04$ | . 52 | 2.11E-03 |
| A1217 | $9.64 \mathrm{E}-04$ | 5.58E-04 | $2.88 \mathrm{E}-0.14$ | 1.79E-04 | 1.75E-04 | $1.80 \mathrm{E}-04$ | 1.85E-04 | 2.10E-04 | 2.33E-04 | 16 | $7.36 \mathrm{E}-04$ |
| RN219 | 3.87E-03 | $4.13 E-03$ | 4.4 UE-03 | $4.93 \mathrm{E}-03$ | $5.49 \mathrm{E}-03$ | $0.31 \mathrm{E}-03$ | 7.12E-03 | 1.07 E | 1.3 | 3.47E-02 | - |
| RN220 | $4.88 \mathrm{E}+00$ | $5.24 \mathrm{E}+00$ | $5.70 \mathrm{E}+00$ | $6.77 \mathrm{E}+00$ | $7.81 \mathrm{E}+00$ | $9.41 \mathrm{E}+00$ | $1.11 \mathrm{E}+01$ | 79E+01 | $2.48 E+01$ | $5.20 \mathrm{E}+01$ | 5.57E+C1 |
| RN222 | 3.21E-05 | $3.32 \mathrm{E}-05$ | $3.45 \mathrm{E}-0.5$ | 3.74E-05 | $4.04 E-05$ | $4.51 \mathrm{E}-05$ | 5.00e-05 | 36 | 1.03 | . | $1.52 \mathrm{E}-03$ |
| F K221 | 6.48E-04 | 3.75E-04 | $1.93 E-04$ | 1.20E-04 | $1.17 \mathrm{E}-04$ | 1.21E-04 | 1.24 | 1.41 E | $1.57 \mathrm{E}-04$ | 2.80E-04 | 4.9 |
| RA223 | 2-24E-03 | $2.34 \mathrm{E}-03$ | $2.54 E-03$ | $2.85 E-03$ | $3.18 \mathrm{E}-0$ | 3.65 E-03 | 4.12E-03 | 6. $21 \mathrm{E}-03$ | 8.05 ¢-03 | $2.01 \mathrm{f}-02$ | 3.54 |
| RA224 | $3.37 E+00$ | $3.62 \mathrm{E}+00$ | .93E+00 | 4.68 E | $5.39 \mathrm{E}+00$ | $6.50 \mathrm{E}+00$ | 7.64E+00 | $1.23 E+01$ | 1.70E+01 | $3.59 \mathrm{E}+01$ | . 85 |
| RA2 | $1.95 \mathrm{E}-0.5$ | 2.02E-05 | 2.10E-05 | 2.26E-05 | 2.44E-0.5 | 2.73E-05 | $3.03 \mathrm{E}-05$ | $4.45 \mathrm{SE-OS}$ | $6.22 \mathrm{E}-05$ | .85E-04 | $9.19 \mathrm{E}-04$ |
| A C | 4.75E-04 | 2.75E-04 | $1.42 \mathrm{E}-04$ | $8.81 \mathrm{E}-05$ | $8.60 \mathrm{E}-05$ | 8.85E-05 | $9.12 \mathrm{E}-05$ | $1.03 \mathrm{E}-04$ | 1-15E-04 | 2.05E-04 | $3.62 \mathrm{E}-04$ |
| A C | $3.16 \mathrm{E}-12$ | 3.36E-12 | $3.56 \mathrm{E}-12$ | 3.90E-12 | $4.38 \mathrm{E}-12$ | $4.99 \mathrm{E}-12$ | 5.59E-12 | 99E-12 | . 03 | 2.59F-11 | 4.50E-11 |
| TH2 | $2.53 \mathrm{E}-03$ | $2.70 E-03$ | 2.87E-03 | 3.21E-03 | 3.56E-03 | $4.08 \mathrm{E}-03$ | 4. | . 77 | 8.78 | 2.1 | 3.8 |
| TH228 | $2.83 \mathrm{E}+00$ | $3.09 E+00$ | $3.36 E+00$ | $3.92 \mathrm{~F}+00$ | $4.52 \mathrm{E}+00$ | 5.4 | 6.4 | $1.04 \mathrm{E}+01$ | 1.43 | 3.02 E | 3.2 |
| TH229 | $4.93 \mathrm{E}-0.5$ | $4.98 \mathrm{E}-05$ | 5.03E-05 | 5.14E-05 | $5.25 \mathrm{E}-05$ | 5.41 | 5.58 | 6.25 | 6.95 | 1.241-04 | 2.19 |
| TH230 | 1.96E-02 | 2.02E-02 | 2.08E-02 | 2.20F-02 | 2.32E-02 | 2.50 E | $2.68 \mathrm{E}-0$ | $3.41 \mathrm{E}-02$ | $4.15 \mathrm{E}-02$ | $9.52 \mathrm{E}-02$ | 1.78 |
| TH232 | $1.60 \mathrm{E}=08$ | 1. | 1.7 | 1.89 | $2.04 \mathrm{E}-08$ | 2.26E-08 | 2.48E-08 | $3.37 \mathrm{E}-08$ | $4.26 E-08$ | $1.05 \mathrm{E}-07$ | . 9 |
| PA231 | $3.58 \mathrm{E}-02$ | 3.5 | 3.59E-02 | 3.60E-02 | 3-61E-02 | 3.63E-02 | 3.64E-02 | $3.69 \mathrm{E}-02$ | 3.75E-02 | $4-12 \mathrm{E}-0$ | $4.65 \mathrm{~F}-02$ |
| U2 | 1.07 E | $1.12 E+01$ | $1.17 \mathrm{E}+01$ | $1.26 E+01$ | $1.35 E+01$ | $1.47 \mathrm{E}+01$ | 1.59L+01 | $1.99 E+01$ | $2.29 \mathrm{E}+01$ | . 07 | $2.97 \mathrm{E}+0$ |
| 02 | 5.59E-02 | $5.61 \mathrm{E}=02$ | $5.62 \mathrm{E}-02$ | 5.65E-02 | 5.68E-02 | $5.73 \mathrm{t}-02$ | 5.77E-02 | 97f-0 | 17 | $7.43 \mathrm{E}-02$ | 9.24E-02 |
| U234 | $8.87 \mathrm{E}+02$ | $8.88 \mathrm{E}+02$ | $8.89 E+02$ | $8.90 \mathrm{E}+02$ | $8.92 \mathrm{E}+02$ | $8.94 \mathrm{E}+02$ | $8.97 \mathrm{E}+02$ | $9.16 \mathrm{E}+02$ | $9.16 \mathrm{E}+02$ | $9.80 \mathrm{E}+\mathrm{U}$ | $1.07 \mathrm{~F}+0$ |
| U235 | 1-76E+01 | $1.76 \mathrm{E}+01$ | $1.76 \mathrm{E}+01$ | 1.76E +01 | $1.76 E+01$ | 9.76E+01 | 1.7 | . 76 | $1.76 \mathrm{E}+01$ | $1.76 t+01$ | $1.76 E+0$ |
| U236 | $2.73 E+02$ | $2.73 E+02$ | $2.73 \mathrm{E}+0$ | 2.73E*02 | 2. | 2.73 | 2.73 | $2.73 \mathrm{E}+$ | $2.73 t+02$ | $2.73 \mathrm{~F}+42$ | 2. |
| U2 | 2.32E+02 | 2.32E+02 | $2.32 \mathrm{E}+02$ | 2.32E+02 | 2.32 | 2.32 E | 2.32 E | $2.32 \mathrm{E}+02$ | $2.32 E+02$ | $2.32 F+02$ | 2. |
| NP2 | $4-24 \mathrm{E}$ | 4.3 | 4.33 | $4.33 \mathrm{E}+0$ | $4.33 E+02$ | $4.33 E+02$ | $4.33 E+02$ | $4.34 \mathrm{E}+02$ | $4.34 \mathrm{E}+02$ | $4.37 t+02$ | $4.44 E+0$ |
| Pu | 8.2 | 8.14 E | $7.98 \mathrm{E}+02$ | 7.67E+02 | $7.36 E+02$ | $0.92 \mathrm{E}+02$ | $6.52 \mathrm{E}+02$ | $5.11 \mathrm{E}+02$ | $4.00 E+02$ | $7.30 \mathrm{t}+01$ | 6-41E+00 |
| PU2 | $5.41 \mathrm{E}+06$ | $5.49 \mathrm{E}+06$ | $5.52 t+06$ | $5.58 \mathrm{E}+06$ | $5.62 \mathrm{E}+00$ | $5.65 E+00$ | $5.68 \mathrm{E}+06$ | . 69 E + +0 | 66t | . 36 E | -4 |
| PU239 | $4-98 \mathrm{t}+05$ | S.U5E +05 | S.05E+05 | 5.05E+05 | $5.05 \mathrm{E}+05$ | $5.05 \mathrm{E}+05$ | $5.05 \mathrm{E}+05$ | . 05 | $5.05 \mathrm{E}+05$ | $5.05 \mathrm{E}+0 \mathrm{~S}$ | 5.05 E |
| PU240 | 7-54E+03 | $7.54 \mathrm{E}+35$ | $7.54 \mathrm{E}+05$ | 7.54E+05 | $7.54 \mathrm{E}+05$ | 7-54E+05 | 7.54 E | 54 | $7.55 \mathrm{E}+05$ | 7-56E+05 | $7.58 \mathrm{E}+0$ |
| PU241 | $4.56 \mathrm{E}-14$ | $4.55 \mathrm{E}-14$ | $4.53 \mathrm{E}-14$ | 4.49E-14 | $4.46 \mathrm{E}-14$ | 40t-14 | 4.35 | 15 | $3.96 E-14$ | $2.84 \mathrm{~F}-14$ |  |
| PU242 | $1.80 E+03$ | 1-80E+03 | $1.80 t+03$ | $80 \mathrm{E}+03$ | 1.80 | 80E | 1.80 | 1.8UE+03 | $1.80 \mathrm{E}+03$ | $1.80 \mathrm{E}+03$ |  |
| PU244 | 1.72E-12 | $2.24 E-12$ | 2.75 E | 88 | $4.84 \mathrm{E}-12$ | $6.42 \mathrm{E}-12$ | 7.98E-12 | $1.42 \mathrm{E}-11$ | $2.05 \mathrm{E}-11$ | -. $.43 \mathrm{E}-11$ | 1.2 |
| AM241 | 1.76 E | 2.04 E | $2.32 E+05$ | 2.88E+05 | $3.45 E+05$ | $4.29 \mathrm{E}+05$ | $5.11 \mathrm{E}+05$ | $8.3 \cup E+05$ | $1.13 \mathrm{E}+06$ | $2.89 E+06$ | . 53 |
| AM243 | 3.2 | $3.27 E+04$ | $3.27 E+04$ | 3.27E 04 | $3.27 E+04$ | $3.27 \mathrm{E}+04$ | $3.27 \mathrm{E}+04$ | $3.27 E+04$ | $3.27 E+04$ | $3.27 \mathrm{E}+04$ | 3.27 |
| CM2 | $9.77 \mathrm{Et07}$ | $8.65 E+07$ | 7.62E+07 | 5.90E+07 | $4.53 \mathrm{E}+07$ | $3.07 \mathrm{E}+07$ | $2.08 \mathrm{E}+07$ | $4.43 \mathrm{E}+06$ | $9.51 \mathrm{t}+05$ | $2.10 \mathrm{E}+04$ |  |
| CM243 | 1.06E+04 | $1.05 E+04$ | $1.05 \mathrm{E}+04$ | 1.05E+04 | $1.05 \mathrm{E}+0.04$ | $1.04 \mathrm{E}+0.4$ | $1.03 \mathrm{E}+04$ | .01E+04 | $9.90 \mathrm{E}+03$ | $8.515+03$ |  |
| CM244 | $5.94 E+06$ | $5.93 E+06$ | $5.91 E+06$ | $5.87 E+06$ | $5.83 \mathrm{E}+06$ | $5.78 \mathrm{E}+06$ | $5.72 \mathrm{E}+06$ | . 1 E +06 | - $\mathrm{E}+06$ | $4.05 \mathrm{~F}+06$ |  |
| CM245 | $5.98 \mathrm{E}+02$ | $6.98 \mathrm{E}+02$ | $6.98 \mathrm{E}+02$ | $6.98 \mathrm{E}+02$ | $6.98 E+02$ | $6.98 E+02$ | .98E +02 | 98E | 6.97 E | $97 \mathrm{E}+02$ |  |
| CM246 | $1.27 E+02$ | $1.27 E+02$ | $1.27 E+02$ | 1.27E+02 | $1.27 E+02$ | $1.27 \mathrm{E}+0$ | $1.27 \mathrm{E}+02$ | $1.27 \mathrm{E}+02$ | $1.27 \mathrm{E}+02$ | $1.27 \mathrm{E}+02$ | 1.27t+0 |
| CM247 | $4-17 \mathrm{E}-04$ | $4-17 E-04$ | 4-17E-04 | 4.17E-04 | 4-17E-04 | $4.17 \mathrm{E}-04$ | $4.17 \mathrm{E}-04$ | $4-17 E-04$ | $4.17 \mathrm{E}-04$ | $4.17 \mathrm{E}-04$ | . $17 \mathrm{E}-0$ |
| CM248 | 7.95E-04 | 7.95E-04 | 7.95E-04 | 7.95E-04 | 7.95E-04 | $7.95 \mathrm{E}-04$ | $7.95 \mathrm{E}-04$ | 7.95E-04 | 7.95E-04 | 7.95E-04 | . 9 |
| BK249 | 9.91E-22 | $9.28 \mathrm{E}-22$ | 8.69E-22 | $7.61 \mathrm{E}-22$ | $6.64 \mathrm{E}-22$ | $5.42 \mathrm{E}-22$ | $4.43 \mathrm{E}-22$ | $1.98 \mathrm{E}-22$ | $8.83 \mathrm{E}-23$ | $3.14 \mathrm{E}-2 \mathrm{~S}$ |  |
| CF249 | $3.48 \mathrm{E}-03$ | $5-22 \mathrm{E}-03$ | 6.86E-03 | 9.82E-03 | 1.25E-02 | $1.58 \mathrm{E}-02$ | 1.85E-02 | $2.52 \mathrm{E}-02$ | 2.82E-02 | 3.02E-02 |  |
| CF250 | 1.02E-01 | 1.02E-01 | $1.02 \mathrm{E}-01$ | 1.01E-01 | $1.00 \mathrm{E}-01$ | $9.87 \mathrm{E}-02$ | $9.74 \mathrm{E}-02$ | $9.24 E-02$ | $8.76 \mathrm{E}-02$ | 6.05E-02 | 3.56 |
| CF251 | 6.91E-04 | 6.91E-04 | 6.91E-04 | 6.91E-04 | 6.91E-04 | $6.91 \mathrm{E}-04$ | 6.91E-04 | 6.90e-04 | $6.90 \mathrm{E}-04$ | 6.86E-04 | $6.81 \mathrm{E}-04$ |
| CF252 | 1.30E-01 | $1.27 E-01$ $3.90 E-22$ | 1.25E-01 | 1-19E-01 | 1.14E-01 | $1.07 \mathrm{E}-01$ | $1.00 \mathrm{E}-01$ | 7.71E-02 | $5.93 \mathrm{E}-02$ | 9.48E-03 | $6.90 \mathrm{E}-0$ |
| CF253 | $1.27 \mathrm{E}-21$ | 3.96E-22 | $1.23 \mathrm{E}-22$ | 1.19E-23 | 1.07E-24 | $2.97 E-26$ | 8.61E-28 | $5.83 \mathrm{E}-34$ | $3.80 \mathrm{E}-40$ | 0.0 | 0.0 |
| CF254 ES 253 | $2.30 \mathrm{E}-25$ | 1-98E-25 | 1.41E-25 | 7.07E-26 | 3-48E-26 | 1.21E-26 | 4.27E-27 | 6.52E-28 | 9.85E-31 | i.88E-43 | 1.26 |
| ES253 | 1.19E-02 | $1.00 \mathrm{E}-02$ | 5-42E-03 | 1.10E-03 | 1.73E-04 | 9.55E-06 | 5.00E-07 | 2.82E-12 | $1.39 \mathrm{E}-17$ | , | - |
| TOTAL | $1.11 \mathrm{E}+08$ | -95E+07 | 8.91E+07 | .21E*07 | .84E+07 | - $38 \mathrm{E}+07$ | . 41 E+0 | . $78 \mathrm{E}+0$ | -43E+ |  |  |

## REFERENCE PWR EQUILIBRIUM FUEL CYCLE -- WASTE DECAY TIMES ALPHA-N NEUTKON SOURCE IN DISCHARGED FUEL, NEUTRONSISEC

BASIS = MT CF HEAVY METAL CHARGED TO REACTOR

|  | INITIAL | 5. Y | 10. Y | . | 90. r | 190. | 490. | 0. | 1990. | 49 | 9990. Y |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 81211 | $3.09 \mathrm{E}-02$ | 4.23E-02 | b.çe-02 | 7.95E-02 | 1.06E-01 | 1.11E-01 | $1.15 \mathrm{E}-01$ | $1.21 \mathrm{E}-01$ | $1.33 \mathrm{E}-01$ | $1.69 \mathrm{E}-01$ | 2.31t-01 |
| -1212 | $4.01 \mathrm{E}-01$ | $0.93 \mathrm{E}-02$ | $1.49 \mathrm{E}-02$ | 3.61E-03 | 2.02E-03 | 7.72E-04 | $4.30 \mathrm{~F}-05$ | 3.57E-07 | $1.19 \mathrm{E}-08$ | 2.53E-08 | $5.16 \mathrm{E}-08$ |
| 81213 | 4.15 E-12 | $4.25 E-12$ | $4.52 \mathrm{E}-12$ | 7.38E-17 | $3.34 \mathrm{E}-11$ | 1.39E-10 | $9.73 \mathrm{E}-10$ | $4.32 \mathrm{E}-09$ | 1.90E-08 | $1.23 \mathrm{E}-07$ | 4.46E-07 |
| P0210 | $5.43 \mathrm{E}-05$ | $1.35 \mathrm{E}-04$ | 2.53E-04 | 1.U0E-03 | $4.47 \mathrm{E}-03$ | $1.15 \mathrm{E}-02$ | $4.05 \mathrm{E}-02$ | $1.04 \mathrm{E}-01$ | 2.86F-01 | 1.08E*00 | 2. $22 \mathrm{E}+00$ |
| P0211 | 1.44E-04 | 1.99E-04 | 2.45E-04 | 3.71E-04 | $4.93 \mathrm{E}-04$ | 5.18E-04 | 5.37E-04 | 5.65E-04 | 6.20F-04 | $7.88 \mathrm{E}-04$ | 1.08E-03 |
| P0212 | 1.13E*02 | 1.96E+01 | $4.20 t+00$ | $1.02 \mathrm{E}+00$ | 5.71E-01 | 2.18E-01 | $1.21 \mathrm{E}-02$ | $1.01 \mathrm{E}-04$ | $3.35 \mathrm{E}-06$ | 7.15E-06 | 1.46E-05 |
| P0213 | 7.54E-04 | 7.77e-04 | 8.27E-04 | 1.35E-03 | $6.11 \mathrm{E}-03$ | 2.53E-02 | 1.78t-01 | 7.91t-01 | $3.48 \mathrm{E}+00$ | $2.24 E+01$ | 8.16E +01 |
| P0214 | $1.62 \mathrm{E}-0.3$ | 2.80E-03 | $4.15 \mathrm{E}-03$ | 9.24t-03 | 2.49E-02 | $5.30 \mathrm{E}-02$ | $1.57 \mathrm{E}-01$ | $4.00 \mathrm{E}-01$ | $1.11 \mathrm{E}+00$ | $4.18 \mathrm{E}+00$ | 1.01E+01 |
| P0215 | 4.66E-02 | 6.41E-02 | 7.91E-02 | 1.2LE-01 | $1.59 \mathrm{E}-01$ | 1.67E-01 | 7.73t-01 | $1.82 \mathrm{E}-01$ | 2.0UE-01 | $2.54 E-01$ | $3.4 \mathrm{kt}-01$ |
| P0216 | $6.85 \mathrm{E}+01$ | $1.18 \mathrm{E}+01$ | $2.54 \mathrm{E}+00$ | 6.181-01 | 3.46E-01 | 1.32E-01 | $7.35 \mathrm{E}-03$ | $6.10 \mathrm{e}-05$ | $2.03 \mathrm{E}-06$ | 4.33F-06 | $8.82 \mathrm{~F}-06$ |
| P0218 | 6.52t-04 | 1.16E-03 | 1.67E-03 | 3.73E-03 | $1.01 \mathrm{E}-02$ | $2.14 \mathrm{E}-02$ | $0.33 \mathrm{t}-02$ | 1.62E-01 | 4.47E-01 | $1.69 \mathrm{E}+0 \mathrm{O}$ | $4.89 \mathrm{E}+00$ |
| A1217 | $4.16 \mathrm{t}-04$ | 4.26E-04 | 4-53E-04 | 7.40E-04 | 3.35E-03 | 1.39t-02 | $9.75 \mathrm{E}-02$ | $4.34 \mathrm{E}-01$ | $1.91 \mathrm{E}+00$ | 1.23E+01 | 4.47E+01 |
| RN219 | $3.47 \mathrm{E}-02$ | $4.78 \mathrm{E}-02$ | 5.90E-02 | 8.94E-02 | $1.19 \mathrm{E}-01$ | 1.25f-01 | 1.29E-01 | $1.36 \mathrm{E}-01$ | $1.49 \mathrm{E}-01$ | $1.90 \mathrm{E}-01$ | 2.60E-01 |
| RN220 | $5.20 E+01$ | $8.99 E+00$ | 1.93E+00 | $4.69 \mathrm{E}-01$ | 2.63E-01 | 1.00e-01 | 5.58f-03 | $4.63 \mathrm{E}-05$ | $1.54 \mathrm{E}-06$ | 3.29E-06 | 6.70E-06 |
| RN222 | $4.70 E-04$ | B.38E-04 | 1.21E-03 | 2.69E-03 | 7.26E-03 | 1.54E-02 | 4.5te-02 | 1.17E-01 | 3.23E-01 | $1.22 \mathrm{E}+00$ | $2.95 \mathrm{E}+00$ |
| FR221 | $2.80 \mathrm{E}-04$ | 2.86F-04 | $3.05 \mathrm{E}-14$ | $4.98 \mathrm{E}-04$ | 2.25E-03 | $9.34 \mathrm{t}-03$ | $0.56 \mathrm{E}-02$ | ?.49E-01 | $1.28 \mathrm{E}+00$ | $8.27 E+00$ | 3.01E+09 |
| RA223 | 2.01E-02 | $2.77 \mathrm{E}-02$ | 3.41E-02 | 5.17E-02 | 6.87E-02 | 7.21t-02 | 7.48E-02 | 7.86E-02 | 8.63E-02 | 1.10E-01 | T.50E-C1 |
| RA224 | $3.59 \mathrm{~F}+01$ | $6.21 E+00$ | $1.33 E+00$ | 3.24t-01 | 1.81E-01 | $0.92 \mathrm{E}-02$ | 3.86E-03 | 3.20E-05 | 1.06E-06 | 2.27E-00 | $4.63 \mathrm{E}-06$ |
| RAP26 | $2.85 E-04$ | 5.07E-04 | 7.30E-04 | 1.t3E-03 | 4.39E-03 | $0.34 \mathrm{E}-03$ | 2.76E-02 | 7.06E-02 | $1.95 \mathrm{t}-01$ | 7.36E-01 | $1.78 \mathrm{E}+00$ |
| AC225 | 2.0SE-04 | 2.10E-04 | 2.23E-04 | 3.05E-04 | $1.65 \mathrm{E}-03$ | $0.85 \mathrm{E}-03$ | $4.81 \mathrm{E}-02$ | $2.14 \mathrm{E}-01$ | $9.39 \mathrm{E}-01$ | $6.06 E+00$ | $2.206+01$ |
| A C227 | $2.59 \mathrm{t}-11$ | $3.56 E-11$ | $4.39 \mathrm{E}-11$ | $6.60 \mathrm{E}-11$ | 8.85E-11 | $9.29 E-11$ | 9.64E-11 | $1.01 \mathrm{E}-10$ | $1.11 \mathrm{E}-10$ | $1.41 \mathrm{E}-10$ | 1.94E-10 |
| TH227 | $2.19 t-02$ | 3.02E-02 | 3.72E-02 | 5.64E-02 | 7.49F-02 | $7.87 \mathrm{E}-02$ | $8.16 \mathrm{~F}-02$ | 8.58E-02 | 9.42E-02 | 1.208-01 | 1.64F-61 |
| 1H228 | $3.02 \mathrm{E}+01$ | $5.20 E+00$ | $1.12 E+00$ | 2.73E-01 | 1.53E-01 | 5.83E-02 | 3.25t-03 | $2.69 \mathrm{E}-05$ | 8.95E-07 | $1.91 \mathrm{E}-00$ | 3.90E-00 |
| 1H229 | 1.24E-04 | 1.27E-04 | 1.35E-04 | 2.21E-04 | 9.9RE-04 | $4.14 \mathrm{E}-0.3$ | 2.91E-02 | $1.29 \mathrm{E}-01$ | 5.68E-01 | $3.67 \mathrm{E}+00$ | $1.33 \mathrm{E}+01$ |
| 1H230 | 9.52t-02? | 9.57E-02 | $9.61 \mathrm{E}-02$ | $9.79 \mathrm{E}-02$ | 1.05E-01 | 1.20E-01 | 1.76E-01 | 2.80E-C1 | 4.91E-0.1 | $1.11 \mathrm{E}+00$ | $2.10 E+00$ |
| 1H232 | $1.05 \mathrm{t}-07$ | $1.05 \mathrm{E}-07$ | $1.05 \mathrm{E}-07$ | 1.07E-07 | 1-13f-07 | $1.22 \mathrm{E}-\mathrm{C} 7$ | 1.49E-07 | 1.96E-07 | 2.90E-07 | 6.34E-07 | -29E-06 |
| PA 231 | $4.12 \mathrm{E}-02$ | $4.12 \mathrm{E}-02$ | 4.12E-02 | $4.13 \mathrm{E}-02$ | 4.16E-02 | 4.20E-02 | 4.34E-02 | 4.56 -02 | 5.00F-02 | $6.36 \mathrm{E}-02$ | $8.71 \mathrm{t}-02$ |
| 4232 | 3.07E-01 | 3.07E-01 | 2.972-01 | $2.47 \mathrm{E}-01$ | 1.38E-01 | 5.29E-02 | 2.94E-03 | 2.39E-05 | 19 | .49E-22 | $5.57 \mathrm{E}-43$ |
| 4233 | 7.43E-04 | 9.79E-33 | 1.89E-02 | 5.59E-02 | 1.73E-01 | 3.83E-01 | $1.11 \mathrm{E}+00$ | $2.54 \mathrm{E}+00$ | 5.705+00 | 1.5SE+01 | $3.17 E+C .1$ |
| U234 | $9.8 U E+0 U$ | $1.02 \mathrm{E}+09$ | $1.07 \mathrm{E}+01$ | $1.23 \mathrm{E}+09$ | 1.61E+01 | $2.03 \mathrm{E}+01$ | $2.51 \mathrm{E}+01$ | $2.64 E+01$ | $2.65 \mathrm{~F}+01$ | $2.63 \mathrm{E}+01$ | $2.60 \mathrm{~F}+\mathrm{C} 1$ |
| U235 | $1.76 \mathrm{E}-01$ | $1.76 \mathrm{E}-\mathrm{J} 1$ | 1.76E-01 | 1-76E-01 | 1.70E-01 | $1.77 \mathrm{E}-01$ | 1.78E-01 | $1.79 \mathrm{t}-01$ | 1.83E-01 | $9.97 t-01$ | 2.22F-01 |
| U236 | $2.73 F+00$ | $2.73 E+30$ | $2.74 \mathrm{E}+00$ | $2.74 E+00$ | $2.75 E+00$ | $2.78 \mathrm{E}+00$ | $2.85 \mathrm{E}+00$ | $2.97 \mathrm{E}+00$ | $3.19 \mathrm{~F}+00$ | $3.73 \mathrm{E}+00$ | $4.33 \mathrm{E}+00$ |
| U238 | $2.32 \mathrm{E}+00$ | $2.32 \mathrm{E}+00$ | $2.32 E+00$ | $2.32 \mathrm{E}+00$ | 2.32E+00 | $2.32 \mathrm{E}+0 \mathrm{C}$ | $2.3 \angle E+00$ | $2.32 \mathrm{~F}+00$ | $2.32 \mathrm{E}+00$ | $2.32 \mathrm{E}+00$ | C.32E+00 |
| VF237 | $4.37 t+02$ | $4.40 \mathrm{E}+0$ | $4.42 \mathrm{E}+0$ | $4.54 \mathrm{E}+02$ | $4.86 \mathrm{E}+02$ | $5.34 \mathrm{E}+02$ | $6.38 \mathrm{E}+02$ | $7.31 \mathrm{E}+02$ | $7.91 \mathrm{E}+02$ | $8.06 E+02$ | ¢. $8.05 E+02$ |
| PU236 | 7.30E-01 | 2.16E-01 | 6.41E-02 | $4.95 \mathrm{E}-04$ | 2.27F-1U | -.21E-21 | 0.0 | 0.0 | 0.0 | 0.0 | $8.05 E+02$ 0.0 |
| PUd38 | $5.36 \mathrm{E}+04$ | $5.21 \mathrm{E}+04$ | $5.06 t+04$ | $4.32 t+04$ | $3.23 \mathrm{E}+04$ | 1.88E+04 | $4.10 \mathrm{E}+03$ | $3.85 E+02$ | $1.79 \mathrm{E}+00$ | -51E-06 | 0.0 $5.63 E-10$ |
| Pu239 | $5.05 \mathrm{~F}+03$ | $5.06 \mathrm{E}+33$ | $5.006+03$ | $5.08 \mathrm{E}+03$ | $5.12 \mathrm{E}+03$ | $5.18 \mathrm{E}+03$ | $5.38 \mathrm{E}+03$ | $5.68 \mathrm{E}+03$ | $6.22 E+03$ | $7.41 \mathrm{E}+03$ | 5.63 8.36 |
| PU240 | 7.56t+03 | $8.80 \mathrm{E}+03$ | $0.81 \mathrm{E}+03$ | 1. $24.2+04$ | $1.43 E+04$ | $1.44 \mathrm{E}+0.4$ | $1.40 \mathrm{E}+04$ | $9.33 E+04$ | $1.20 \mathrm{E}+04$ | $8.80 \mathrm{CoE}+03$ | $5.27 E+0$ ? |
| Pu241 | $2.84 \mathrm{E}-16$ | $2.24 E-16$ | 1.77t-16 | $6.84 \mathrm{E}-17$ | 4.10t-18 | 1.82t-19 | 1.44E-19 | $1.38 \mathrm{E}-19$ | 1.27E-19 | $9.80 \mathrm{E}-2 \mathrm{U}$ | $5.278 E-2 U$ |
| PUZ42 | $1.80 \mathrm{E}+01$ | $1.80 E+01$ | $1.80 E+01$ | $1.81 \mathrm{E}+01$ | 1.83E+01 | $1.85 \mathrm{E}+01$ | 1.88E+01 | $1.89 \mathrm{E}+01$ | $1.90 \mathrm{E}+01$ | $1.42 \mathrm{E}+01$ | $1.93 \mathrm{E}+01$ |
| PU244 | $6.43 \mathrm{E}-13$ | $3.19 \mathrm{E}-11$ | $6.32 \mathrm{E}-11$ | $1.88 \mathrm{E}-10$ | 5.64F-10 | 1.19E-09 | 3.07E-09 | $6.19 \mathrm{E}-09$ | 1.24E-08 | 3.11E-08 | 6.19E-08 |
| AM241 | $2.89 t+06$ | 2.87E+06 | 2.86E + 06 | $2.79 \mathrm{E}+06$ | $2.54 \mathrm{E}+00$ | $2.17 \mathrm{E}+06$ | $1.34 \mathrm{E}+06$ | 6.02E+05 | $1.22 E+05$ | $1.48 \mathrm{E}+03$ | $3.06 \mathrm{E}+02$ |
| AM243 | $3.27 E+04$ | $3.27 E+04$ | $3.27 E+04$ | $3.26 E+04$ | $3.24 \mathrm{E}+04$ | $3.21 E+04$ | $3.13 E+04$ | $2.99 E+04$ | $2.73 \mathrm{E}+04$ | $2.08 E+04$ | . $32 \mathrm{FF}+04$ |
| CM242 | $2.10 \mathrm{E}+04$ | $2.05 \mathrm{E}+04$ | $2.00 E+04$ | $1.83 \mathrm{E}+04$ | $1.39 E+04$ | $8.81 \mathrm{E}+03$ | $2.24 E+03$ | $2.29 E+02$ | $2.40 \mathrm{E}+00$ | 2.75E-06 | $3.44 \mathrm{E}-16$ |
| CM243 | $8.51 \mathrm{t}+03$ | 7.63E+03 | $6.85 E+03$ | $4.44 \mathrm{E}+03$ | $1.21 E+03$ | $1.39 \mathrm{E}+02$ | 2.10E-01 | 4.16E-06 | $1.63 \mathrm{E}-15$ | 0.0 | - 0 |
| CM244 | $4.05 E+06$ | $3.35 E+06$ | $2.76 E+06$ | $1.28 E+06$ | $1.29 \mathrm{E}+05$ | $2.80 \mathrm{E}+03$ | 2.87E-02 | $1.58 \mathrm{E}-10$ | $3.85 \mathrm{E}-11$ | $9.62 \mathrm{E}-11$ | $1.92 \mathrm{E}-10$ |
| CM245 | $6.97 \mathrm{E}+02$ | $6.97 E+02$ | $6.97 E+02$ | $6.95 \mathrm{E}+02$ | $6.92 E+02$ | 6.86E +02 | $6.69 \mathrm{E}+02$ | $6.42 \mathrm{E}+02$ | $5.90 \mathrm{E}+02$ | $4.59 \mathrm{E}+02$ | 3.02E+02 |
| CM246 | $1.27 \mathrm{E}+02$ | $1.27 \mathrm{E}+02$ | $1.27 E+02$ | $1.26 E+02$ | $1.25 E+02$ | $1.23 \mathrm{E}+02$ | $1.18 \mathrm{E}+02$ | $1.10 \mathrm{E}+02$ | $9.47 \mathrm{E}+01$ | 6.09E+01 | $2.97 \mathrm{E}+01$ |
| CM247 | $4.17 \mathrm{E}-04$ | 4.17E-04 | 4.17E-04 | 4.17E-04 | 4.17E-04 | 4-17E-04 | $4.17 \mathrm{E}-04$ | 4.17E-04 | $4.17 \mathrm{E}-04$ | $4.17 \mathrm{E}-04$ | 4.97E-04 |
| CM248 | 7.95E-04 | 7.95E-04 | 7.95E-04 | 7.95t-04 | 7.95E-04 | 7.95E-04 | 7.95E-04 | 7.94E-04 | 7.92E-04 | 7.88E-04 | 7.80E-04 |
| -6K249 | $3.14 \mathrm{E}-25$ | 5.57E-27 | 9.88E-29 | 9.79E-36 | 9.54E-57 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| CF249 | 3.02E-02 | 2.99E-02 | 2.96E-02 | 2.85E-02 | 2.53E-02 | 2.08E-02 | 1.15E-02 | $4.30 \mathrm{E}-03$ | 6.00E-04 | 1.04E-06 | 8.65 E-11 |
| CF250 | $6.05 \mathrm{E}-02$ | $4.64 \mathrm{E}-02$ | 3.56E-02 | $1.23 \mathrm{E}-02$ | 5.13E-04 | $2.56 \mathrm{E}-06$ | $2.47 \mathrm{E}-10$ | 2.42E-10 | $2.32 \mathrm{E}-10$ | 2.06E-10 | $1.69 \mathrm{E}-10$ |
| CF251 | $6.86 \mathrm{t}-04$ | 6.83E-04 | -.81E-04 | 6.70E-04 | 0.40E-04 | 5.93E-04 | $4.70 \mathrm{E}-04$ | 3-20E-04 | 1.48E-04 | 1.47E-05 | $3.12 \mathrm{E}-07$ |
| CF252 | $9.48 \mathrm{E}-03$ | $2.56 \mathrm{E}-03$ | 6.90E-04 | 3.66E-06 | $5.45 E-13$ | 2.28E-24 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| CF254 | $1.88 \mathrm{E}-43$ | $1.54 \mathrm{E}-52$ | 1.26E-61 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| TOTAL | 7.07E+06 | $6.35 E+06$ | $5.75 \mathrm{E}+06$ | $4.19 \mathrm{E}+06$ | $2.77 \mathrm{E}+06$ | $2.25 E+06$ | $1.40 \mathrm{E}+06$ | 6.53E+05 | $1.69 \mathrm{E}+05$ | $4.00 \mathrm{E}+04$ | $2.86 E+04$ |

neutron source in fuel as a function of time

REGERENCE PGR EQUILIBRIUM FUEL CYCLE -- FUEL decay Times
SPONTANEOUS FISSION NEUTRON SOURCE IN DISCHARGED FUEL, NEUTRONS/SEC basis = MT OF heavy metal charced to reactor

|  |  |  |  |  |  |  |  |  |  | REPRO |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | INITIAL | 30. ${ }^{\circ}$ | 60. D | 120. 0 | 182. ${ }^{\text {d }}$ | $274 . \mathrm{D}$ | 365.0 | 730.0 | 1096. D | 3652. D | 7305. D |
| PU238 | $3.74 \mathrm{E}+05$ | $3.80 E+0 S$ | $3.82 \mathrm{E}+05$ | - $8.86+05$ | $3.89 \mathrm{E}+05$ | $3.91 \mathrm{E}+05$ | $3.93 E+0.5$ | $3.44 E+05$ | $3.91 F+05$ | $3.71 \mathrm{~F}+05$ | $3.43 E+05$ |
| PU240 | $2.04 E+06$ | $2.04 \mathrm{E}+06$ | $2.04 \mathrm{E}+00$ | $2.04 \mathrm{E}+06$ | $2.04 \mathrm{E}+06$ | $2.04 E+06$ | $2.04 E+06$ | 2. $04 \mathrm{E}+06$ | $2.04 \mathrm{E}+06$ | $2.05 \mathrm{E}+46$ | 2.0SE+0G |
| PU242 PU244 | $7.10 E+05$ $2.25 E-07$ | $7.10 E+05$ $2.93 E-07$ | $7.10 \mathrm{E}+05$ $3.00 \mathrm{E}-07$ | 7-10E 4.05 | 7. $90 \mathrm{E}+0 \mathrm{~S}$ | 7-10E+0S | 7.10E+05 | 7.10E+05 | $7.10 \mathrm{E}+05$ | $7.10 t+05$ | $7.10 \mathrm{e}+05$ |
| CM242 | 1.99E+08 | $2.93 E-07$ $1.76 E+08$ | $3.60 E-07$ $1.55 E+08$ | 4. $94 \mathrm{E}-07$ $1.20 E+08$ | $6.33 E-07$ $9.23 E+07$ | 6.39E-07 | $1.04 \mathrm{E}-06$ | 1.86E-U6 | 2.68E-06 | 8.41E-06 | $1.66 \mathrm{E}-05$ |
| CM244 | $3.47 E+08$ | $3.46 \mathrm{E}+08$ | $3.45 E+08$ | 3.43E+08 | 3-41E+08 | $6.24 E+07$ $3.37 E+08$ | $4.24 E+07$ $3.34 E+08$ | $9.02 E+06$ $3.22 E+08$ | $1.94 E+06$ $3.09 E+08$ | $4.27 E+04$ $2.37 \mathrm{t}+08$ | $4.07 E+C 4$ $1.01 F+08$ |
| CM240 | $2.07 E+06$ | $2.07 \mathrm{E}+06$ | $2.07 E+06$ | 2.07E+06 | $2.07 E+06$ | $2.07 E+06$ | $2.07 \mathrm{E}+06$ | 2.07F+06 | $2.07 \mathrm{E}+06$ | $2.07 \mathrm{E}+06$ | 1.07E+08 |
| CM248 CM250 | $8.26 E+03$ $1.25 E-03$ | $8.26 E+03$ $1.25 E-03$ | $8.26 E+03$ | $8.26 E+03$ | $8.26 E+03$ | $8.26 E+03$ | $8.205+03$ | 8.26E 03 | $8.26 E+03$ | $8.26 E+03$ | $8.27 F+03$ |
| CF250 | $4.02 \mathrm{E}+$ U3 | $4.02 \mathrm{E}+03$ | $4.01 E+03$ | $1.25 E-03$ $3.97 E+03$ | 1.25E-03 | $1.25 E-03$ | $1.25 E-03$ | 1.25t-03 | $1=25 \mathrm{E}-03$ | $1.25 \mathrm{E}-03$ | 1.25E-03 |
| CF252 | $2.26 E+05$ | 2.22E+05 | $2.17 \mathrm{E}+05$ | 2.08E+05 | $1.99 \mathrm{E}+05$ | 1.86E+0S | $3.83 E+03$ $1.74 E+0 S$ | $3.64 E+03$ $1.34 F+05$ | 3.45E+03 | $2.38 E+03$ | $1.4 \mathrm{UE}+\mathrm{US}^{\text {a }}$ |
| CF254 | $1.70 \mathrm{E}+02$ | $1.21 E+02$ | $8.57 \mathrm{E}+01$ | $4.31 \mathrm{E}+01$ | $2.12 \mathrm{E}+01$ | $7.38 \mathrm{E}+00$ | $1.74 \mathrm{E}+05$ $2.60 \mathrm{t}+06$ | $1.34 E+05$ $3.97 E-02$ | $1.03 E+05$ $6.00 E-04$ | $\begin{aligned} & 1.65 E+04 \\ & 1.15 t-16 \end{aligned}$ | $\begin{aligned} & 1.20 F+0] \\ & 7.65[-35 \end{aligned}$ |
| total | $5.51 \mathrm{t}+08$ | $5.28 \mathrm{E}+08$ | 5.0SE+08 | $4.08 \mathrm{E}+08$ | $4.38 \mathrm{E}+08$ | $4.05 E+08$ | $3.82 \mathrm{E}+08$ | $3.36 E+08$ | $3.17 \mathrm{t}+08$ | $2.42 \mathrm{E}+08$ | $1.67 E+08$ |
| TOTAL | $6.67 \mathrm{E}+08$ | $6.27 E+08$ | $5.24 E+08$ | $5.40 E+08$ | $4.97 \mathrm{E}+08$ | $4.49 E+08$ | $4.16 \mathrm{t}+08$ | $3.54 \mathrm{E}+08$ | 3.31F*08 | 2.56 tal 8 | $1.80 E+08$ |

neutron suurct in ouel as a funcilun uf ime

Reference pwr equilibrium fuel cycle -- waste decar times
SFONTANEOUS FISSION NEUTRON SOURCE IN DISCHARGED FUEL, NEUTRONS/SEC
RASIS = MT OF HEAVY METAL CHARGED TO REACTOR

|  | REPRO |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | H+7AE | 5. r | 10. Y | 30. Y | 90. Y | 190. Y | 490. Y | 990. Y | 1990. Y | 4990. Y | 9990. Y |
| PU238 | $3.71 E+03$ | $3.60 t+03$ | $3.50 E+03$ | $3.13 \mathrm{E}+03$ | $2.24 E+03$ | $1.30 \mathrm{E}+03$ | $2.84 \mathrm{E}+02$ | $2.66 E+01$ | $1.24 \mathrm{E}-01$ | 3.12E-07 | 3.90E-17 |
| PU240 | $2.05 k+04$ | $2.38 \mathrm{E}+04$ | ?.65E+04 | $3.35 E+04$ | $3.87 \mathrm{E}+04$ | $3.89 E+04$ | $3.78 E+04$ | $3.59 \mathrm{E}+04$ | $3.24 \mathrm{E}+04$ | $2.38 E+04$ | $1.43 \mathrm{E}+04$ |
| PU242 | $7.10 \mathrm{E}+013$ | $7.11 \mathrm{E}+03$ | 7.11E+U3 | $7.14 \mathrm{E}+03$ | $7.21 E+03$ | $7.30 E+03$ | $7.41 \mathrm{E}+03$ | 7.47E +03 | $7.51 \mathrm{E}+03$ | $7.58 \mathrm{E}+03$ | $7.62 F+03$ |
| PU244 | 8.41E-08 | $4.18 \mathrm{E}-\mathrm{DC}$ | -.27E-06 | $2.46 E-05$ | 7.37E-05 | 1.56E-04 | $4.01 \mathrm{E}-04$ | 8.09E-04 | $1.63 \mathrm{E}-03$ | $4.06 t-03$ | $8.09 \mathrm{E}-07$ |
| CM242 | $4.27 E+04$ | $4.17 \mathrm{E}+04$ | $4.08 \mathrm{E}+04$ | $3.72 \mathrm{E}+04$ | $2.83 \mathrm{E}+04$ | $1.79 E+04$ | $4.56 \mathrm{E}+03$ | $4.07 E+02$ | $4.88 \mathrm{E}+00$ | 5.60E-06 | $6.99 \mathrm{E}-16$ |
| CM244 | $2.37 E+08$ | $1.95 \mathrm{E}+08$ | $1.61 E+08$ | 7. $50 \mathrm{E}+07$ | 7.54E+06 | $1.64 \mathrm{E}+05$ | $1.67 E+00$ | 9.20E-09 | $2.25 E-09$ | 5.62F-09 | 1.12E-C8 |
| CM246 | $2.07 \mathrm{E}+06$ | $2.07 \mathrm{E}+06$ | $2.07 E+06$ | $2.06 E+06$ | $2.04 \mathrm{E}+00$ | $2.01 t+06$ | $1.92 \mathrm{E}+06$ | 1.79E+06 | 1.54E+06 | $9.92 \mathrm{E}+05$ | $4.76 \mathrm{~F}+05$ |
| CM248 | $8.20 E+03$ | $8.27 \mathrm{E}+03$ | $\varepsilon .27 E+03$ | $8.26 E+03$ | $8.265+03$ | $8.26 E+03$ | $8.26 E+03$ | $8.25 E+03$ | $8.23 E+03$ | $8.18 \mathrm{E}+03$ | $8.10 \mathrm{E}+03$ |
| CM250 | $1.25 E-03$ | $1.25 E-33$ | $1.25 \mathrm{E}-03$ | $1.25 \mathrm{~F}-03$ | 1.24E-03 | 1.24E-03 | $1.22 \mathrm{E}-03$ | 1.20E-03 | 1.15E-03 | $1.02 \mathrm{E}-03$ | 8.38E-04 |
| CF250 | $2.38 \mathrm{E}+03$ | $1.83 \mathrm{E}+03$ | $1.40 E+03$ | $4.85 \mathrm{~F}+02$ | $2.02 \mathrm{~F}+01$ | $1.01 \mathrm{E}-01$ | 9.72E-06 | 9-51E-06 | 9.14E-06 | 8.11E-06 | $6.65 \mathrm{E}-06$ |
| CF2S2 | $1.65 E+04$ | $4.45 E+03$ | $1.20 E+03$ | $6.36 \mathrm{E}+00$ | 9.49E-07 | 3.97E-18 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| CF254 | $1.15 \mathrm{E}-16$ | 9.38t-26 | 7.67E-35 | 0.0 | 0.0 | U.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| total | $2.39 E+08$ | $1.98 t+08$ | $1.64 E+08$ | 7.72E+07 | $9.66 E+06$ | $2.25 E+06$ | $1.98 E+06$ | $1.84 \mathrm{E}+06$ | $1.59 E+06$ | $1.03 E+06$ | $5.06 E+65$ |
| total | 2.40E+08 | $2.04 \mathrm{E}+08$ | $1.69 E+08$ | $8.14 \mathrm{E}+07$ | $1.24 \mathrm{E}+07$ | $4.50 \mathrm{E}+06$ | 3.38E406 | $2.49 E+06$ | $1.76 E+06$ | $1.07 \mathrm{E}+06$ | $5.34 \mathrm{E}+0.5$ |

## A N N E X 3

THERMAL POWER IN CONTAINERS

The; thermal power of one 150 1iters container has been calculated by the Origen program with some hypothesis as for spectra.

## RESULTS:

They are given in annexed tables.
3.1 - Element thermal power and total thermal power for actinides - Storage period.
3.2 - $d^{\circ}$ for period after reprocessing.
3.3 - $d^{\circ}$ for fission products - Storage period.
3.4 - $d^{\circ}$ for fission products - After reprocessing.

Ex : For 10,40 and 1000 years after discharge, the total thermal power is in watts:

|  | 10 years | 40 years | 1000 years |
| :--- | :---: | :---: | :---: |
| Actinides | 108.0 | 66.2 <br> fission products <br> 1030.0 | 447.0 |
| Total | 1138.0 | 513.2 | 22.6 |

150 1/MTU H L W GLASS CONTAINER

TIIERMAL ENERGY

reference pwr equilierium fuel cycle -- fuel decay times
POWFR $=30.00 \mathrm{MW}$, RURNUP $=33000$. MWD. FLUX $=2.93 E+13 \mathrm{~N} / \mathrm{CM} * 2-$ SEC

|  | charge | discharge | 30. D | 60. D | 120. | 18 |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BI | 0.0 | 1.99E-05 | $2.14 \mathrm{E}-0.5$ | 2.33E-05 | $2.76 \mathrm{E}-05$ | $3.19 \mathrm{E}-05$ | 3.84E-05 | $4.51 E-05$ | $\begin{array}{r} 730=0 \\ 7.29 E-05 \end{array}$ | $\begin{array}{r} 109600 \\ 1.005-04 \end{array}$ | $\begin{array}{r} 3652=0 \\ 2.92 F=04 \end{array}$ | $\begin{array}{r} 7305=D \\ 2.285=04 \end{array}$ |
| PO | 0.0 | $1.135-04$ | 1.21E-04 | $1.31 \mathrm{E}-04$ | $1.56 \mathrm{E}-04$ | $1.80 \mathrm{E}-\mathrm{O}_{4}$ | 3.84E-05 | 4.5.55E-04 |  | $\begin{aligned} & 1.00 E-04 \\ & 5.67 E-04 \end{aligned}$ | $\begin{aligned} & 2.12 F-04 \\ & 1.20 \varepsilon-03 \end{aligned}$ | $\begin{aligned} & 2.28 \mathrm{E}-04 \\ & 1.28 \mathrm{E}-03 \end{aligned}$ |
| AT | 0.0 | 8.26E-09 | 4.78E-09 | 2.47E-09 | $1.53 \mathrm{E}-09$ | $1.50 \mathrm{E}-0 \mathrm{C}$ | 1.54E-09 | 2.59E-09 | 4.12E-04 $1.80 E-09$ | $5.67 \mathrm{E}-04$ $2.00 \mathrm{E}-09$ | $1.20 F-013$ $3.56 F-09$ | $1.29 E-03$ |
| RN | 0.0 | 5.? $1 \mathrm{E}-05$ | 6.12E-05 | -6.66E-05 | $7.92 \mathrm{E}-05$ | $9.13 \mathrm{E}-05$ | 1.1 UE-04 | 1.29E-04 | 2.09E-04 | $2.87 \mathrm{E}-04$ | -. $6.08 \mathrm{E}=04$ | $\begin{aligned} & t-31 E-09 \\ & 6.51 E-04 \end{aligned}$ |
| FR | 3.0 0.0 | $7.41 E-09$ $5.16 E-05$ | $4=29 E-09$ | 2.21E-09 | 1-37E-09 | $1.34 \mathrm{E}=09$ | $1.38 \mathrm{E}-09$ | $1.42 \mathrm{E}-09$ | $1.61 \mathrm{E}-09$ | $1.79 \mathrm{~F}-09$ | 3.20E-09 | $6-51 E-04$ $5.06 E-09$ |
| RA ${ }_{\text {c }}$ | 3.0 | S.16E-05 $7.21 E-09$ | $5.53 E-05$ $4.37 E-09$ | 6.C2e-05 | 7.15E-05 | $8.25 E-05$ | $9.94 \mathrm{E}-05$ | 1.17E-04 | 1.89E-04 | 2.601-04 | 5.49E-04 | 5.80E-04 |
| TH | 0.0 | $4.95 \mathrm{E}-05$ | 5.41E-05 | $5.88 \mathrm{t}-05$ | $6.85 E-05$ | 1.79E-09 $7.90 E-05$ | 1.91E-09 | 2.03E-09 | 2-51E-09 | 2.98E-09 | $6.26 \mathrm{E}-69$ | $1.11 \mathrm{E}-08$ |
| PA | 0.0 | 7.45E-07 | 7.48E-07 | 7.49E-07 | 7.51E-07 | 7.53E-07 | $7.50 \mathrm{E}-07$ | 1.12t-04 | $1.80 \mathrm{E}-04$ | 2.48E-04 | $5.26 \mathrm{E}-04$ | $5.60 \mathrm{E}-04$ |
| U | $5.72 \mathrm{E}-02$ | $3.80 \mathrm{E}-0 \mathrm{C}$ | $3.80 E-02$ | 3.81E-02 | $3.81 \mathrm{E}-0$ く | 3.82E-02 | 3.8.2E-02 | $3.83 \mathrm{~F}-02$ | 7.70E-07 | 7.81E-07 | 8.58E-07 | $9.09 E-07$ |
| NP | 0.0 | 9. $77 \mathrm{E}-03$ | -. 08 E-03 | $9.99 \mathrm{~F}-03$ | $9.99 \mathrm{E}-03$ | $9.99 \mathrm{E}-\mathrm{C} 3$ | $9.99 \mathrm{E}-03$ | $3.85 F-02$ $9.99 F-03$ | 9.99E-03 | $3.89 E-02$ $1.00 \mathrm{E}-02$ | $4.00 E-02$ $1.01 E-02$ | $4.27 E-02$ |
| PU | 0.0 | $1.15 E+02$ | $1.17 \mathrm{E}+02$ | $1.18 \mathrm{t}+02$ | $1.18 \mathrm{E}+02$ | $1.19 \mathrm{E}+\mathrm{C} 2$ | $1.20 E+02$ | $1.20 \mathrm{E}+02$ | $1.20 \mathrm{e}+02$ | 1. $20 \mathrm{O}+02$ | $1.15 \mathrm{E}+02$ | $1.02 E-02$ $1.08 E+02$ |
| AM | 0.0 | $3.46 E+00$ | $3.92 E+00$ | $4.38 E+00$ | $5.30 \mathrm{E}+0 \mathrm{C}$ | $6.23 E+00$ | $7.61 \mathrm{~F}+00$ | 8.95E+00 | $1.42 k+01$ | $1.92 \mathrm{E}+01$ | $4.74 \mathrm{E}+01$ | 7.48E+08 |
| CM | 0.0 | $1.32 \mathrm{E}+03$ | $1.18 t+03$ | $1.05 \mathrm{E}+03$ | $8.31 \mathrm{f}+02$ | $6.58 \mathrm{E}+02$ | $4.72 \mathrm{E}+02$ | $3.47 \mathrm{E}+02$ | $1.36 t+02$ | $8.91 F+01$ | $5.92 \mathrm{E}+\mathrm{Cl}$ | 4.05 COT |
| BK | 0.0 | 2-60E-99 | $2.44 E-09$ | 2.28E-09 | 2.00E-09 | 1.74E-09 | 1.42E-09 | 1.16E-09 | 5.19E-10 | $2.3<2-10$ | $8.23 E-13$ | 2.59E-16 |
| CF | 0.0 | 3. $28 \mathrm{E}-06$ | 3.26E-06 | 3-24E-06 | $3.19 \mathrm{E}-06$ | $3.13 \mathrm{E}-\mathrm{CO}$ | $3.05 \mathrm{E}-06$ | 2.97E-06 | $2.60 \mathrm{E}-00$ | 2.38E-06 | $1.33 \mathrm{E}-06$ | $8.63 \mathrm{E}-07$ |
| ES TOTALS | 0.0 5 - $721-12$ | 1. $21 \mathrm{E}-07$ | $1.02 \mathrm{E}-07$ | S. $52 \mathrm{E}-08$ | $1.12 \mathrm{E}-08$ | 1.76E-09 | $9.72 \mathrm{E}-11$ | 5.09E-12 | 2.87F-17 | $1.42 \mathrm{E}-22$ | 0.0 | 0.0 |
| totals | $5.72 \mathrm{E}-0 \mathrm{c}^{2}$ | $1.44 \mathrm{E}+03$ | 1.30E +03 | 1.17E +03 | $9.55 E+02$ | 7.63E+02 | $5.99 \mathrm{E}+1 \mathrm{~L}$ | $4.76 E+02$ | $2.71 t+02$ | < $28.28+02$ | 2.22E+02 | $2.23 \mathrm{E}+42$ |

REfERENCE PWR EQUILIGRIUM FUEL cycle -- waste decay times
POWER $=30.00 \mathrm{MW}, \mathrm{BURNUP}=33000$. MWD. FLUX $=2.93 E+13 \mathrm{~N} / \mathrm{CM} * 2-\mathrm{SEC}$

REPRO
ELEMENT THERMAL POWER, WATTS
BASIS = MI OF HEAVY METAL CHARGED TO KEACTOR

|  | CHARGE | - | $5 . Y$ | 10. Y | 30. Y | 9 | 190 | 490. Y |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 81 | 0.0 | 2.12E-04 | $3.71 E-05$ | 8.41E-06 | 2.73E-06 | 2-16E-06 | $1.53 \mathrm{E}-06$ | 1.22E-06 | $1.31 \mathrm{E}-00$ | 1.66E-U6 | $3.66 E-16$ | $9.39 E-00$ |
| P0 | 0.0 | ?.20t-03 | 2.08E-04 | $4.52 \mathrm{E}-05$ | $1.19 \mathrm{E}-05$ | 7.69E-06 | $4.58 \mathrm{E}-06$ | 5.08E-06 | $1.25 \mathrm{E}-05$ | 3.92E-05 | 1.95E-C4 | t. $-19 \mathrm{E}-04$ |
| AT | 0.0 | 3. 50E-09 | 3.65E-09 | 3.89F-09 | 6.34E-09 | $2.87 \mathrm{E}-08$ | 1.19E-07 | 8.36t-07 | 3.72E-06 | $1.63 \mathrm{E}-05$ | $1.05 \mathrm{E}-14$ | 2.84t-04 |
| RN | 0.0 | $6.08 \mathrm{E}-04$ | $1.06 E-04$ | $2.31 E-05$ | 6.37E-06 | $4.31 \mathrm{E}-\mathrm{U} 6$ | $2.615-06$ | $2.05 \mathrm{E}-06$ | 3.24F-06 | 6.82E-06 | 2.225-05 | $5.19 \mathrm{E}-05$ |
| FR | 0.0 | 3.20E-09 | 3.27E-09 | 3.49E-09 | 5.69E-09 | $2.58 \mathrm{E}-08$ | $1.07 \mathrm{E}-07$ | 7.50f-07 | $3.33 \mathrm{E}-06$ | 1.46E-05 | . 46 E-05 | 3.4.4E-04 |
| RA | 0.0 | 5.49E-04 | $9.53 \mathrm{E}=05$ | 2-09t-05 | $5.72 \mathrm{E}-00$ | $3.84 \mathrm{E}-06$ | 2.30E-06 | $1.78 \mathrm{E}-06$ | $2.81 \mathrm{E}-06$ | 5.93E-06 | $1.93 \mathrm{f}-05$ | 4.52[-0.5 |
| $A C$ | 0.0 | 0. 20 E-09 | 7.59E-OY | 8. $25 E-09$ | 1.38E-08 | 3.50f-08 | $1.10 \mathrm{E}-07$ | $7.01 \mathrm{E}-07$ | $3.07 \mathrm{E}-00$ | 1.35E-05 | $8.69 \mathrm{k}-05$ | ?.16E-04 |
| TH | 0.4 | 5.26E-J4 | 9.30E-05 | 2.23E-05 | 7.97E-06 | 6.32t-06 | 5.18t-06 | $6.22 \mathrm{E}-06$ | 1.09E-05 | $2.55 \mathrm{E}-05$ | $1.06 \mathrm{E}-54$ | $3.31 \mathrm{E}-04$ |
| PA | 0.0 | 8. S¢E-07 | 8.59E-07 | 8.59E-07 | 8.61E-07 | 6.67F-07 | $8.76 \mathrm{E}-07$ | $9.04 \mathrm{E}-07$ | $9.501-07$ | 1.04E-06 | $1.33 \mathrm{E}-06$ | 1.82E-n6 |
| U | 5.72E-0 | $4.06 E-04$ | $4.17 \mathrm{E}-14$ | 4-28F-04 | 4.67E-04 | $5.02 \mathrm{E}-04$ | 6.68t-04 | $8.03 \mathrm{E}-04$ | $8.71 \mathrm{E}-14$ | $9.54 \mathrm{E}-04$ | 1.20e-03 | $1.59 E-0 \%$ |
| NP | 0.0 | 1.01E-0? | $1.111 \mathrm{E}-02$ | $1.02 \mathrm{E}-02$ | 1.05e-02 | 1-12E-02 | $1.23 \mathrm{t}-02$ | 1.47E-02 | $1.08 \mathrm{~F}-02$ | 1.82E-02 | 1.86t-02 | 1.6UF-U2 |
| PU1 | 0.0 | T.15E+00 | $1.15 E+00$ | $1.14 \mathrm{E}+00$ | $1.10 \mathrm{E}+0 \mathrm{O}$ | $9.26 E-01$ | $7.03 \mathrm{E}-01$ | $4.51 \mathrm{E}-01$ | $3.81 \mathrm{~F}-01$ | $3.60 \mathrm{E}-01$ | $3.21 \mathrm{E}-01$ | $2.70 \mathrm{E}-\mathrm{Cl}$ |
| AM | 0.0 | $4.79 \mathrm{E}+01$ | $4.76 E+04$ | $4.74 \mathrm{E}+01$ | $4-62 E+01$ | $4.22 \mathrm{E}+01$ | $3.60 E+01$ | $2.25 E+01$ | $1.04 \mathrm{E}+01$ | $2.49 \mathrm{E}+00$ | $3.99 \mathrm{E}-01$ | C.43E-61 |
| CM | 0.0 | $5.92 E+01$ | $4.89 E+01$ | $4.05 \mathrm{E}+01$ | $1.89 \mathrm{E}+01$ | $2.08 \mathrm{E}+00$ | 9.67E-01 | $4.15 \mathrm{E}-0$ ? | $1.54 \mathrm{E}-02$ | $1.14 \mathrm{E}-02$ | $8.64 \mathrm{E}-03$ | $5.49 \mathrm{E}-03$ |
| 日K | 0.0 | $8.23 E-13$ | $1.40 E-14$ | 2-59E-16 | 2.57E-23 | 2.50E-44 | 0.6 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| C ${ }^{\text {F }}$ | 0.0 | $1.33 E-06$ | 1.04E-06 | $8.63 \mathrm{E}-07$ | 5.25E-07 | $3.25 E-07$ | $2.63 \mathrm{E}-07$ | 1.47E-07 | $5.71 \mathrm{~F}-08$ | $9.45 \mathrm{E}-0.4$ | $2.30 \mathrm{E}-10$ | 4-40t-12 |
| TOTALS | $5.721-02$ | $1.08 \mathrm{E}+02$ | $9.77 E+01$ | $8.90 E+U 9$ | $6.62 \mathrm{~F}+01$ | $4.52 \mathrm{E}+01$ | $3.69 E+01$ | $2.30 E+01$ | $1.08 \mathrm{E}+01$ | $2.88 E+00$ | 7.49E-01 | 5.41F-01 |

POWER $=30.00 \mathrm{FIW}$, BURNUP $=33000$.MWD, FLUX $=2.93 E+13 \mathrm{~N} / \mathrm{CM} * 2-$ SEC

|  | charge | D ISCharge | 30. 0 | 60.0 | 120.0 | 182. | 274. | 365. D | 730. | 1096. D | 3652.0 | 7305.0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| H | 0.0 | 2.52E-02 | 2.51E-02 | 2.50E-02 | 2.48E-02 | 2.45E-02 | 2.42E-02 | 2.38E-02 | $2.25 \mathrm{E}-02$ | 2.13E-02 | 1.44F-012 |  |
| 2N | 0.0 | $2.70 \mathrm{E}-02$ | 5.90E-07 | 1.29E-11 | $6.13 \mathrm{E}-29$ | 1.43E-30 | 7.26E-45 | 5.27E-59 | 2.0 0 | $2.13 \mathrm{E}-02$ 0.0 | $1.44 r-012$ 0.0 |  |
| 6A | 0.0 | $1.91 \mathrm{E}+01$ | $1.37 \mathrm{E}-05$ | 2.99E-10 | $1.42 \mathrm{E}-19$ | 3-32E-29 | $1.69 \mathrm{E}-43$ | 1.23E-57 | 0.0 | 0.0 | 0 | 0.0 |
| GE | 0.0 | $3.81 F+01$ | 5.83E-19 | 3.85E-38 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 0.0 |
| AS | 0.0 | $4.47 E+02$ | 9.35 E-06 | $2.35 F-11$ | $1.48 \mathrm{E}-22$ | 3.93E-34 | $2.63 \mathrm{t}-51$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 1.0 |
| SE | 0.0 | $5.35 \mathrm{E}+03$ | $1.51 \mathrm{E}-04$ | 1-51E-04 | $1.51 \mathrm{E}-04$ | $1.51 \mathrm{E}-04$ | $1.51 \mathrm{t}-04$ | 1.51E-04 | $1.51 \mathrm{E}-04$ | 1.51E-04 | $1.51 \mathrm{E}-\mathrm{C} 4$ | $7.51 \mathrm{E}-04$ |
| BR | 0.0 0.0 | $4.43 E+04$ $5.18 E+34$ | $2.84 \mathrm{E}-05$ | $2.06 E-11$ | $1.08 \mathrm{E}-23$ | $2.21 E-36$ | 3-28E-55 | 0.0 | 0.0 | 0.0 | 0.0 | C.0) |
| R ${ }^{\text {R }}$ | 0.0 | $5.18 E+04$ $1.16 E+05$ | $1.83 E+01$ $7.67 E-01$ | $1.82 E+01$ $2.52 E-01$ | $1.80 \mathrm{E}+01$ $2.73 \mathrm{E}-02$ | $1.78 E+01$ $2.74 E-03$ | $1.75 E+01$ $9.06 E-05$ | $1-72 \mathrm{E}+01$ | $1.62 \mathrm{E}+01$ | 1. $52 \mathrm{E}+09$ | $9.67 \mathrm{E}+10$ | $5.09 \mathrm{E}+0 \mathrm{C}$ |
| SR | 0.0 | $5.61 \mathrm{E}+04$ | $1.8 .8 \mathrm{t}+03$ | $1.26 E+03$ | 6.22E+02 | 3.28E+02 | $9.06 E-05$ $1.67 E+02$ | $3.12 E-00$ $1.19 E+02$ | $1.28 E-08$ $9.69 E+01$ | $1.28 \mathrm{~F}-08$ $9.44 \mathrm{E}+01$ | $1.28 E-08$ $7.94 \mathrm{E}+01$ | $1.28 E-08$ $6.20 E+01$ |
| Y | 0.0 | 1. U6E +05 | $2.98 E+03$ | $2.23 E+03$ | $1.33 E+03$ | $8.71 \mathrm{E}+02$ | $5.90 \mathrm{E}+02$ | $4.94 E+02$ | $4.35 \mathrm{E}+02$ | $4.24 t+02$ | 3.57E+02 | $6.2 \mathrm{CE}+\mathrm{Cl}$ 2.79 C |
| 2 L | 0.0 | 2.47E 4.04 | $5.21 E+03$ | 3. $79 E+03$ | $2.00 E+03$ | $1.03 \mathrm{E}+03$ | $3.87 E+02$ | $1.40 \mathrm{E}+02$ | $2.99 \mathrm{E}+00$ | - . $05 \mathrm{SE}-\mathrm{O} 2$ | $2.24 E-04$ | $2.24 E-64$ |
| NB | 0.0 | 1.08E+05 | $6.20 E+03$ | $5.24 \mathrm{E}+03$ | $3.28 E+03$ | $1.85 \mathrm{E}+03$ | $7.39 \mathrm{~F}+02$ | $2.87 \mathrm{E}+02$ | $6.12 \mathrm{E}+00$ | 1.24E-01 | $1.49 \mathrm{E}-0.4$ | 2.24E- 24 |
| T0 | 0.0 0.0 | $5.95 E+04$ $7.64 E+04$ | $3.61 E+00$ $7.28 \mathrm{E}-01$ | $2.10 E-03$ $1.01 E-02$ | $7.13 E-10$ $8.69 E-03$ | $1.47 E-10$ $9.69 E-03$ | 1.77E-26 | $2.72 \mathrm{E}-36$ | 0.0 | 0.0 | 0.0 | 0.0 |
| RU | 0.0 | $1.67 \mathrm{E}+04$ | $2.47 \mathrm{E}+03$ | $1.01 E-02$ $1.44 E+03$ | $9.69 \mathrm{E}-03$ | 9.69E-03 | $9.69 E-03$ | $9.64 \mathrm{E}-03$ | $9.69 \mathrm{~F}-03$ | 9.69E-03 | 9.69E-03 | 9-6ye-0.3 |
| RH | 0.0 | $2.815+04$ | $5.76 \mathrm{~F}+03$ | $5.32 \mathrm{E}+03$ | $4.64 \mathrm{t}+03$ | $4.09 E+03$ | $5.25 t+01$ $3.42 \mathrm{t}+03$ | 2.3CE+01 | $8.16 \mathrm{E}+00$ | $4-08 \mathrm{E}+00$ | $3.27 \mathrm{E}-\mathrm{C}$ ? | 3.22F-05 |
| PD | 0.0 | $1.30 E+03$ | $9.18 \mathrm{E}-0.6$ | $9.18 \mathrm{E}-06$ | $9.18 \mathrm{E}-06$ | $9.18 \mathrm{E}-00$ | 9.18E-00 | $9.18 \mathrm{E}-06$ | $9.18 \mathrm{E}-03$ | 7-24E+02 | 79E+00 | $5.84 E-03$ $9.88 t-0 t$ |
| AG | 0.0 | $2.34 E+03$ | $6.70 E+01$ | $5.63 \mathrm{E}+01$ | $4.75 \mathrm{E}+01$ | $4.00 \mathrm{E}+\mathrm{C} 1$ | $3.11 \mathrm{E}+01$ | $2.43 \mathrm{E}+01$ | $8.92 E+00$ | 3.27E+00 | 2.98E-03 | $\begin{aligned} & 9.18 t-0 t \\ & 1.34 t-07 \end{aligned}$ |
| CD | 0.0 | $2.94 t+02$ | $1.26 E+00$ | 7.81e-01 | 3.05e-01 | 1.21E-01 | 3-78E-02 | $1.89 \mathrm{E}-02$ | 1.26E-0.2 | 1.20E-02 | 8.50E-0. | 5.18t-03 |
| 1 N | 0.0 | $1.20 E+03$ | 7.28E-03 | $3.82 \mathrm{E}-03$ | $1.66 \mathrm{E}-03$ | 7.04E-04 | 1.97E-04 | $5.57 E-05$ | 3.53E-07 | 2.21E-09 | $9.03 \mathrm{~F}-25$ | 0.01 |
| SN | 0.0 | $4.17 t+04$ | $3.61 E+01$ | $2.36 E+01$ | $1.62 \mathrm{E}+01$ | $1.15 \mathrm{E}+01$ | $6.89 \mathrm{E}+00$ | $4.16 E+00$ | 5.52E-01 | 7.361-U2 | 5.921-04 | 5.918-04 |
| St | 0.0 | 1.04E+05 | $4-14 \mathrm{E}+01$ | $3.72 \mathrm{E}+01$ | $3.44 E+01$ | $3.23 E+01$ | $2.98 \mathrm{E}+01$ | $2.79 E+01$ | 2-15E+01 | $1.66 E+01$ | $2.77 \mathrm{E}+00$ | 2.2?E-C1 |
| TE | 1.0 0.0 | $9.04 E+04$ $1.00 E+05$ | $1.74 E+02$ $2.69 E+02$ | $1.02 \mathrm{E}+02$ | $4.31 \mathrm{E}+01$ | 2-23E+01 | 1.21E*01 | $8.37 \mathrm{E}+00$ | $4.10 \mathrm{E}+00$ | $2.95 E+00$ | 4.84E-01 | 3.71E-02 |
| XE | 0.0 | $7.00 E+05$ $7.26 E+04$ | $2.69 E+02$ $0.47 \mathrm{E}+01$ | 1.78E+01 | 1.01E-01 | S.09E-04 | $2.48 \mathrm{E}-05$ | $2.46 E-05$ | $2.46 E-05$ | 2.46E-03 | $2.46 E-05$ | 2-46E-05 |
| CS | 0.0 | $1.11 E+05$ | $2.89 \mathrm{E}+03$ | 2.68E+03 | 3.60E-02 | $9.53 \mathrm{E}-04$ | $4-31 \mathrm{E}-0 \mathrm{C}$ | 2.06E-08 | $1.01 \mathrm{E}-17$ | $4.63 \mathrm{E}-27$ | 0.0 | 0.0 |
| Ha | 0.0 | $5.91 \mathrm{E}+04$ | $1.35 \mathrm{E}+03$ | $5.83 \mathrm{E}+02$ | $4.01 E+02$ | $3.92 E+02$ | $3.89 \mathrm{E}+0 \mathrm{~L}$ | $87 \mathrm{E}+02$ | 1.50E+03 | $1.11 \mathrm{E}+03$ | 2-24E+02 | $1.14 E+02$ $2.50 \%$ |
| La | 0.0 | $9.26 E+04$ | $5.81 \mathrm{E}+03$ | 1.14E+03 | $4.44 \mathrm{E}+01$ | $1.55 \mathrm{E}+00$ | 1. $\mathrm{DoE}-02$ | . 69 E-05 | 18F+02 | $3.70 E+02$ | $3.14 \mathrm{E}+02$ | $2.50 \mathrm{C}+\mathrm{Cz}$ |
| Cs | 0.0 | $3.20 E+04$ | $2.29 t+03$ | 1.55E+03 | $8.89 \mathrm{E}+02$ | $0.39 \mathrm{t}+02$ | $4.74 E+02$ | $3.74 \mathrm{E}+02$ | $1.53 \mathrm{E}+02$ | $6.26 E+01$ | 1.23E-01 | C.0 $1.65 E-05$ |
| PR | 0.0 | $4.04 \mathrm{E}+04$ | $8.64 \mathrm{E}+03$ | 7.58E+U3 | $6.43 E+03$ | $5.52 \mathrm{E}+03$ | $4.41 E+03$ | $3.53 \mathrm{E}+03$ | $1.45 \mathrm{E}+03$ | $5.93 \mathrm{E}+02$ | $1.16 \mathrm{E}+0 \mathrm{C}$ | 1. 56 F -04 |
| ND | 0.0 | $5.35 \mathrm{E}+03$ | $2.94 E+02$ | $4.51 \mathrm{~F}+01$ | $1.06 F+00$ | 2-22E-02 | 7.09E-05 | $2.41 \mathrm{E}-07$ | $3.04 \mathrm{E}-17$ | $3.61 \mathrm{E}-27$ | 0.0 | 0.0 |
| PM | 0.0 | $6.75 E+03$ | $3.97 E+n 2$ | $2.43 \mathrm{E}+02$ | $1.21 F+02$ | $7.44 \mathrm{E}+01$ | $5.16 \mathrm{E}+01$ | $4.44 \mathrm{E}+01$ | $3.32 \mathrm{E}+01$ | $2.54 \mathrm{E}+01$ | $3.99 \mathrm{t}+00$ | $2.83 \mathrm{E}-01$ |
| SM | 0.0 | $1.36 E+03$ | ?.21E+00 | 2. $18 \mathrm{E}+00$ | $2.18 \mathrm{E}+0 \mathrm{U}$ | $2.17 E+00$ | $2.17 \mathrm{E}+00$ | 2.17E+00 | $2.15 \mathrm{E}+00$ | $2.13 E+00$ | $2.02 \mathrm{E}+00$ | $1.86 t+00$ |
| EU | 0.0 0.0 | $2.53 E+03$ $1.54 F+01$ | $6.29 E+02$ $4.71 \mathrm{E}-02$ | $2-05 E+02$ | 7.13E+01 | O.22E+09 | - 0 -06E + 01 | $5.98 \mathrm{E}+01$ | $5.59 \mathrm{E}+01$ | $5-27 E+01$ | $3.76 E+01$ | $2.43 \mathrm{~F}+01$ |
| 18 | 0.0 | 1.54E+01 | $4.71 E-02$ $8.25 E+00$ | $4.32 \mathrm{E}-02$ | $3.64 \mathrm{E}-02$ | $3.05 \mathrm{E}-02$ | 2.34E-02 | 1.80E-02 | $6.34 \mathrm{E}-03$ | 2.22E-03 | 1.47E-06 | 4.2LE-11 |
| Dr | 0.0 | $1.44 E+00$ | $2.09 \mathrm{E}-05$ | 6. $-15 \mathrm{E}+00$ $4-06 \mathrm{E}-08$ | $3.45 E+00$ | 1.90E+00 | 7.85E-01 | $3.27 \mathrm{E}-01$ | $9.80 \mathrm{E}-03$ | 2.90E-C4 | $6.18 \mathrm{E}-15$ | 3.46F-30 |
| H0 | 0.0 | $5.49 \mathrm{E}-01$ | $1.68 \mathrm{E}-04$ | 7.85E-06 | 7.49E-06 | $7.49 \mathrm{E}-06$ | 6 | 6 | 0. | 0.0 | 0.0 | 0.0 |
| totals | 0.0 | $1.52 \mathrm{E}+00$ | $4.74 E+04$ | $3.36 E+04$ | 2. $30 \mathrm{E}+04$ | $1.76 E+04$ | $1.30 E+04$ | $1.05 \mathrm{E}+0.4$ | $7.49 E-10$ $5.62 E+03$ | $48 E-06$ $50 E+03$ | $7.45 E-06$ $1.04 F+03$ | 7.41E-06 |

## REfERENCE PWR EQUILIBRIUM fUEL CYCLE -- WASte decay times

POWER $=30.00 \mathrm{MW}, ~ B U R N U P=33000$. MWD, FLUX $=2.93 E+93 N / C M * * 2-S E C$

|  |  | REPRO |  |  | $\begin{aligned} & \text { ENT THERM } \\ & =\quad M T \end{aligned}$ | AL POWER, heavy | WATTS TAL CHAR | To Rea |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | CHARGE | - | 5. Y | 10. Y | $30 . r$ | 90. | 190. Y |  |  |  |  |  |
|  | 0.0 | $1.44 \mathrm{E}-02$ | $1.08 \mathrm{E}-02$ | $8.17 \mathrm{E}-03$ | $2.65 E-03$ | $9.00 \mathrm{E}-05$ | 3.22E-07 | $1.46 E-14$ | $8.44 E-27$ | 0.0 | 2.090. ${ }^{\text {2 }}$ |  |
| SE | 0.0 | $1.51 \mathrm{E}-04$ | $1-51 \mathrm{E}-04$ | 1-51E-04 | $1.51 \mathrm{E}-04$ | 1.51E-04 | 1.51E-04 | $1.50 \mathrm{E}-04$ | 1.50E-04 |  |  |  |
| R ${ }^{\text {B }}$ | 0.0 | 1-28E-08 | $1.28 E-08$ | $1.28 \mathrm{t}-08$ | $1.28 \mathrm{E}-08$ | $1.28 \mathrm{E}-08$ | $1.515-04$ $1.28 \mathrm{E}-08$ | 1.28E-08 | $1.38 \mathrm{E}-04$ | $1.48 E-04$ $1.28 E-08$ | $1.43 E-04$ $1.28 E-08$ | $\begin{aligned} & 1.36 E-04 \\ & 1.28 E-08 \end{aligned}$ |
| SR $Y$ | 0.0 0.0 | $7.94 E+01$ $3.57 E+02$ | $7.02 \mathrm{E}+01$ | 6. $20 \mathrm{E}+01$ | $3.79 \mathrm{E}+01$ | $8-62 E+00$ | 7.31E-01 | 4.47E-04 | 1.96E-09 | $3.80 \mathrm{E}-20$ | 0.0 | $\begin{aligned} & 7.28 \mathrm{E}-08 \\ & 0.0 \end{aligned}$ |
| $\begin{array}{r}Y \\ \text { Z } \\ \hline\end{array}$ | 0.0 0.0 | $3.57 E+02$ $2.24 E-04$ | $3.15 E+02$ $2.24 E-04$ | $2.79 t+02$ $2.24 E-04$ | $1.70 \mathrm{E}+02$ | $3.87 E+01$ | $3.29 \mathrm{E}+00$ | $2.01 E-03$ | 8.83F-09 | 1.71E-19 | 0.0 | C. 0 |
| NB | 0.0 | $1.49 \mathrm{E}-04$ | 1.91E-04 | 2.24E-04 | E-04 | 4 | 2.24E-0.4 | $2.24 E-04$ | $2.23 \mathrm{E}-0.4$ | 2.23E-04 | 2.23E-04 | 2.23t-04 |
| TC | 0.0 | $4.69 \mathrm{E}-03$ | $9.69 E-03$ | $9.69 \mathrm{E}-03$ | $9.69 \mathrm{E}-03$ |  | 3.35E-0.4 | 3.35E-04 | 3.35E-04 | 3.35E-04 | 3.35E-04 | $3.34 \mathrm{E}-04$ |
| RU | 0.0 | $3.27 E-02$ | $1.04 \mathrm{E}-03$ | 3.30E-05 | $3.36 \mathrm{E}-11$ | 3.54E-29 | $9.68 t-03$ 0.0 | $9.67 \mathrm{E}-03$ 0.0 | $9.605-03$ 0.0 | $9.62 \mathrm{~F}-03$ | $9.53 \mathrm{E}-03$ | $9.38 \mathrm{E}-13 \mathrm{l}$ |
| RH | 0.0 | $5.79 \mathrm{E}+00$ | $1.84 \mathrm{E}-01$ | $5.85 E-03$ | 5.95E-09 | 6.27E-27 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | C.0 r .0 |
| PD | 0.0 | $9.18 \mathrm{E}-06$ | 9-18E- 46 | $9.18 \mathrm{E}-0 \mathrm{C}$ | 9.18E-00 | $9.18 t-06$ | 9.18t-06 | $9.18 \mathrm{E}-06$ | $9.18 \mathrm{E}-06$ | $9.18 \mathrm{E}-\mathrm{DO}$ | 9.-17E-6 6 | 9.17E-06 |
| AG $C D$ | 0.0 | $2.98 \mathrm{E}-03$ | $2.00 \mathrm{E}-05$ | $1.34 \mathrm{E}-07$ | 2.73E-16 | 2.40E-33 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| C ${ }^{\text {d }}$ - IN | 0.0 0.0 | $8.50 E-03$ $9.03 E-25$ | $6.63 E-03$ | $5.18 \mathrm{E}-03$ | $9.92 \mathrm{E}-03$ | $9.86 \mathrm{E}-03$ | 0.98E-07 | 2-47E-13 | $4.37 \mathrm{E}-24$ | $1.37 E-45$ | 0.0 | 0.0 |
| SN | 0.0 | 5.92E-04 | $5.91 \mathrm{E}-06$ | $9.19 E-48$ $5.915-04$ | 0.0 $5.91 E-04$ | 0.0 $5.90 E-04$ | 0.0 $5.9 U E-04$ | 0.0 | 0.0 | 0.0 | 0.0 | C.0 |
| SB | 0.0 | $2.77 E+30$ | $7.74 \mathrm{E}-01$ | 2. $22 \mathrm{E}-01$ | $1.20 \mathrm{E}-02$ | 1.07E-02 | $5.9 U E-04$ $1.07 E-02$ | $5.88 \mathrm{~F}-04$ $1.07 \mathrm{E}-02$ | 5.86E-04 | $5.82 \mathrm{~F}-0.4$ | $5.70 t-C 4$ | 5.51E-C4 |
| re | 0.0 | $4.84 \mathrm{E}-01$ | $1.34 \mathrm{E}-01$ | $3.71 \mathrm{E}-02$ | 2.19t-04 | 4.47E-11 | 3.16t-2c | 0.0 | 0.0 | 1.0 | 1.04t-02 | 1.0UE-02 |
| I | 0.0 | $2.46 \mathrm{E}-05$ | $2.46 \mathrm{E}-015$ | $2.46 E-05$ | $2.46 E-05$ | 2.46E-05 | 2.46E-05 | $2.46 \mathrm{E}-05$ | $2.46 \mathrm{E}-05$ | - 20 FF-05 | 0.0 | C.U |
| CS | 0.0 | $2.29 E+32$ | $1.41 E+02$ | 1.14E+0? | 7.00E+01 | 1.75E*01 | 1.74E+00 | $1.83 \mathrm{E}-03$ | $2.46 E-05$ $1.39 E-04$ | $2.46 F-05$ $1.39 \mathrm{~F}-04$ | $2.46 E-05$ $1.39 E-04$ | 2.46E-05 |
| BA | 0.0 | $3.14 \mathrm{E}+02$ | $2.80 \mathrm{E}+02$ | 2. $50 \mathrm{E}+02$ | $1.57 \mathrm{E}+02$ | $3.93 \mathrm{E}+01$ | $3.90 E+00$ | $3.80 \mathrm{E}-03$ | $3.65 \mathrm{E}-\mathrm{C8}$ | $3.37 \mathrm{E}-18$ | $1.39 \mathrm{t}-04$ 0.0 | 1.39t-04 |
| CE | 0.0 | 1.23E-01 | $1.42 \mathrm{E}-03$ | $1.65 E-05$ | $2.97 \mathrm{E}-13$ | 0.0 | U.0 | 0.0 | 0.0 | 0.0 | 0.0 | C.0 |
| PR | 0.0 | $1.16 E+00$ | $1.35 \mathrm{E}-02$ | $1.56 \mathrm{E}-04$ | $2.81 \mathrm{E}-12$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| PM | 0.0 | $3.99 \mathrm{E}+00$ | $1.06 E+00$ | $2.83 \mathrm{E}-01$ | $1.42 \mathrm{E}-03$ | $1.81 \mathrm{E}-10$ | $5.86 \mathrm{E}-22$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| SM | 0.0 | $2.02 \mathrm{E}+00$ | 1.94E +00 | $1.86 \mathrm{E}+00$ | $1.59 \mathrm{E}+00$ | $9.84 \mathrm{E}-01$ | 4-44E-01 | 4.006-02 | 7. 56F-04 | 2.62F-07 | $1.09 \mathrm{E}-17$ | $5.43 \mathrm{E}-35$ |
| EU G0 | 0.0 | 3. $70 \mathrm{E}+01$ | $3.01 \mathrm{E}+01$ | $2.43 E+01$ | $1.02 \mathrm{E}+01$ | 7.56E-01 | 9.92E-03 | 2.25E-08 | $8.78 \mathrm{E}-18$ | $1.34 \mathrm{E}-36$ | 0.0 | 0.0 |
| 18 | 0.0 | $1.47 E-06$ $6.18 E-15$ | $7.85 \mathrm{E}-\mathrm{n} 9$ $1.40 \mathrm{E}-22$ | $4.20 E-11$ $3-47 E-30$ | $3.43 \mathrm{E}-20$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| H0 | 0.0 | 7.45E-06 | $7.43 \mathrm{E}-06$ | 7.47E-30 | 7.32E-00 | 0.078-0.00 | U.0 | 0.0 | 0.0 | 4.0 | 0.0 | 0.0 |
| totals | 0.0 | $1.03 \mathrm{E}+03$ | $8.41 \mathrm{E}+02$ | $7.31 \mathrm{E}+02$ | $4.47 E+02$ | $1.06 \mathrm{E}+\mathrm{O} 2$ | 1.01E+01 | $5.61 E-06$ $7.04 \mathrm{E}-02$ | $4-21 E-06$ $2.26 F-02$ | $2.36 \mathrm{E}-06$ | $4.17 E-07$ $?-14 E-02$ | $2.32 E-0 K$ |

## A N N E X 4

SHIELDING $\gamma$ AND $N$ OF THE
NTL 12 TRANSPORTATION CASK AT 10 YEARS

1 - GAMMA SHIELDING

- According to annex 2, the total amount of $\gamma$ rays at 10 years is for one container :

$$
\frac{4.33 \mathrm{E} 15}{3.7 \mathrm{E} 10}=1.17 \mathrm{E} \mathrm{O5} \mathrm{Ci} X
$$

and for 15 containers (maximum in one cask):

$$
1.17 \mathrm{E} \mathrm{O5} \times 15=1.75 \mathrm{E} 06 \mathrm{Ci}
$$

- Others assumptions:
- The fifteen containersare dispatched in five longitudinal axis, regularly spaced around the main axis of the cask.
- The thickness of the shielding is 300 mm around the internal diameter 1220 mm but, with the thermal evacuator and surrounding neutron shielding compound, the external effective diameter is 2500 mm 。
- So, the dose is calculated at a distance 1250 mm from the main axis, in contact with the external diameter.

The theoretical geometry of the source is decomposed into 125 pieces and the calculation includes the absorption effect of each piece for the others.

The build-up factor is calculated by Taylor formula for the lead shield.

- The calculation is performed by the program $S N$ o18 PROTEC, with a eight energy groups distribution, according to annex 2-4.

| Mev. | 0.3 | 0.63 | 1.10 | 1.55 | 1.99 | 2.38 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Ci} / \mathrm{TY}$ | 4.70 E 03 | 1.08 E 05 | 3.67 E 03 | $2.84 \mathrm{E} \mathrm{O2}$ | $3.32 \mathrm{E}-01$ | $9.43 \mathrm{E}-01$ |

Total dose: $146 \mathrm{mR} / \mathrm{h}$

```
A N N E X 4 (cont.)
```

2 - CALCULATION OF NEUTRON FLUX OF ONE GONTATNER

Are taken into account (see annexed tables 2.6 and 2.8)
$\mathcal{\chi}-\mathrm{N}$ neutron sources coming from $1 \%$ of following nuclides:

Pu 238
Pu 239
Pu 240
Pu 242
The $N / o x$ production ratio is assumed as $2 \cdot 10^{-6}$ that corresponds to 2 N for $10^{6} \alpha$ in glass.
$\alpha-N$ neutron sources coming from $100 \%$ of following nuclides:

Am 241
Cm 242
Cm 244

Spontaneous fission neutron sources coming from following nuclides:

| Cm 242 | Cf 250 | Pu 238 | $(1 \%)$ |
| :--- | :--- | :--- | :--- |
| Cm 244 | Cf 252 | Pu 240 | $(1 \%)$ |
| Cm 246 | Cf 254 | Pu 242 | $(1 \%)$ |
| Cm 248 |  | Pu 244 | $(1 \%)$ |
| Cm 250 |  |  |  |

Supposing the neutrons are fast, the conversion dose factor in the material is assumed:
$0.14 \mathrm{mRem} / \mathrm{h}$ for $1 \mathrm{~N} / \mathrm{cm} 2 / \mathrm{s}$
The total surface of the cylinder is assumed to be approximately 2 m 2 or $2.10^{4} \mathrm{~cm} 2$.

The results of this calculation is given in the following table for times after discharge (in years):
$10,15,20,40,100,200,500,1000,2000,5000,10000$
./...

$$
\mathrm{ANNEX}^{4} \quad(\text { Cont. })
$$

| Years | Ci $\alpha$ | $\alpha / \mathrm{s}$ | $\frac{N(\alpha \cdot N)}{s}$ | $\frac{\text { Spont. } \mathrm{N}}{\mathrm{s}}$ | $\frac{\text { Total } N}{s}$ | $\begin{aligned} & \text { Contact } \\ & \text { dose } / \text { rate } \\ & \text { mRem } / \mathrm{h} \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 10 | 3.78 E 03 | 1.40 E 14 | 2.80 E 08 | 2.39 E O8 | 5.19 E 08 | 3.63 E 03 |
| 15 | 3.36 E O3 | 1.24 E 14 | 2.48 E 08 | 1.98 E O8 | 4.46 E 08 | 3.12 E O 3 |
| 20 | 3.00 E 03 | 1.11 E 14 | 2.22 E 08 | $1.64 \mathrm{E} \mathrm{O8}$ | 3.86 E 08 | 2.70 E 03 |
| 40 | 1.56 E O 3 | 5.77 E 13 | 1.15 E 08 | 7.72 E 07 | 1.92 E 08 | 1.39 E O 3 |
| 100 | 1.34 E O 3 | 4.96 E 13 | 9.92 E O 7 | 9.66 E 06 | 1.09 E 08 | 7.63 E O2 |
| 200 | 1.09 E O 3 | 4.03 E 13 | 8.06 E 07 | 2.25 E 06 | 8.28 E 07 | $5.80 \mathrm{E} \mathrm{O2}$ |
| 500 | 6.71 E O2 | 2.48 E 13 | 4.96 E 07 | 1.98 E 06 | 5.16 E 07 | $3.61 \mathrm{E} \mathrm{O2}$ |
| 1000 | 3.08 E O 2 | 1.14 E 13 | 2.28 E 07 | $1.84 \mathrm{E} \mathrm{O6}$ | 2.46 E 07 | $1.72 \mathrm{E} \mathrm{O2}$ |
| 2000 | 7.15 E 01 | 2.65 E 12 | 5.30 E 06 | 1.59 E 06 | 6.89 E 06 | 4.82 E 01 |
| 5000 | 1.13 E 01 | 4.18 E 11 | 8.36 E 05 | $1.03 \mathrm{E} \mathrm{O6}$ | 1.86 E 06 | 1.30 E 01 |
| 10000 | 8.98 E OO | 3.22 E 11 | 6.44 E 05 | 5.06 E 05 | 1.15 E 06 | 8.05 E 00 |

## 3 - NEUTRON SHIELDING

- We suppose there is no absorption before the iron shield.
- Supplementary neutron snielding $=100 \mathrm{~mm}$ anti-neutron compound
- The neutron and gamma fission + capture dose-rates are calculated by Sabine program.


## RESULTS

- neutron dose-rate : $3.8 \mathrm{mRem} / \mathrm{h}$
- gamma dose-rate : $1.6 \mathrm{mRem} / \mathrm{h}$

4 - TOTAL BIOLOGICAL DOSE-RATE

```
146+3.8+1.6=151.4 mRem/h
```


## A N N E X 5

FLUX AND DOSE RATES $\gamma$ AND N
AROUND LEAD TITANIUM SHIELDED
CANISTER (AT 40 YEARS)

1 - GAMMA DOSE RATE
The calculation has been made by SGN program SN 018 PROTEC, taking into account :

- homogeneous source $=\mathrm{D}=400 \mathrm{~mm} ; \quad \mathrm{L}=1400 \mathrm{~mm}$
glass density : $2.8 \mathrm{~kg} / \mathrm{dm} 3$
containing quantity of fission products equivalent to 1 metric ton of heavy material.
- main shielding: lead thickness $=100 \mathrm{~mm}$
- build-up factor calculated by Taylor formula for lead
- the source is decomposed into 125 pieces and the calculation includes the absorption effect of each piece for the others.
- detector located at 300 mm from the axis-cylinder on contact of the titanium.
- the spectrum gamma is in total curies: (see annex 2)

| Mev. | 0.3 | 0.63 | 1.10 | 1.55 | 1.99 | 2.38 | 2.75 | 3.25 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Ci} / \mathrm{TU}$ | 2.07 E 03 | 4.32 E 04 | 9.54 E 02 | 1.32 E 01 | $3.92 \mathrm{E}-01$ | $9.62 \mathrm{E}-107.59 \mathrm{E}-1192.40 \mathrm{E}-12$ |  |  |

RESULTS

- Total dose rate at the detector : $2450 \mathrm{mRem} / \mathrm{h}$
- According to table of annex 4, the total neutron flux for one canister is at 40 years:

$$
1.92 \mathrm{E} 08 \mathrm{~N} / \mathrm{s}
$$

- The approximate internal surface of the cylinder is:

$$
2 \mathrm{~m} 2=20000 \mathrm{~cm} 2
$$

- The flux per cm2 is:
- on the glass ( 400 mm diameter) $\quad 9.6 \mathrm{E} 03 \mathrm{~N} / \mathrm{cm} 2 / \mathrm{s}$
- on the titanium ( 600 mm diameter) $6.4 \mathrm{E} 03 \mathrm{~N} / \mathrm{cm} 2 / \mathrm{s}$
- Dose-rate calculated by Sabine program on the titanium external surface:
- neutron dose-rate $=558 \mathrm{mRem} / \mathrm{h}$
- gamma (fission + capture) dose-rate $=4 \mathrm{mRem} / \mathrm{h}$

3 - TOTAL BIOLOGICAL DOSE-RATE
$2450+558+4=3012 \mathrm{mRem} / \mathrm{h}$

$1501 / \mathrm{MTU}$ H L W
NEUTRON DOSE-RATE
$\gamma \operatorname{DOSE-RATE}$$\left\{\begin{array}{l}\text { GLASS CANISTER } \\ 10 \mathrm{~cm} \begin{array}{l}\text { LEAD SHIELDING } \\ \text { CONTACT }\end{array}\end{array}\right.$
Dose-rate

## A N N E X 6

## GAMMA AND N SHIELDING OF THE TRANSFER CASKS

1 - UNDERGROUND CANISTER TRANSFER CASK (CANISTER 40 Years)
1.1 - Gamma Shielding

The titanium-lead canister is supposed located inside a 10 cm lead supplementary shielding representing by the transportation cask, the outside diameter of that supplementary shield is 900 mm .

The source is the same as in the Annex 5 and the calculation procedure too.

The gap between the canister and the internal wall of the cask is 50 mm .

RESULTS
Total dose-rate at the detector calculated by program SN 018 : $1.25 \mathrm{mRem} / \mathrm{h}$ at 650 mm from center.
1.2 - N Shielding

According to Annex 5:

- Supplementary neutron shielding: 200 mm anti-neutron compound
- We assume the external diameter of the cask is: 1300 mm

$$
\Sigma=0.13 \mathrm{~cm}^{-1}
$$

- The neutron and gamma dose-rates are calculated by Sabine program.

| neutron dose-rate: $0.34 \mathrm{mRem} / \mathrm{h}$ |  |
| :--- | :--- |
| gamma dose-rate $:$ | $0.07 \mathrm{mRem} / \mathrm{h}$ |

1.3 - Total Biological Dose-Rate

$$
1.25+0.34+0.07=1.66 \mathrm{mRem} / \mathrm{h}
$$

```
2 - CONTAINER TRANSFER CASK (10 years)
    The shielding will be made by :
        250 mm lead
        + 220 mm anti-neutron compound
        The gap between the container and the internal wall of the cask is
        supposed 50 mm.
    So, the external diameter is : }1440\textrm{mm
    2.1 - Gamma Dose-Rate
            Calculated by program SN 018 : 0.60 mRem/h
2.2 - Neutron and Fission + Capture Gamma Dose-Rate
    Calculated by program Sabine:
    - neutron dose-rate : 1.20
    _ gamma dose-rate : 0.65
2.3 - Total Biological Dose-Rate
    0.60 + 1.20 + 0.65 = 2.45 mRem/h
```


## A $\mathrm{N} N \mathrm{E} \mathrm{X}$ 7

## WATER RADIOLYSIS AND ABSORPTION CAPACITY

IN THE WET PACKING OF THE PIT

1 - The quantity of water which would be radiolysed by irradiation in the pit can be estimated in following conditions:

- the total irradiation rate coming from the canister may be known:
- in $\mathrm{Mev} / \mathrm{s} / \mathrm{cm} 2$ at the titanium limit for the $\gamma$ rays (Origen and Protec programs)
- in $\mathrm{Mev} / \mathrm{s}$ at the titanium limit for the N rays (last column of the table Annex 4). Those figures must be divided by a factor 2.72 taking into account the decreasing of energy accross the lead container and multiplied by the mean energy $=8 \mathrm{Mev}$.
- the approximate external surface of the lead container is:

$$
3.0 \mathrm{E} 04 \mathrm{~cm}^{2}=3.0 \mathrm{~m} 2
$$

- according to the Reactor Handbook AECD 3646 p. 673, the ratio of water radiolysis is the following:
. for $\gamma$ rays : 0.3-0.7 molecule per 100 EV
- for fast neutrons : 1.5 molecule per 100 EV

So, we admit a general mean factor equal to 0.8 and we suppose $20 \%$ of the total irradiation energy is delivered into the intersticial water in the packing material, which is probably a pessimistic hypothesis.

The recombination of H 2 and O 2 is not considered.

Results in the next table.

The quantity of radiolysed water per year is very low, in the range of the mg/y.

## A N N E X 7 (Cont.)

## WATER RADIOLYSIS FOR ONE CYLINDER

| Years | $\frac{\text { Total }}{\text { S }}$ | $\frac{\text { Mev N }}{\text { year }}$ | $\frac{\mathrm{Mev} \text { 仡 }}{\mathrm{cm} 2 / \mathrm{s}}$ | $\frac{\mathrm{Mev}}{\text { year }}$ | $\frac{\text { Mev Total }}{\text { year }}$ | $\frac{\mathrm{g} \mathrm{H} 2 \mathrm{O}}{\text { year }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 40 | $1.92 \mathrm{E} \mathrm{O8}$ | 1.78 E 16 | 5.19 E O4 | 1.92 E 15 | 1.97 E 16 | 9.26 E-04 |
| 100 | $1.09 \mathrm{E} \mathrm{O8}$ | 1.01 E 16 | 1.12 E O 4 | 4.14 E 14 | 1.05 E 16 | $4.94 \mathrm{E}-04$ |
| 200 | 8.88 E 07 | 8.24 E 15 | 8.23 E 02 | 3.04 E 13 | 8.27 E 15 | 3.88 E-04 |
| 500 | 5.16 E 07 | 4.79 E 15 | 2.73 E 02 | 1.01 E 13 | 4.80 E 15 | 2.26 E-04 |
| 1000 | 2.46 E 07 | 2.28 E 15 | 2.70 E 02 | 9.96 E 12 | 2.29 E 15 | $1.08 \mathrm{E}-04$ |
| 2000 | 6.89 E 06 | 6.39 E 14 | 2.68 E 02 | 9.92 E 12 | 6.49 E 14 | $3.05 \mathrm{E}-05$ |
| 5000 | 1.86 E 06 | 1.72 E 14 | 2.62 E O 2 | 9.69 E 12 | 1.82 E 14 | 8.55 E-06 |
| 10000 | 1.15 E 06 | 1.07 E 14 | 2.50 E 02 | 9.25 E 12 | 1.16 E 14 | $5.45 \mathrm{E}-06$ |

## N.B.:

- Coefficient transforming $N / s$ in Mev $N / y$ (mean energy : 8 Mev )

$$
86400 \times 365.2 \times 8 / 2.72=9.28 \mathrm{E} \mathrm{06}
$$

- Coefficient transforming Mev/cm2/s in Mev ${ }^{6} / \mathrm{y}$
$86400 \times 365.2 \times 3.0 \mathrm{E} \mathrm{O} 4=9.46 \mathrm{E} 11$
- Conversion coefficient for Mev/y into $\mathrm{g} \mathrm{H}_{2} \mathrm{O}$ $20 \%$ of 0.8 molecule $\mathrm{H}_{2} \mathrm{O}$ for 100 EV
i.e.: in $\mathrm{g} \mathrm{H}_{2} \mathrm{O}$ for 1 Mev:

$$
\frac{0.8 \times 18 \times 1 \mathrm{E} 04 \times 0.2}{6.02 \mathrm{E} 23}=4.78 \mathrm{E}-20
$$

## A N NEXX 7 (Cont.)

## 3 - ABSORPTION CAPACITY

According to Origen program results, the total weight of fission products in one canister corresponding to 1 MT of heavy metal is:

$$
29.1 \mathrm{~kg}
$$

The absorption capacity of ion exchanging is defined as follows:

- zeolite 24.06 milliequivalent for 100 g
- bentonite 64.26 milliequivalent for 100 g
or
- zeolite 0.24 equivalent for 1 kg
- bentonite 0.64 equivalent for 1 kg

We assume the mean atomic mass of the fission products is 110 and the mean valency is 2 .

The absorption possibility will be :

- zeolite $0.24 \times 110 / 2=13.2 \mathrm{~g} / \mathrm{kg}$
- bentonite $0.64 \times 110 / 2=35.2 \mathrm{~g} / \mathrm{kg}$

The theoritical total quantity of absorbing material for one canister will be:

$$
\frac{29.1 \times 1000}{13.2}=2205 \mathrm{~kg} \text { of zeolite }
$$

or

$$
\frac{29.1 \times 1000}{35.2}=826 \mathrm{~kg} \text { of bentonite }
$$

Those quantities are to be compared to the useful volume of the pit ( 1 m diameter x 3 m ).

$$
V=\frac{\pi_{x} 1^{2} \times 3}{4}-\frac{\pi x 0.6^{2} \times 1.6}{4}=2 \mathrm{~m} 3
$$

## AN NE X 8

## HEAT TRANSFER FROM THE CANISTER IN THE PIT

1 - THERMAL TRANSFER IN THE MASS OF A CYLINDER
The thermal power per unit volume is:

$$
\mathrm{p}=\mathrm{dP} / \mathrm{dV}
$$

The transfer coefficient is $\lambda$

Write the equilibrium of thermal effects for:

- an area $d R$ along the radius $R$
- an unit of length

The total thermal power to be transmitted is:

$$
P=\pi R^{2} p
$$

The surface of crossing is:
$S=2 \pi_{R}$

The temperature gap across the area is:

$$
\frac{d \theta}{d R}=\frac{\pi R^{2} \times p}{2 \prod R \lambda}=\frac{R \times p}{2 \lambda}
$$

The total temperature difference between the axis and a surface of radius Row is:

$$
\begin{aligned}
& \Delta \theta=\frac{p}{2 \lambda} \int_{0}^{R o} \quad R d R=\left[\frac{p}{4 \lambda}\right]_{o}^{2} \\
& \Delta \theta_{\mathrm{Ro}}=\frac{\mathrm{p} \mathrm{Ro}^{2}}{4 \lambda}
\end{aligned}
$$

For a cylindrical layer of material between two radius $R 1$ and $R 2$ and with a constant power $P$ per unit of length, the calculation is:

$$
\frac{d \theta}{d R}=\frac{P}{2 \prod R \lambda}
$$

$\Delta \theta \quad{ }_{R 1} 2=\frac{p}{2 \Pi \lambda} \quad \operatorname{Ln}(R 2 / R 1)$

## A N N E X 8 (Cont.)

For such calculations, we make the assumption that the whole behaviour of the system is completely axisymetric.

2 - APPLICATION TO THE DISPOSAL AT 40 YEARS
The total power of a canister is $\approx 520$ Watts (Annex 3)

The transfer coefficient are :

- in the glass : $1.2 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}$
- in the lead : $30 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}$
- in the packing: 0.5 or 1 or $1.5 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}$

So the diagram of temperature differences from the center of the canister to the granite will be approximately:

- in the glass: $p=520 / 150=3.47 \mathrm{~W} / 1=3470 \mathrm{~W} / \mathrm{m} 3$

$$
\text { Ro }=0.2 \mathrm{M}
$$

$\Delta \theta \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots . . \begin{aligned} & 29^{\circ} \mathrm{C}\end{aligned}$ (at 10 years $=64^{\circ} \mathrm{C}$ )

- in the stainless steel contact surfaces
$\Delta \theta$ (estimated) $\ldots \ldots . . . . . . . . . . .$.
- in the lead : $P=520 / 1.6=325 \mathrm{~W} / \mathrm{M}$ $\mathrm{R} 1=0.2 \mathrm{M}$ $\mathrm{R} 2=0.3 \mathrm{M}$
$\Delta \theta$ $1^{\circ} \mathrm{C}$
- in the titanium contact surfaces
$\Delta \theta$ (estimated) $\ldots . . . . . . . . . . . . .$.
- in the packing material

$$
\begin{aligned}
\mathrm{P} & =520 / 20=260 \mathrm{~W} / \mathrm{M} \\
\mathrm{R} 1 & =0.3 \mathrm{M} \\
\mathrm{R} 2 & =0.5 \mathrm{M}
\end{aligned}
$$

$\Delta \theta$

$$
\begin{aligned}
& \text { for } \lambda=0.5 \quad \frac{42^{\circ} \mathrm{C}}{} \begin{array}{l}
\text { for } \lambda=1.0 \quad \frac{21^{\circ} \mathrm{C}}{\text { for }^{\circ} \lambda} \begin{array}{l} 
\\
\text { for }
\end{array}
\end{array} \text { 1.5 } 14^{\circ} \mathrm{C}
\end{aligned}
$$

The total difference of temperature is:

- with $\lambda$ packing $=0.5 \ldots . . . . . . . . . . .$.
- with $\lambda$ packing $=1.0 \ldots . . . . . . . . .$.
- with $\lambda$ packing $=1.5 \ldots . . . . . . . . .$.

For a vertical cylinder in free air, the approximate transfer coefficient is given by:

$$
k=4.5 \sqrt[4]{\frac{\Delta \theta}{T \times H}}
$$

with

$$
\mathrm{P}=\mathrm{k} \times \mathrm{s} \times \Delta \theta
$$

$$
\left\{\begin{aligned}
& \mathrm{k}= \text { coef in watts } / \mathrm{m} 2^{\circ} \mathrm{C} \\
& \Delta \theta= \text { mean temperature } \\
& \text { difference between } \\
& \text { wall and air at } \\
& \text { infinite } \\
& \mathrm{T}= \begin{array}{l}
\text { absolute temperature } \\
\\
\text { of air } \\
\mathrm{H}=
\end{array} \\
& \quad \begin{array}{l}
\text { total height of } \\
\\
\text { cylinder in m }
\end{array} \\
& \mathrm{S}= \text { surface in contact } \\
& \text { ma }
\end{aligned}\right.
$$

So, the complete formula is:

$$
\begin{aligned}
& \text { te formula is: } \\
& P=4.5 \times s \times \Delta \theta \times \sqrt[4]{\frac{\Delta \theta}{T \times H}}
\end{aligned}
$$

and we can obtain $\Delta \theta$ :

$$
\Delta \theta=\left(\frac{\mathrm{P}^{\mathrm{T} \mathrm{~T}^{1 / 4} \times \mathrm{H}^{1 / 4}}}{4.5 \times 5}\right)^{4 / 5}
$$

$\Delta \theta=$
$160^{\circ} \mathrm{C}$

The temperature difference between the center of the glass and the air will be:

$$
\sum \theta=160+64+4=228^{\circ} \mathrm{C}
$$

N. B. : Temperature diff. in glass : see annex 8

By the same way with $P=520$ Watts

$$
\Delta \theta=\quad 72.5^{\circ} \mathrm{C}
$$

The temperature difference between the center of the glass and the air will be:

$$
\theta=72.5+29+2=103.5^{\circ} \mathrm{C}
$$

3 - TEMPERATURE OF GLASS CONTAINERS AT 10 YEARS IN THE NTL 12 CASK

- The cask will include 15 containers at 10 years.
- We suppose an air gap between the containers and the aluminium solid packing, and a quasi direct contact between the packing and the internal surface of the iron cask itself.
- On outside, the surface of the cask is made of radial blades which widely increases the useful contact of thermal evacuation surface.
heat transfer coefficient $:$ in aluminium $: \lambda=146 \mathrm{Watts} / \mathrm{m}{ }^{\circ} \mathrm{C}$ in steel $: \lambda=43 \mathrm{Watts} / \mathrm{m}^{\circ} \mathrm{C}$
3.1 - Temperature difference in aluminium

The total thermal power for the whole arrangement of 15 containers is:
$P=1138 \times 15=17070$ Watts
For a 1 m long section:
$P=17070 / 4.5=3800$ Watts
Approximately half of this quantity has to go through the space between cylinders, in the aluminium packing.

This useful section is:
$5 \mathrm{x}\left[\left(2 \mathrm{x} 0.39 \mathrm{x} \sin 36^{\circ}\right)-0.418\right]=0.20 \mathrm{~m} 2$
The distance to go to the internal surface of the cask is approximately: 0.30 m

The temperature difference in the aluminium will be:
$\Delta \theta_{\mathrm{Al}}=\frac{3800 \times 0.30}{2 \times 0.20 \times 146}=20^{\circ} \mathrm{C}$
A N N E X
9 (Cont.)

## 3.2 - Temperature difference in air gap

We suppose a 2 cm air gap between the external surface of the glass cylinder and the packing.

The thermal transfer is due to:

- conduction
- natural convection
- radiation

Considering the size of the gap and the range of temperature the most important effect is the radiation.

A - Conduction - Convection

The transfer coefficient for steady air is (at $\sim 200^{\circ} \mathrm{C}$ )

$$
\begin{aligned}
& \lambda=0.035 \mathrm{kcal} / \mathrm{h} \mathrm{~m}^{\circ} \mathrm{C} \\
& \lambda=0.040 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}
\end{aligned}
$$



For a 2 cm gap, the combined coefficient is (Cf.Tideström)

$$
\begin{aligned}
& \alpha \mathrm{cc}=\lambda \mathrm{ac} / \mathrm{d}=0.040 \cdot 1.5 / 0.02 \\
& \alpha \mathrm{cc}=3 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}(\sim 6 \text { for each surface })
\end{aligned}
$$

## B - Radiation

Surfaces are supposed made of rough metal, not polished.
Their emission coefficient is approximately $\Sigma=\sim 0.5$
The total power to be transferred is given by:

$$
P=\sum \operatorname{Cn} \cdot\left(T_{1}^{4}-T_{2}^{4}\right) \mathrm{S}
$$

with Cn (black surface emission) $=5.66 \cdot 10^{-8} \mathrm{~W} / \mathrm{m}^{2}{ }^{\circ} \mathrm{K}^{4}$
The total lateral emitting surface of the cylinder is: 1.88 m 2 The complete equilibrium will be given by the formula:

$$
3 x\left(T_{1}-T_{2}\right)+2.83 E 08 \times\left(T_{1}^{4}-T_{2}^{4}\right)=1138 / 1.88
$$

We make an assumption about the possible temperature of the cylinder surface: $\mathrm{cc}=130^{\circ} \mathrm{C}$

```
T1}\approx130+273=403\mp@subsup{}{}{\circ}\textrm{k
```

By iterating computation, we find: $\mathrm{T} 2 \approx 333^{\circ} \mathrm{k}$ So, the temperature difference in the air gap is: $\quad 70^{\circ} \mathrm{C}$

## A N N E X 9 (Cont.)

3.3 - Temperature difference in aluminium-iron contact

We assume a good contact between aluminium packing and cast iron.
So, the temperature difference will be approximately: $\underline{5^{\circ} \mathrm{C}}$
$3.4-\frac{\text { Temperature difference in the stainless steel contact surfaces }}{(\text { assumed }) \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots}$
3.5 - Temperature difference in the cast iron wall of the cask

We apply the second formula of Annex 8:


| with P (per unit of length) | $=3800 \mathrm{~W}$ |
| ---: | :--- |
| $\lambda$ | $=43 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}$ |
| $\mathrm{R}_{1}$ | $=1.220 / 2 \mathrm{~m}$ |
| $\mathrm{R}_{2}$ | $=1.820 / 2 \mathrm{~m}$ |

$$
\Delta \theta=5.62^{\circ} \mathrm{C} \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \approx \ldots \ldots \ldots \ldots{ }^{6} \mathrm{C}
$$

3.6 - Temperature difference outside the cask

The cylindrical surface is increased by a lot of blades in an approximate ratio 8.

The equivalent surface heat exchange coefficient will be approximately:

$$
\mathrm{k}=3 \times 8=24 \mathrm{~W} / \mathrm{m}^{\circ} \mathrm{C}
$$

The temperature difference:

$$
\Delta \theta=\frac{3800}{1.82 \times \pi \times 24}=
$$

3.7 - Total difference between center of containers and atmosphere
$\sum \Delta \theta=64+20+70+4+5+6+28=197^{\circ} \mathrm{C}$
This result is in accordance with the preceding assumption made in $\$ 3.2$ above.

A N N E X 9 (Cont.)

## $4^{-}$- TEMPERATURE OF A CONTAINER AT 10 YEARS IN THE INTERMEDIATE TRANSFER CASK

The diagram of temperature difference from the center of the canister to the atmosphere will be approximately:

- in the glass (see Annex 8)

$$
\Delta \theta=\ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots
$$

- in the stainless steel contact surface

$$
\Delta \theta=\ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots^{4^{\circ} \mathrm{C}}
$$

- in the air gap (see § 2.2)
- in the lead shield ( 20 cm )

- in the polyethylene $N$ shield

$$
\Delta \theta \mathrm{R}_{1}^{\mathrm{R}_{2}} \quad=\frac{\mathrm{P}}{2 \pi \lambda} \operatorname{Ln}\left(\mathrm{R}_{2} / \mathrm{R}_{1}\right)
$$

with $\mathrm{P}=\quad 760 \mathrm{~W}$

| $\lambda$ | $=$ | $0.4 \mathrm{~W} / \mathrm{m} \mathrm{o}^{\circ} \mathrm{C}$ |
| :--- | :--- | :--- |
| $\mathrm{R}_{1}$ | $=$ | 1.0 m |
| $\mathrm{R}_{2}$ | $=$ | 1.4 m |

$\Delta \theta=\ldots \ldots \ldots \ldots \ldots \ldots \cdot \ldots \cdot 101^{\circ} \mathrm{C}$
This last figure is high but could be decreased by thermal evacuating blades as for the big transportation cask:

In that case, we assume $\Delta \theta=\ldots . . .$|  |
| :--- | :--- |
| $0^{\circ} \mathrm{C}$ |

- in the outside contact surface with atmosphere

$$
\Delta \theta=\frac{760}{3 \times \pi \times 1.4} \quad=\ldots \ldots \ldots \ldots \ldots \ldots \ldots .
$$

The total difference will be:

$$
\sum \Delta \theta=64+4+70+7+30+56=\ldots \ldots \ldots \ldots \ldots+231^{\circ} \mathrm{C}
$$

## A N N E X 9 (Cont.)

5 - TEMPERATURE OF A LEAD-TITANIUM GLASS CANISTER AT 40 YEARS IN THE
UNDERGROUND TRANSFER CASK
The diagram of temperature differences from the center of the canister
to the atmosphere will be approximately:
- in the glass (see annex 8)
$\Delta \theta=$
$29^{\circ} \mathrm{C}$
- in the stainless steel contact surfaces

- in the lead

- in the titanium contact surfaces

- in the air gap between titanium and internal surface
of cask (see § 2.2 above)
$3 \mathrm{x}(\mathrm{T} 1-\mathrm{T} 2)+2.83 \mathrm{E}-08 \mathrm{x}\left(\mathrm{T} 1^{4}-\mathrm{T} 2^{4}\right)=520 / 1.50$
assuming $\mathrm{T} 1 \sim 100^{\circ} \mathrm{C}$
we find $\mathrm{T} 2=52^{\circ} \mathrm{C}$
and the difference:


- in the complementary lead shield of cask

    - in the polyethylene $N$ shielding of cask
$\Delta \theta_{R 1}^{R 2}=\frac{P}{2 \pi \lambda} \operatorname{Ln}(R 2 / R 1)$
with: $P$ (per unit of length) $=346 \mathrm{~W}$
$\lambda=\sim 0.4$
$\mathrm{R} 1=0.80 \mathrm{~m}$
R2 $=1.20 \mathrm{~m}$
$\Delta \theta=$........................................................... $55^{\circ} \mathrm{C}$
N.B.: This last figure is high, it will be useful to provide thermal
evacuating blades on the cask in the same way as for the main
transportation are
In that case, we assume $\Delta \theta=$
$15^{\circ} \mathrm{C}$

```
A N N E X }9\mathrm{ (Cont.)
```

- in the outside contact surface with atmosphere

$$
\Delta \theta=\frac{346}{3 \times \times 1.2}=\ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots
$$

The total difference will be:

$$
\sum \Delta \theta=29+2+1+2+48+1+15+30=\cdots \cdots{ }^{128^{\circ} \mathrm{C}}
$$

6 - NORMAL COOLING BY FORCED AIR FLOW IN THE TEMPORARY STORAGE
Each main storage cell is provided as follows:

- 3000 containeis of 150 l each (1 MT equivalent U)
- The containers are put into vertical pits by set of 10 - There are 300 pits.
- The charging of pits is performed at 10 years for all containers, but all along a 10 years period.
- So the final composition of the storage includes containers from 10 years to 20 years.
- At 10 years the thermal power is 1138 W

At 20 years the thermal power is 850 W

- So the approximate logarithmic mean power is, per container, 1000 W and the total thermal power when the cell is full, is ..... 3000 kW
- The forced ventilation is calculated for an entering air at $20^{\circ} \mathrm{C}$ an exit air at ... $80^{\circ} \mathrm{C}$ and the total flow-rate will be:

$$
\text { Qo }=\frac{3000 \times 860}{60 \times 0.29}=\underline{150.000 \mathrm{Nm} 3 / \mathrm{h}}
$$

- The total pressure drop in the circuit will be, approximately 35 mbars including:

- Mean surface container temperature .................................. $48^{\circ} \mathrm{C}$
- Upper surface containet temperature ..................................... $72^{\circ} \mathrm{C}$


## A $\mathrm{N} N \mathrm{E} \mathrm{X} \quad 9$ (Cont.)

7 - VENTILATION BY ONE FAN (HALF BREAK-DOWN)
With one fan operating instead of two, we may assume the air flow-rate is approximately $65 \%$ of the total flow-rate because of falling down pressure drop.

In such conditions, the exit temperature is increased up to $112^{\circ} \mathrm{C}$ (instead of $80^{\circ} \mathrm{C}$ ).

In the case of the most powerful filling with 1000 W mean container thermal power, the other figures are:

- mean surface container temperature: .................................. $77^{\circ} \mathrm{C}$
- upper surface container temperature: ........................... $123^{\circ} \mathrm{C}$
and in the case of an half thermal power:
- temperature of exit air .................................................. $66^{\circ} \mathrm{C}$
- mean surface container temperature...................................... $48^{\circ} \mathrm{C}$
- upper surface container temperature.................................... $72^{\circ} \mathrm{C}$

8 - TRANSIENT EFFECT OF VENTILATION BREAK-DOWN
We assume the forced ventilation is completely stopped at a time to $=0$ and we look at the transient effect on the temperature of the air and the temperature of the container.

We assume the filling of pits homogeneous.

The air flow gets an equilibrium for each mean temperature of the containers but the air cannot evacuate the whole thermal power. So the exceeding quantity of heat remains in the containers and the temperature of which increases.

Later on, the system gets a new constant flow in an asymptotic way.
The dynamic calculation has been programmed and performed in following conditions:

- pressure drop in cool part ( $\mathrm{Zo}=15$ mbars in normal flow) function of mass air flow-rate at power 2
- pressure drop in hot part ( $Z^{\prime} o=10$ mbars in normal flow) function of mass air flow-rate at power 1
- useful evacuating surface of a container: 2.1 m 2
- heat transfer coefficient given by the Kern formula: $K=0.027 \times \frac{c_{0}^{0,666} \times C_{5}^{0,333} \times v^{0,8} \times M_{s}^{0,8}}{D e^{0,2} \times H 0,666 \times 1,62}$


## A N N E X 9 (Cont.)

The equilibrium of flow is given by:

$$
z+z^{\prime}=\left(\int_{\text {sair }}-\left(20^{\circ} \mathrm{C}\right) x \mathrm{~g} x \frac{\mathrm{H}}{100}\right.
$$

with:
$Z$ and $Z^{\prime}$ function of mass air flow-rate
$g=$ gravity coefficient
$\mathrm{H}=$ useful height of chimney
(assume $\left.\begin{array}{c}30 \mathrm{~m} \text { underground } \\ 20 \mathrm{~m} \text { outside }\end{array}\right\}$

The computing has been made with a spacing of time $=1 \mathrm{~h}$
Other assumptions:

- volumic mean mass of container.............................. $2.600 \mathrm{~kg} / \mathrm{m} 3$
- mass heat capacity ............................................ $0.18 \mathrm{kcal} / \mathrm{kg}^{\circ} \mathrm{G}$

RESULTS : Annexed diagrams give the curves of temperature of:

- exit air
- mean container surface
- warmest container surface.




SAINT GOBAIN
TECHNIQUES NOUVELLES - 1055/77

## A N N E X <br> 10

DRAWINGS

Drawing no.
Title

1879 D 0010
1879 D 0011
1879 D 0012/16
1879 D 0013
1879 D 0014
1879 D 0015
1879 D 0017
1879 D 0020
1879 D 0021
1879 D 0022
General perspective
Lay-out of the incells installation
Containers receiving and temporary storage
Final conditioning and storage of canisters
Titanium container
Operations diagram
Temporary storage and encapsulation
Temporary storage ventilation
Air inlet filtration
Filtration and air exhaust room
(1) - iorrt entranca
(2.) - Lorrt exir
(3) - Building chanct. Rooms entrance
(4) - Building access (Stain +LIFT)
(5) - emengenct exit
(2)
(7) - mine ventilation facilitr
(B) - Mine lift machinery
(9) - connecting ghllear
(10) - Ventlation: bullding + machine ry
(11) - Mean butbing (Neceiving and conditioning cells)
(12) - temporari storage
(13) - storace ventilatian
(14) - ExTENSION OF TEMPORARY STORAGE
(15) - Ventilation pit
(16) - Mine shaft
(17) - Working area
(186) - access and ventilation ghlifer
(19) - Storage tunnel
(20) - Storage pit
(21) - mine mir exhaust duct
(22) - Stand br machinerr

|  | Accul | $9 / 77$ | $u_{p}$ dating | A |
| :---: | :---: | :---: | :---: | :---: |
| FINAL REPOSITORY FOR VITRIFIED HLW |  |  |  |  |
|  | DESIGN STUDIES |  |  |  |

General perspective




[^2]




Weight of container Jainless steel + glase Weight of lead titanium container Total weight $\begin{array}{r}500 \mathrm{Kgp} \\ 3300 \mathrm{Kgp} \\ 87 \mathrm{Kgp} \\ \hline\end{array}$ $\frac{81 \mathrm{Kgp}}{3887 \mathrm{Kgp}}$


TEMPORARY STORAGE VENTILATION

|  | - |
| :---: | :---: |
| S | 187900020 |



Titanium contoiner






## GEOLOGICAL STORAGE

TEMPORARY STORAGE VENTILATION

| SGN SAINT GOBAAN FITMS | 1/1/ ${ }^{\text {a }}$ |
| :---: | :---: |
|  | 1879 D0020 |


$\xrightarrow{B}$
C.C SECTION


```
*
```


$x-2+\quad \mid$
GEOLOGICAL STORAGE

AIR INLET FILTRATION

$\leqslant A$
$B+C$


PLAN VIEW



GEOLOGICAL STORAGE

01 Källstyrkor i utbränt bränsle och hägaktivt avfall frin on PWR beraiknade med ORIGEN
Nils Kjellbert
AB Atomenergi 77-04-05
02 PM angiende värmeledningstal hos jordmaterial
Sven Knutsson
Roland Pusch
Högskolan i Luleå 77-04-15
03 Deponering av högaktivt avfall i borrhål med buffertsubstans Arvid Jacobsson
Roland Pusch
Högskolan i Lulea 77-05-27
04 Deponering av högaktivt avfall i tunnlar med buffertsubstans Arvid Jacobsson
Roland Pusch
Högskolan i Luleã 77-06-01
05 Orienterande temperaturberäkningar för slutförvaring i berg av radioaktivt avfall, Rapport 1
Roland Blomqvist
AB Atomenergi 77-03-17
06 Groundwater movements around a repository, Phase 1 , State of the art and detailed study plan Ulf tindblom
Hagconsult AB 77-02-28
07 Resteffekt studier för KBS
De1 1 Litteraturgenomgång
Del 2 Beräkningar
Kim Ekberg
Nils Kjellbert
GÖran Olsson
AB Atomenergi 77-04-19
08 Utlakning av franskt, engelskt och kanadensiskt glas med högaktivt avfall
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[^0]:    Reference pwr equiligrium fuel cycle -- fuel decay times

[^1]:    A N N E X
    2 (cont.)

[^2]:    1 Cask receiving (cask externai washing, shock absorber dismouting)
    2 Cask lay down on lifting cart
    3 Monitoring of the internal activity of rask
    4 fixing of honding pari on cask lid
    5 Cask lid unbolting
    
    Cask lid removal
    8 Giose corthinere unloading
    915 g as: curfainery conso sorants
    
    11 Lid welding on sn steel on, to mers for contaminated gione con ala ans only
    12 Possibility of si steel containct washing
    13 Trans $t$ con: for on contamer
    14 Cast fo: gion are's men on the sioroge pit
    15 Tenturary sharos

